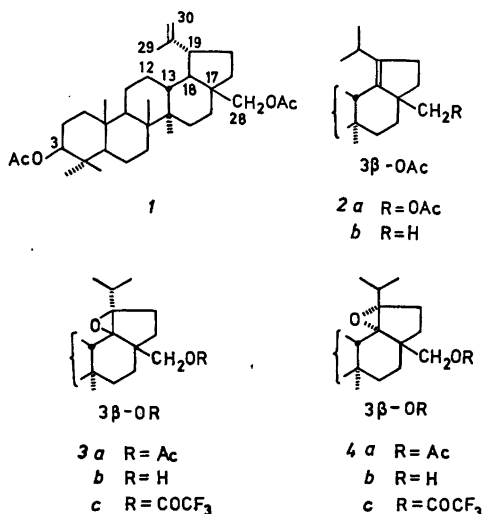


Triterpenes. BF_3 Catalysed Rearrangement of $3\beta,28$ -Diacetoxy- $18(\alpha \text{ or } \beta),19(\alpha \text{ or } \beta)$ -epoxylupane. A Novel Route to Baccharane Derivatives

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Both the $3\beta,28$ -diacetoxy- $18\alpha,19\alpha$ -epoxylupane ($4a$) and the corresponding $18\beta,19\beta$ -isomer ($3a$) gave $3\beta,28$ -diacetoxy- $18,19$ -secolup- $13(18)$ -en- 19 -one ($5a$) on treatment with BF_3 -etherate. $18\beta,19\beta$ -Epoxy isomer ($3a$) gave, in addition, $3\beta,28$ -diacetoxy-lupa- $12,18$ -diene (6). The baccharane derivative $5a$ was also synthesized *via* another route. The stereochemistry of the isomeric $18,19$ -epoxy-lupane derivatives 3 and 4 is discussed.



The rather rare tetracyclic triterpenes with a six-membered D-ring include the baccharane series, the parent hydrocarbon of which has been shown to be $18,19$ -secolupane.^{1,2} In this communication we report a novel route to a polyfunctional baccharane derivative starting from lupane derivatives. Nomenclature based on the lupane skeleton is applied throughout.

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Betulin diacetate (1) can be transformed to its Δ^{18} -isomer ($2a$), either through successive dehydrogenation and hydrogenation steps,^{3,4} or *via* acid catalysed double-bond migration.⁵ On epoxidation of $2a$, both *m*-chloroperbenzoic acid and performic acid give as the main product the same epoxide isomer; only a few percent of the other epoxide isomer was formed. Treatment of $2a$ with ozone yields only the epoxide obtained as a minor product from the peracid epoxidation. Epoxidation of triterpenes with peracids usually occurs by attack from the α -side, on account of steric hindrance due to methyl groups on the β -side. When considering oxidative attack of the double bond in compound $2a$, it is difficult to conclude from Dreiding models, however, which side is the less hindered. The ^1H NMR spectra of the epoxides show different chemical shifts for the protons at the acetoxy carbon C-28. In the epoxide obtained with peracids the C-28 protons have higher δ -values (in $3a$ doublets at δ 3.80 and 4.50) and are magnetically more different than those of the epoxide produced by ozone (in $4a$ doublets at δ 3.74 and 4.20). This difference can be explained by hindered rotation about the C-17–C-28 axis and the influence of the magnetic anisotropy of a β -epoxy oxygen, thus indicating the β -epoxy structure 3 for the main epoxide produced by peracid and the α -epoxy structure 4 for the epoxide obtained by treatment with ozone.

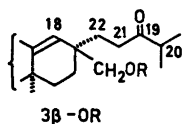
Further information was sought from the NMR shift reagent spectra of epoxides $3c$ and $4c$, where hydroxy groups are protected as

trifluoroacetates. The $\text{Eu}(\text{DPM})_3$ or $\text{Eu}(\text{fod})_3$ induced shifts in the spectrum of the presumed β -epoxide agree with the β -structure **3c** but they do not strictly exclude the possibility of the α -structure **4c**. As to the presumed α -epoxide **4c**, the induced shifts are so small that it is evident that Eu-reagent does not form a dipolar complex with the epoxy oxygen thus preventing clear deductions.

