

## Synthesis of Some Pyrazol-5-ols Related to Muscimol

JYTTE LYKKEBERG

Royal Danish School of Pharmacy, Department of Chemistry BC, DK-2100 Copenhagen, Denmark

The syntheses of the zwitterions 3-aminomethylpyrazol-5-ol (*5a*), 3-(2-aminoethyl)pyrazol-5-ol (*5b*), 3-(1-aminoethyl)pyrazol-5-ol (*5c*), and 4,4-dimethyl-4,5,6,7-tetrahydropyrazolo[4,3-*c*]pyridin-3-ol (*7b*) are described. *5a–c* were prepared from the corresponding dihydrochlorides *4a–c* by treatment with triethylamine. The 3-(2-aminoethyl)- and the 3-(1-aminoethyl)pyrazol-5-ol dihydrochloride (*4b,c*) were obtained by treatment of *N*-phthaloyl  $\beta$ - and  $\alpha$ -alanylmalonic ester (*8b,c*), respectively, with hydrazine hydrate in glacial acetic acid followed by refluxing of the formed 3-(*N*-phthaloylalanyl)-4-ethoxycarbonylpyrazol-5-ols *9b,c* with 7 M hydrochloric acid. The 3-(2-aminoethyl)pyrazol-5-ol dihydrochloride (*4b*) reacted with acetone and triethylamine to give the 4,4-dimethyl-4,5,6,7-tetrahydropyrazolo[4,3-*c*]pyridin-3-ol (*7b*).

The syntheses of 3-[*N*-(*o*-carbazoylbenzoyl)-2-aminoethyl]pyrazol-5-ol hydrochloride (*6b*), *N*-phthaloyl- $\beta$ -alanine hydrazide (*10b*), and the new tetracyclic ring system 4,5-dihydro-1-hydroxy-7*H*-pyrazolo[4',3':3,4]pyrido[2,1-*a*]isindol-7-one (*11b*) are also described.