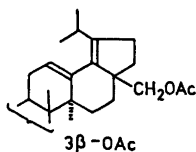
Both epoxides are highly resistant toward anionic species (HO^- , H^- , several R^- , etc.) and attempted determination of the stereostructure of the epoxides with chemical methods based on these reagents failed.

Results from X-ray crystallographic measurements,⁶ however, confirmed the assumption that the epoxide obtained with peracids has β -epoxy structure **3** and thus the epoxide produced by ozone the α -epoxy structure **4**.

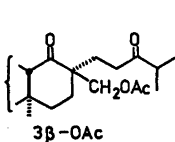
The reason for different stereochemical outcome of ozone and peracid epoxidation is not clear. Any directive effect of C-28 oxygen function is excluded as the corresponding Δ^{18} -lupane derivative **2b**, lacking C-28 oxygen, similarly gives different epoxides as main products when treated with ozone³ and peracid.⁷ The different stereochemistry may be due to the fact that ozone is diradical⁸ and peracid ionic in character.



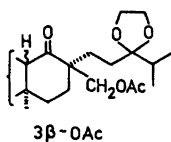
5a R = Ac
b R = COCF_3



6



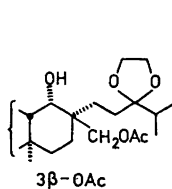
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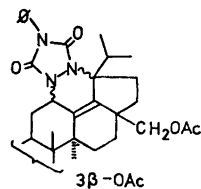
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Both epoxides **3a** and **4a** rearrange to the baccharane derivative **5a** on treatment with BF_3 -etherate in benzene, and the β -epoxide **3a** gives, in addition, as a minor product, the conjugated diene **6**. The double-bond structure of

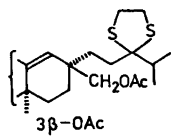
the latter **6** is confirmed by the formation of Diels-Alder adduct **10**. **5a** is reduced to the methylene compound **12** via formation of the thioacetal **11** followed by Raney-Ni desulfurisation. The $\text{Eu}(\text{DPM})_3$ induced shifts to lower field in ^1H NMR spectrum of the trifluoroacetate derivative **5b** are in the expected order: protons at C-22 (2 H, tr, $J=8$ Hz) \simeq at C-20 (1 H, m) $>$ at C-21 (2 H, tr, $J=8$ Hz) $>$ at C-29 and C-30 (6 H, d, $J=7$ Hz).



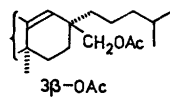
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10



11



12

The baccharane derivative **5a** was also synthesised *via* an independent route. Oxidation of the double bond in **2a** with RuO_4 - NaIO_4 results in the diketone **7**, the side chain carbonyl of which is selectively protected as an ethylene acetal to give the ketoacetal **8**. NaBH_4 reduction of **8** gave a mixture of compounds, among them the hydroxy acetal **9**. The $13\beta\text{H}$ - $18\alpha\text{OH}$ -structure follows from the 4 Hz half-height width of the broad singlet of $18\beta\text{H}$ at δ 3.33. Dehydration of hydroxy acetal **9** in HMPT and subsequent regeneration of the side chain carbonyl at C-19 produces the baccharane derivative **5a**, identical (TLC, m.p., $[\alpha]_D$, IR, ^1H NMR, and MS) with the compound **5a** obtained from the BF_3 -catalysed rearrangement of epoxides **3a** and **4a**.

EXPERIMENTAL

Melting points are uncorrected. ^1H NMR spectra were recorded on a Jeol JNM-PMX 60 spectrometer, the spectra with shift reagent in CS_2 and others in CDCl_3 (unless stated otherwise) and related to internal TMS, the IR

spectra on a Perkin-Elmer 125 spectrometer using KBr pellets, mass spectra on a Perkin-Elmer 270 B mass spectrometer, specific rotations in CHCl_3 solution on a Perkin-Elmer 141 polarimeter, and elemental analyses with a F & M 185 CHN analyser.