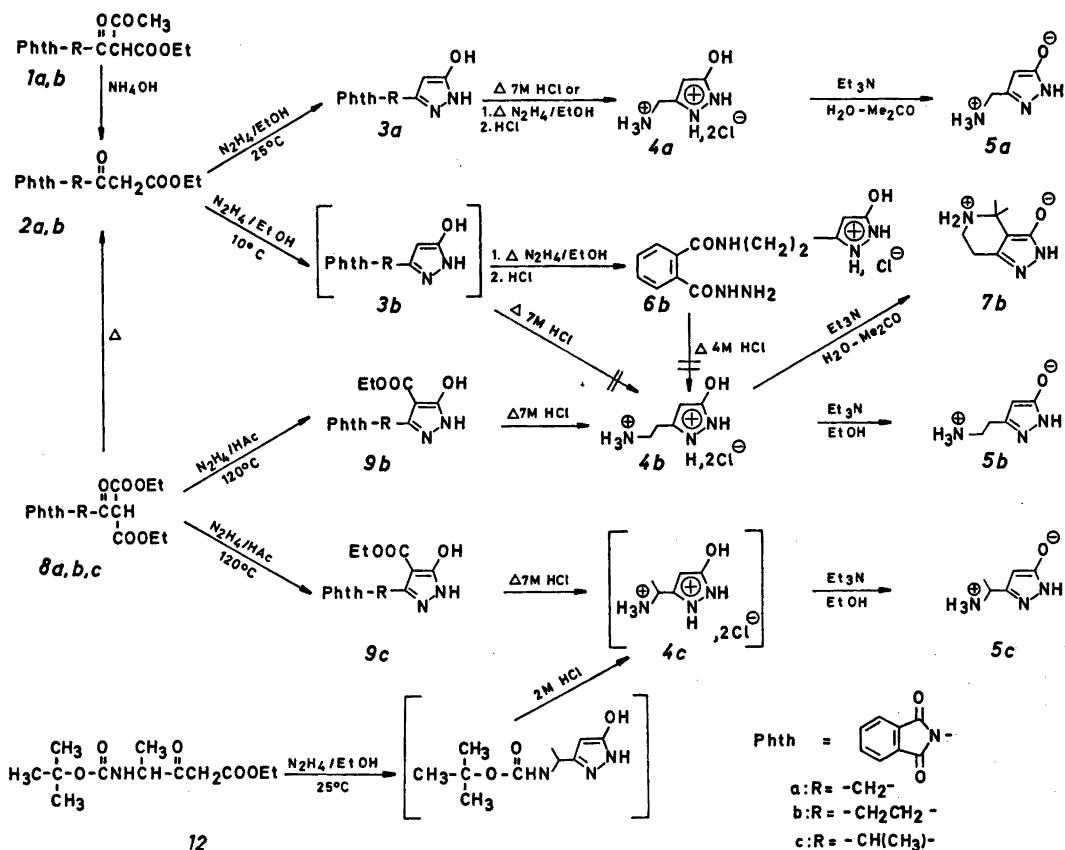
Muscimol (5-aminomethyl-3-isoxazolol), a constituent of *Amanita muscaria*, is a powerful GABA ( $\gamma$ -aminobutyric acid) agonist.<sup>1,2</sup> Several muscimol analogues containing various aminoalkyl groups in position 4 or 5 of the 3-isoxazolol nucleus have been synthesized<sup>3–6</sup> and tested biologically.<sup>7</sup> As part of our interest in muscimol analogues in which the heterocyclic ring system has been changed,<sup>8</sup> the zwitterions 3-aminomethylpyrazol-5-ol (azamuscimol) (*5a*), 3-(2-aminoethyl)pyrazol-5-ol (homoazamuscimol) (*5b*), 3-(1-aminoethyl)pyrazol-5-ol (*5c*), and 4,4-dimethyl-4,5,6,7-tetrahydropyrazolo[4,3-*c*]pyridin-3-ol (*7b*) have now been prepared. Azamuscimol (*5a*) has a relatively weak depressant action on spinal neurons in cats (microelectrophoresis),<sup>7</sup> and it has been shown

to be an inhibitor of L-glutamate decarboxylase (GAD).<sup>9</sup> Finally azamuscimol (*5a*) has been tested *in vitro* for antitubercular activity, but no antibacterial activity could be shown. The biological properties of the azamuscimol analogues *5b*, *5c*, and *7b* will be examined in the near future.

Azamuscimol dihydrochloride (*4a*)<sup>10</sup> was converted into the zwitterion *5a* by treatment with triethylamine followed by addition of acetone.

A new improved synthesis of ethyl (*N*-phthaloyl- $\beta$ -alanyl)acetate (*2b*)<sup>11</sup> was based on treatment of ethyl (*N*-phthaloyl- $\beta$ -alanyl)acetate (*1b*) with concentrated aqueous ammonia in ethanol. Attempts to synthesize homoazamuscimol (*5b*) by the reaction sequence used for the preparation of azamuscimol (*5a*) (Scheme 1) failed, as *2b* with hydrazine hydrate in ethanol at room temperature gave *11b* (Scheme 2) as the main product. The tetracyclic ring system of *11b* has not previously been described, but the corresponding tricyclic ring system without the pyrazolo ring is well-known, as *1b* by treatment with base (pyridine) yielded 1-acetyl-3,4-dihydropyrido[2,1-*a*]isindole-2,6-dione.<sup>11</sup> Treatment of *2b* with hydrazine hydrate in ethanol at 0–12 °C yielded a mixture of 3-(*N*-phthaloyl-2-aminoethyl)pyrazol-5-ol (*3b*) and unreacted *2b*. Attempts to remove the phthaloyl group of *3b* by treatment with hydrazine hydrate in boiling ethanol led to *6b*, which was stable to boiling with 4 M hydrochloric acid for 3 h. Refluxing of *3b* in 7 M hydrochloric acid for 3 h gave a complex reaction mixture containing unreacted *3b*.

Finally the dihydrochloride *4b* of homoazamuscimol was prepared by treatment of *3b* with hydrazine hydrate in glacial acetic acid followed by refluxing of the pyrazolol *9b*



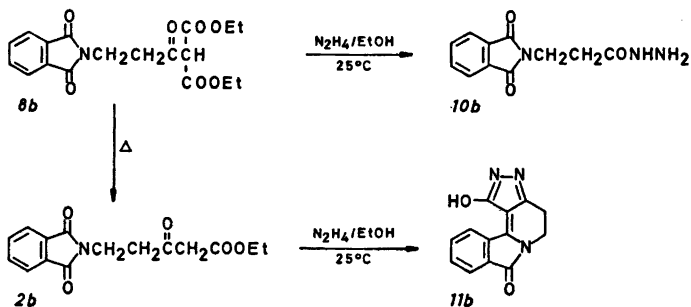
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Scheme 1.

with 7 M hydrochloric acid Treatment of **8b** with hydrazine hydrate in ethanol at ambient temperature yielded *N*-phthaloyl- $\beta$ -alanine hydrazide (**10b**) (Scheme 2). Homoazamuscimol (**5b**) was prepared by treatment of the dihydrochloride **4b** with triethylamine in ethanol. Attempts to convert **4b** into **5b** by treatment with triethylamine and acetone led to the

tetrahydropyrazolo[4,3-*c*]pyridin-3-ol **7b**, instead of the expected zwitterion **5b**.

The 3-(1-aminoethyl)pyrazol-5-ol zwitterion (**5c**) could be prepared by the procedure used for homoazamuscimol (**5b**) using **8c** as the starting material (Scheme 1). In an attempt to elevate the yield of **5c**, **12** was used as starting material but the sequence outlined in



Scheme 2.

Scheme 1 also yielded *5c* in a rather poor yield. Attempts to convert *8c* into ethyl (*N*-phthaloyl- $\alpha$ -alanyl)acetate (*2c*) by refluxing of *8c* with water or pyridine-water failed.

The structure determinations of the new compounds *4b*, *5a-c*, *11b*, *6b*, *9b,c*, *7b* and *10b* were based on IR, UV, and  $^1\text{H}$  NMR spectroscopy and supported by elemental analyses. The structure of the new ring system of *11b* was further confirmed by mass spectrometry. The described pyrazoles were assigned the pyrazolol structure rather than the pyrazolinone structure due to absence of infrared absorption at 1710–1700  $\text{cm}^{-1}$  and absence of ultraviolet absorption about 250 nm with an  $\epsilon$  value of  $1.8\text{--}2.5 \times 10^4$  and about 310 nm with an  $\epsilon$  value of  $8.0 \times 10^3$ , all characteristics of pyrazolinones.<sup>12,13</sup> The structure determination of the pyrazolols was supported by  $^1\text{H}$  NMR spectroscopy and by the finding that the pyrazolols formed coloured complexes with ferric chloride.<sup>12</sup> Broad IR absorptions of *5a-c* and *7b* in the range 3600–1800  $\text{cm}^{-1}$  are in agreement with the presence of ammonium groups in these compounds, and consequently these compounds are zwitterions. The zwitterion *5c* crystallized with 1 mol of water, the removal of which by heating was accompanied by destruction of the compound.