Derivatives of 18 β ,19 β -epoxylupane 3a, 3b and 3c. (a) 3 β ,28-Diacetoxy-lup-18-ene^{3,6} (2a) was epoxidised with *m*-chloroperbenzoic acid, as described earlier,⁵ to 3 β ,28-diacetoxy-18 β ,19 β -epoxylupane (3a) m.p. 222–223 °C (EtOH), $[\alpha]_D + 25^\circ$ (c 1.0). (Ref. 5, m.p. 222 °C, $[\alpha]_D + 25.5^\circ$). δ 4.50 (1 H, m, 3 α H), 4.50 and 3.80 (δ 1 H, d, 11 Hz, AB system of C-28 protons). X-Ray crystallographic data will be published⁶ later. A small amount (about 5% from 3a) of 3 β ,28-diacetoxy-18 α ,19 α -epoxylupane (4a), characterised below, was also obtained.

(b) 3 β ,28-Diacetoxy-lup-18-ene (2a) (0.2 g), CH_2Cl_2 (3 ml), EtOAc (5 ml), HCO_2H (2 ml), and 30% H_2O_2 (1 ml) were stirred for 1 h at room temperature. The mixture was washed with water, NaHCO_3 -solution, dried, and the solvent evaporated. Crystallisation from Et_2O –MeOH gave 3 β ,28-diacetoxy-18 β ,19 β -epoxylupane (3a) (0.17 g) with physical constants as above.

Hydrolysis of 3a (1.0 g) with KOH (1.0 g) in boiling EtOH (30 ml) for 1 h, work up, and chromatography on silica plates (CHCl_3 –EtOAc 9:1 eluent), and crystallisation from EtOH gave 3 β ,28-dihydroxy-18 β ,19 β -epoxylupane (3b) (0.75 g), m.p. 245 °C, $[\alpha]_D + 29^\circ$ (c 1.04). Reacetylation in boiling Ac_2O gave the starting epoxide 3a.

3 β ,28-Dihydroxy-18 β ,19 β -epoxylupane (3b) (1.0 g) in CH_2Cl_2 (20 ml) and $(\text{CF}_3\text{CO})_2\text{O}$ (1.0 ml) was allowed to stand for 15 min at room temperature. Most of the solvent was evaporated under reduced pressure. The residue crystallised from EtOH resulting in 3 β ,28-di(trifluoroacetoxy)-18 β ,19 β -epoxylupane (3c) (0.5 g), m.p. 124 °C, $[\alpha]_D + 23^\circ$ (c 1.28); M^+ 650; ν 1775, 1225, 1160, 950, 870, 775, 730, 655; δ 4.58 (1 H, m, 3 α H), 4.46 and 4.15 (δ 1 H, d, 11 Hz, AB system of C-28 protons), 1.05 and 1.00 (δ 3 H, d, 7 Hz, methyls of isopropyl group), 1.12, 1.06, 0.94 (δ 3 H, s), 0.86 (6 H, s).

Derivatives of 18 α ,19 α -epoxylupane 4a, 4b and 4c. 3 β ,28-Diacetoxy-lup-18-ene (2a) (1.0 g) in EtOAc– CH_2Cl_2 (1:1, 50 ml) was ozonised at –75 °C until the solution remained faintly blue. Excess ozone was driven off with a stream of nitrogen and the solution allowed to reach room temperature. Evaporation of the solvent, chromatography on silica plates (CHCl_3 eluent), and crystallisation from EtOH gave 3 β ,28-diacetoxy-18 α ,19 α -epoxylupane (4a) (0.7 g), m.p. 210 °C, $[\alpha]_D + 36.6^\circ$ (c 1.02); M^+ 542; δ 4.44 (1 H, m, 3 α H), 4.20 and 3.74 (δ 1 H, d, 11 Hz, AB system of C-28 protons), 2.02 and 1.98 (δ 3 H, s, acetyl groups), 1.10–0.85 (group of 7 Me). Anal. $\text{C}_{28}\text{H}_{44}\text{O}_5$: C, H.

Hydrolysis of 4a (1.0 g) as the corresponding β -epoxide 3a above gave 3 β ,28-dihydroxy-18 α ,19 α -epoxylupane (4b) (0.5 g) crystallised

from Et_2O –MeOH m.p. 214 °C, $[\alpha]_D + 22.6^\circ$ (c 1.0). Reacetylation in boiling Ac_2O gave the starting diacetate 4a.