## EXPERIMENTAL

Melting points are corrected and were determined with a hot stage microscope (Mikroskop-Heiztisch, 350 Ernst Leitz G.m.b.H., Wetzlar). Thin layer chromatography (TLC) was accomplished by using silica gel GF<sub>254</sub> plates (Merck). The recording of IR (KBr technique), UV (methanol solutions), and  $^1\text{H}$  NMR spectra, and the performance of microanalyses were accomplished as described in a previous paper.<sup>8</sup> The mass spectrum was obtained with an AEI MS 902 mass spectrometer operating at 70 eV. pH values were measured on a Radiometer pH meter 26 and the  $\text{p}K_{\text{A}}$  values were determined according to the method of Albert and Serjeant.<sup>14</sup>

*3-Aminomethylpyrazol-5-ol zwitterion (5a)*. A solution of 1.86 g (10 mmol) of *4a*<sup>10</sup> in 3 ml of water was chilled in an ice bath and treated with 2.23 g (22 mmol) of triethylamine. The reaction mixture was diluted with 10 ml of acetone, stirred thoroughly, and filtered. Recrystallization (ethanol–water) gave 870 mg (66%) of *5a*· $\text{H}_2\text{O}$ , m.p. 220 °C (decomp.). Anal.  $\text{C}_4\text{H}_7\text{N}_3\text{O}_2\text{H}_2\text{O}$ : C, H, N. The water of crystallization could be removed upon heating

*in vacuo* at 110 °C for 3 h: Anal.  $\text{C}_4\text{H}_7\text{N}_3\text{O}$ : C, H, N.  $\lambda_{\text{max}}$  240 nm ( $\epsilon = 4.00 \times 10^3$ ). IR data: 3600–2000 (s) (with submaxima at 3350, 3200, 3050, 2900, 2800, 2700, 2600, 2450, 2100), 1660 (m), 1620 (s), 1580 (s), 1550 (s), 1470 (s)  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR data (DMSO-*d*<sub>6</sub>):  $\delta$  5.24 (s, 1 H, C=CH); 4.2 (broad signal from  $\text{H}_2\text{O}$  and NH protons); 3.52 (s, 2 H, N- $\text{CH}_2$ ).  $\text{p}K_{\text{A}}$  values ( $\text{H}_2\text{O}$ , 21 °C):  $1.85 \pm 0.09$ ,  $6.15 \pm 0.02$ , and  $9.78 \pm 0.02$ .

*Ethyl (N-phthaloyl- $\beta$ -alanyl)acetate (2b)*. A solution of 13 g (40 mmol) of *1b*<sup>11</sup> in 60 ml of ethanol was treated with 4 ml of aqueous ammonia ( $\rho$  0.88) during 10 min at 80 °C. After cooling to room temperature the precipitate was collected and washed with two 5 ml portions of ethanol. After drying 6.1 g (53%) of *2b* was obtained, m.p. 90–91 °C (Ref. 11, m.p. 90 °C).  $\lambda_{\text{max}}$  290 nm ( $\epsilon = 1.80 \times 10^3$ ), 241 nm ( $\epsilon = 9.79 \times 10^3$ ), 232 nm ( $\epsilon = 1.36 \times 10^4$ ), and 220 nm ( $\epsilon = 3.63 \times 10^4$ ). IR data: 3450 (m), 1770 (m), 1740 (s), 1710 (s), 1440 (m)  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR data (DMSO-*d*<sub>6</sub>):  $\delta$  7.87 (s, 4 aromatic protons); 4.10 (q,  $J \sim 7$  Hz, 2 H, C- $\text{CH}_2$ -OCO); 3.82 (t,  $J \sim 7$  Hz, 2 H, N- $\text{CH}_2$ -C); 3.63 (s, 2 H, CO- $\text{CH}_2$ -COO); 2.98 (t,  $J \sim 7$  Hz, 2 H, C- $\text{CH}_2$ -CO); 1.17 (t,  $J \sim 7$  Hz, 3 H,  $\text{CH}_3$ ).

*4,5-Dihydro-1-hydroxy-7H-pyrazolo[4',3':3,4]-pyrido[2,1-a]isoindol-7-one (11b)*. To a suspension of 5.2 g (18 mmol) of *2b*<sup>11</sup> in 75 ml of ethanol was added 900 mg (18 mmol) of hydrazine hydrate. After stirring overnight at room temperature, the precipitate was collected and dried to give 1.2 g (28%) of *11b*. An analytical sample was recrystallized in ethanol to give *11b* as red needles, m.p. 284–285 °C. Anal.  $\text{C}_{13}\text{H}_{10}\text{O}_2\text{N}_3$ : C, H, N.  $\lambda_{\text{max}}$  306 nm ( $\epsilon = 1.61 \times 10^4$ ) and 232 nm ( $\epsilon = 2.18 \times 10^4$ ). IR data: 3250 (m), 1710 (s), 1680 (s), 1620 (s), 1570 (m), 1470 (m), 1410 (m)  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR data (CF<sub>3</sub>COOD):  $\delta$  9.1 (broad signal, 1 H, NH); 8.0 (s, 4 aromatic protons); 4.40 (t, 2 H,  $\text{CH}_2$ ); 3.53 (t, 2 H,  $\text{CH}_2$ ). MS [ $m/e$  (% rel. int.)]: 237 (100, M), 210 (32 [M-N<sub>2</sub>H]),<sup>15</sup> 182 (71, [M-CON<sub>2</sub>H]).<sup>15</sup>

*3-[N-(o-Carbazoylbenzoyl)-2-aminoethyl]-pyrazol-5-ol hydrochloride (6b)*. To a suspension of 2.6 g (9 mmol) of *2b*<sup>11</sup> in 60 ml of ethanol was added 450 mg (9 mmol) of hydrazine hydrate. After stirring at 12 °C for 3 h the precipitate was collected and dried to give 1.4 g of a mixture of *3b* and *2b* (rendered probable by TLC). The crude mixture (1.4 g) and 300 mg (6 mmol) of hydrazine hydrate in 15 ml of ethanol was heated at reflux for 1 h. After cooling to room temperature the precipitate was collected and dissolved in 4 N hydrochloric acid. The solution was concentrated *in vacuo* and the crystalline residue recrystallized from methanol-ether to give 580 mg of *6b* (20%), m.p. 285 °C (decomp.). Anal.  $\text{C}_{13}\text{H}_{16}\text{N}_5\text{O}_3\text{Cl}$ : C, H, N, Cl.  $\lambda_{\text{max}}$  299 nm ( $\epsilon = 6.73 \times 10^3$ ), 249 nm ( $\epsilon = 1.02 \times 10^4$ ), and 219 nm ( $\epsilon = 2.69 \times 10^4$ ). IR data: 3700–1900 (s), 1660–1560 (s) (with

submaxima at 1655, 1645, 1635, 1630, 1620, 1610, 1545 (m), 1530 (s), 1480 (m), 1465 (s)  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR data (DMSO- $d_6$ ):  $\delta$  12.7 (s, 1 H, NH); 8.2 (m, 2 H, NH); 7.8 (m, 4 aromatic protons); 5.8 (broad, 5 H, NH, OH, and C=CH); 2.9 (s, 4 H,  $2 \times \text{CH}_2$ ).