3 β ,28-Di(trifluoroacetyl)-18 α ,19 α -epoxylupane (4c) was prepared as the corresponding β -epoxy derivative 3c above. Crystallisation from MeOH– Et_2O gave m.p. 225 °C, $[\alpha]_D + 30^\circ$ (c 1.0); M^+ 650; ν 1770, 1210, 1155, 920, 875, 775, 730; δ 4.74 (1 H, m, 3 α H), 4.59 and 4.05 (δ 1 H, d 11 Hz, AB system of C-28 protons), 1.15–0.89 (group of 7 Me).

BF_3 -catalysed rearrangement of 3 β ,28-diacetoxy-18 β ,19 β -epoxylupane (3a) and trifluoroacetate 3c. 3 β ,28-Diacetoxy-18 β ,19 β -epoxylupane (3a) (0.5 g) in dry benzene (30 ml) and BF_3 -etherate (1 ml) was allowed to stand for 20 min at room temperature. The reaction was interrupted by addition of methanol (5 ml) and the mixture washed with water and NaHCO_3 -solution, dried and the solvent evaporated. Chromatography on silica plates (CHCl_3 –light petroleum, b.p. 60–80 °C, 3:1 eluent) gave two compounds. The less polar 3 β ,28-diacetoxy-lupa-12,18-diene (6) crystallised from EtOH (30 mg), m.p. 140 °C, $[\alpha]_D + 220^\circ$ (c 1.03). (Ref. 4, m.p. 140–143 °C, $[\alpha]_D + 219^\circ$) λ_{max} (EtOH) 234 nm (ϵ 7600); M^+ 524; ν 3020, 1730, 1240; δ (CCl_4) 5.35 (1 H, br. tr, 3 Hz, C-12 vinyl proton), 4.45 (1 H, m, 3 α H), 3.28 (2 H, AB quart, 11 Hz, C-28 protons), 1.05–0.89 (group of 7 Me).

The more polar 3 β ,28-diacetoxy-18,19-secolup-13(18)-en-19-one (5a) crystallised from EtOH (0.35 g), m.p. 115 °C, $[\alpha]_D - 34.8^\circ$ (c 1.15); M^+ 542; ν 1730, 1710, 1240; δ (CCl_4) 4.93 (1 H, br. s, C-18 vinyl proton), 4.40 (1 H, m, 3 α H), 3.79 (2 H, AB quart, 11 Hz, C-28 protons), 2.00 and 1.96 (δ 3 H, s, acetyl methyls), 1.12 (6 H, s), 1.00–0.80 (group of 5 Me). Anal. $\text{C}_{24}\text{H}_{34}\text{O}_5$: C, H.

Similar treatment of 3 β ,28-di(trifluoroacetoxy)-18 β ,19 β -epoxylupane (3c) (1.0 g) with BF_3 -etherate gave 3 β ,28-di(trifluoroacetoxy)-18,19-secolup-13(18)-en-19-one (5b) (0.7 g), m.p. 125 °C (from EtOH), $[\alpha]_D - 24.8^\circ$ (c 1.25); ν 1775, 1710, 1220, 1160.

BF_3 -catalysed rearrangement of 3 β ,28-diacetoxy-18 α ,19 α -epoxylupane (4a). The treatment with BF_3 -etherate of 3 β ,28-diacetoxy-18 α ,19 α -epoxylupane (4a) as the β -epoxide 3a above resulted in quantitatively 3 β ,28-diacetoxy-18,19-secolup-13(18)-en-19-one (5a) identical with the compound 5a obtained from the β -epoxide 3a.