**3-(N-Phthaloyl-2-aminoethyl)-4-ethoxycarbonylpyrazol-5-ol (9b).** A mixture of 1.8 g (5 mmol) of **8b**<sup>11</sup> and 250  $\mu\text{l}$  (5 mmol) of hydrazine hydrate in 8 ml of glacial acetic acid was refluxed for 1 h. The solution was evaporated *in vacuo* and upon recrystallization (ethanol) of the residue 0.9 g (55 %) of **9b** was obtained; m.p. 175–177 °C. Anal.  $\text{C}_{16}\text{H}_{15}\text{N}_3\text{O}_5$ : C, H, N.  $\lambda_{\text{max}}$  263 nm ( $\epsilon = 6.12 \times 10^3$ ), 240 nm ( $\epsilon = 1.31 \times 10^4$ ), and 220 nm ( $\epsilon = 4.67 \times 10^4$ ). IR data: 3250 (m), 1770 (s), 1710 (s), 1590 (s), 1530 (m), 1470 (m), 1440 (m)  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR data (DMSO- $d_6$ ):  $\delta$  9–12 (broadened signal, 2 H, NH and OH); 7.83 (s, 4 aromatic protons); 4.05 (q,  $J \sim 7$  Hz, 2 H, C- $\text{CH}_2$ -OCO); 3.85 (t,  $J \sim 6$  Hz, 2 H, N- $\text{CH}_2$ -C); 3.03 (t,  $J \sim 6$  Hz, 2 H, C- $\text{CH}_2$ -C=N); 1.22 (t,  $J \sim 7$  Hz, 3 H,  $\text{CH}_3$ ).

**3-(2-Aminoethyl)pyrazol-5-ol dihydrochloride (4b).** A mixture of 3.7 g (11 mmol) of **9b** and 90 ml of 7 M hydrochloric acid was refluxed for 2 h. After cooling to room temperature the precipitate was removed and the solution was concentrated *in vacuo*. The crystalline residue was recrystallized from methanol-ether to give 1.2 g (55 %) of **4b**, m.p. 185–188 °C (decomp.). Anal.  $\text{C}_6\text{H}_{11}\text{N}_3\text{OCl}_2$ : C, H, N, Cl.  $\lambda_{\text{max}}$  246 nm ( $\epsilon = 1.73 \times 10^3$ ), and 218 nm ( $\epsilon = 5.49 \times 10^3$ ). IR data: 3600–2100 (s) (with submaxima at 3350 and 3150), 1625 (s), 1570 (m), 1540 (m), 1470 (s)  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  13.37 (s, 3 H, NH and OH, exchangeable with  $\text{D}_2\text{O}$ ); 8.40 (s, 3 H, NH); 5.87 (s, 1 H, C=CH); 3.07 (s, 4 H,  $2 \times \text{CH}_2$ ).

**3-(2-Aminoethyl)pyrazol-5-ol zwitterion (5b).** A solution of 400 mg (2 mmol) of **4b** in 10 ml of ethanol was chilled in an ice bath and treated with 600  $\mu\text{l}$  (4.4 mmol) of triethylamine. The precipitate was collected and dried to give 195 mg of crude **5b**. Recrystallization (water-ethanol-ether) gave 95 mg (37 %) of **5b**, m.p. 200 °C (decomp.). Anal.  $\text{C}_6\text{H}_9\text{N}_3\text{O}$ : C, H, N.  $\lambda_{\text{max}}$  240 nm ( $\epsilon = 3.30 \times 10^3$ ), and 226 nm ( $\epsilon = 3.77 \times 10^3$ ). IR data: 3500–1800 (s) (with submaxima at 3250, 2950, 2750, 2650, 2400), 1660–1490 (s) (with submaxima at 1580, 1570, 1560, 1550), 1490–1390 (s) (with submaxima at 1470, 1460, 1450, 1420)  $\text{cm}^{-1}$ .  $\text{pK}_A$  values ( $\text{H}_2\text{O}$ , 21 °C):  $2.08 \pm 0.06$ ,  $6.98 \pm 0.01$ ,  $10.31 \pm 0.05$ .

**4,4-Dimethyl-4,5,6,7-tetrahydropyrazolo[4,3-c]pyridin-3-ol zwitterion (7b).** **7b** was synthesized as described above for **5a** using 600 mg (3 mmol) of **4b** and 920  $\mu\text{l}$  (6.6 mmol) of triethylamine to give 500 mg of crude product. Recrystallization (methanol-ether) gave 380 mg (68 %) of **7b**. $\text{H}_2\text{O}$ ; m.p. 200 °C (decomp.). Anal.  $\text{C}_8\text{H}_{13}\text{N}_3\text{O} \cdot \text{H}_2\text{O}$ : C, H, N. The water of crystallization could be removed upon heating

*in vacuo* at 120 °C for 18 h. Anal.  $\text{C}_8\text{H}_{13}\text{N}_3\text{O}$ : C, H, N.  $\lambda_{\text{max}}$  243 nm ( $\epsilon = 4.82 \times 10^3$ ). IR data: 3600–2000 (s) (with submaxima at 3200, 3100, 2980, 2700, 2600, 2480, 2250), 1630 (s), 1580 (s), 1530 (s), 1490 (s), 1430 (s)  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR data (DMSO- $d_6$ ):  $\delta$  5.5 (broad s, 5 H,  $\text{H}_2\text{O}$  and 3 NH, exchangeable with  $\text{D}_2\text{O}$ ); 2.85 (t,  $J \sim 5$  Hz, 2 H,  $\text{CH}_2$ ); 2.32 (t,  $J \sim 5$  Hz, 2 H,  $\text{CH}_2$ ); 1.22 (s, 6 H,  $2 \times \text{CH}_3$ ).

**N-Phthaloyl- $\beta$ -alanine hydrazide (10b).** **10b** was synthesized as described above for **11b** using 1.8 g (5 mmol) of **8b**<sup>11</sup> and 250  $\mu\text{l}$  (5 mmol) of hydrazine hydrate. After stirring for 5 min 0.9 g (77 %) of **10b** was obtained. An analytical sample was recrystallized (ethanol) to give **10b**; m.p. 174–176 °C. Anal.  $\text{C}_{11}\text{H}_{11}\text{N}_3\text{O}_3$ : C, H, N.  $\lambda_{\text{max}}$  293 nm ( $\epsilon = 1.67 \times 10^3$ ), 241 nm ( $\epsilon = 1.06 \times 10^4$ ), 232 nm ( $\epsilon = 1.59 \times 10^4$ ), and 220 nm ( $\epsilon = 4.01 \times 10^4$ ). IR data: 3300 (m), 1770 (m), 1710 (s), 1640 (s), 1610 (m), 1530 (m), 1440 (m)  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR data (DMSO- $d_6$ ):  $\delta$  7.85 (s, 4 aromatic protons); 5.1 (broad signal, 3 H, NH, exchangeable with  $\text{D}_2\text{O}$ ); 3.75 (t,  $J \sim 7$  Hz, 2 H, N- $\text{CH}_2$ -C); 2.37 (t,  $J \sim 7$  Hz, 2 H, C- $\text{CH}_2$ -CO).

**3-(N-Phthaloyl-1-aminoethyl)-4-ethoxycarbonylpyrazol-5-ol (9c).** A mixture of 1.8 g (5 mmol) of **8c**<sup>16</sup> and 250  $\mu\text{l}$  (5 mmol) of hydrazine hydrate in 8 ml of glacial acetic acid was refluxed for 10 min. The solution was evaporated *in vacuo* to give a yellow gum. The residue was taken up in a saturated sodium carbonate solution. The basic solution was extracted with ether and adjusted to pH 1–2 with 4 M hydrochloric acid. The precipitate was collected and recrystallized from ethanol-water to give 0.8 g (49 %) of **9c**; m.p. 150–152 °C. Anal.  $\text{C}_{18}\text{H}_{19}\text{N}_3\text{O}_5$ : C, H, N.  $\lambda_{\text{max}}$  260 nm ( $\epsilon = 3.80 \times 10^3$ ), 240 nm ( $\epsilon = 1.13 \times 10^4$ ), and 220 nm ( $\epsilon = 3.75 \times 10^4$ ). IR data: 3500 (m), 3400 (m), 3200 (m), 1780 (m), 1710 (s), 1540 (m), 1515 (m), 1505 (m), 1470 (m)  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR data (DMSO- $d_6$ ):  $\delta$  11.4 (broad signal, 2 H, NH and OH); 7.82 (s, 4 aromatic protons); 5.60 (q,  $J \sim 7$  Hz, 1 H, N-CH-C); 4.03 (q,  $J \sim 7$  Hz, 2 H, C- $\text{CH}_2$ -OCO); 1.75 (d,  $J \sim 7$  Hz, 3 H,  $\text{CH}_3$ -C-N); 1.08 (t,  $J \sim 7$  Hz, 3 H,  $\text{CH}_3$ -C-OCO).

**3-(1-Aminoethyl)pyrazol-5-ol zwitterion (5c).** (a) A mixture of 500 mg (1.5 mmol) of **9c** and 12 ml of 7 M hydrochloric acid was refluxed for 2 h. After cooling to room temperature the precipitate was removed and the solution was concentrated *in vacuo* to give 200 mg (1 mmol) of **4c** as an oil. The oily product was dissolved in 5 ml of ethanol and treated with 280  $\mu\text{l}$  (2 mmol) of triethylamine and 100  $\mu\text{l}$  of ether. The precipitate was collected, dried and upon recrystallization (water-ethanol-ether) 45 mg (21 %) of **5c**. $\text{H}_2\text{O}$  was obtained; m.p. 207–210 °C. Found: C 41.05; H 7.10; N 28.50. Calc. for  $\text{C}_6\text{H}_9\text{N}_3\text{O} \cdot \text{H}_2\text{O}$ : C 41.37; H 7.64; N 28.95.  $\lambda_{\text{max}}$  240 nm ( $\epsilon = 3.12 \times 10^3$ ), and 221 nm ( $\epsilon = 4.85 \times 10^3$ ). IR data: 3600–2000 (s) (with submaxima at 3250, 3050, 2875, 2675,

2500, 2100), 1660–1490 (s) (with submaxima at 1640, 1565, 1550, 1540), 1490–1420 (s) (with submaxima at 1460, 1450, 1440)  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR data (DMSO- $d_6$ ):  $\delta$  7.9 (broad signal, 6 H, 4 NH and  $\text{H}_2\text{O}$ ); 5.3 (s, 1 H, C=CH); 4.0 (broad signal, 1 H, CH); 1.4 (broad signal, 3 H,  $\text{CH}_3$ ).

(b) A solution of 1.3 g (5 mmol) of  $12^{18}$  in 20 ml of ethanol was treated with 250  $\mu\text{l}$  (5 mmol) of hydrazine hydrate for 10 min. The solution was concentrated *in vacuo* and the oily residue was dissolved in 20 ml of 2 M hydrochloric acid. The acid solution was evaporated *in vacuo* to give 450 mg (2.3 mmol) of *4c* as an oil. The oily product was dissolved in 10 ml of ethanol and treated with 550  $\mu\text{l}$  (4 mmol) of triethylamine. The precipitate was collected, dried and upon recrystallization (water-ethanol) 70 mg (10 %) of *5c*· $\text{H}_2\text{O}$  was obtained.

**Biological test.** Azamuscimol (*5a*) was tested *in vitro* on Löwenstein-Jensen medium against strains of *M. tuberculosis* (No. E 10883/45 and 1921/41) as described by Engbæk *et al.*<sup>17</sup> Azamuscimol (*5a*) had no influence on the bacteria growth at concentrations where isoniazid caused complete inhibition.

**Acknowledgements.** The author is very grateful to cand. pharm. Bodil Vergmann, Tuberculosis Department, Statens Seruminstitut, Copenhagen, Denmark, for testing azamuscimol (*5a*), and to M. Sc. A. Kjær Sørensen, Danish Civil Defence Analytical-Chemical Laboratory, Copenhagen, for the measurements of the mass spectra. The work has been supported by the Danish Medical Research Council.

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Received June 9, 1977.