Diels-Alder adduct (10). 3 β ,28-Diacetoxy-lupa-12,18-diene (6) (0.1 g) and 4-phenyl-1,2,4-triazoline-3,5-dione (50 mg) in dry benzene (10 ml) were allowed to stand overnight. The solvent was evaporated and the residue chromatographed on silica plates (CHCl_3 eluent). Crystallisation from EtOH gave adduct 10 (60 mg), m.p. 160 °C decomp., $[\alpha]_D + 162^\circ$ (c 0.56); ν 1730, 1690; δ (CCl_4) 7.40 (5 H, m, phenyl), 4.65 (2 H, m, 3 α H and C-12 proton), 3.95 (2 H, br. s, C-28 protons), 2.04 and 2.00

(δ 3 H, s, acetyl methyls), 1.12–0.85 (group of 7 Me). Anal. $C_{10}H_{17}N_2O_6$: C, H, N.

Reduction of 3 β ,28-diacetoxy-18,19-secolup-13(18)-en-19-one (5a). 3 β ,28-Diacetoxy-18,19-secolup-13(18)-en-19-one (5a) (0.5 g), ethane-1,2-dithiol (2.0 g), and BF_3 -etherate (0.3 ml) was allowed to stand for 15 min at room temperature, methanol (15 ml) was added and the precipitated thioacetal 11 (0.3 g) m.p. 160 °C, $[\alpha]_D -29^\circ$ (c 1.0), refluxed with Raney-Ni in EtOH for 1 h. The Raney-Ni was filtered off, the solvent evaporated and the residue chromatographed on silica plates ($CHCl_3$ -light petroleum, b.p. 60–80 °C, 3:1 eluent). Crystallisation from Et_2O -MeOH gave 3 β ,28-diacetoxy-18,19-secolup-13(18)-ene (12) (0.16 g), m.p. 90 °C, $[\alpha]_D -40^\circ$ (c 1.47); ν 3020, 1735, 1265, 1245, 756; δ 5.00 (1 H, br. s, C-18 vinyl proton), 4.50 (1 H, m, 3 α H), 3.88 (2 H, AB quart, 11 Hz, C-28 protons), 2.05 (6 H, s, acetyl methyls), 0.95–0.80 (group of 7 Me). Anal. $C_{24}H_{36}O_4$: C, H.

3 β ,28-Diacetoxy-18,19-secolupan-18,19-dione (7). 3 β ,28-Diacetoxy-lup-18-ene (2a) (1.0 g) in CCl_4 (30 ml), $NaIO_4$ (1.0 g) in water (20 ml), and a catalytic amount of RuO_4 was shaken vigorously for 20 h. The water layer was removed and the CCl_4 phase washed with water, added MeOH (1 ml), dried, and chromatographed on a short silica column ($CHCl_3$ -light petroleum, b.p. 60–80 °C, 2:1 eluent). Crystallisation from EtOH gave 3 β ,28-diacetoxy-18,19-secolupan-18,19-dione (7) (0.8 g), m.p. 166 °C, $[\alpha]_D +38.7^\circ$ (c 1.2); M^+ 588; ν 1740, 1720, 1705, 1250; δ 4.45 (1 H, m, 3 α H), 4.44 and 3.95 (δ 1 H, d, 11 Hz, AB system of C-28 protons), 3.9–2.2 (4 H, m), 2.03 (6 H, s, acetyl methyls), 1.15–0.8 (group of 7 Me). Anal. $C_{34}H_{54}O_6$: C, H.

3 β ,28-Diacetoxy-18,19-secolupan-18,19-dione 19-ethylene acetal (8). 3 β ,28-Diacetoxy-18,19-secolupan-18,19-dione (7) (1.0 g), ethylene glycol (5 ml), *p*-toluenesulfonic acid (0.1 g), and benzene (50 ml) were refluxed for 4 h and water removed from the condensed azeotrope. The mixture was washed with $NaHCO_3$ -solution, dried, and the solvent evaporated. Crystallisation from Et_2O -MeOH gave 3 β ,28-diacetoxy-18,19-secolupan-18,19-dione 19-ethylene acetal (8) (0.6 g), m.p. 192 °C, $[\alpha]_D +43^\circ$ (c 1.0); M^+ 602; δ 4.43 (1 H, m, 3 α H), 4.42 and 3.90 (δ 1 H, d, 11 Hz, AB system of C-28 protons), 3.87 (4 H, s, ethylene acetal protons). Anal. $C_{38}H_{58}O_7$: C, H.

3 β ,28-Diacetoxy-18 α -hydroxy-18,19-secolupan-19-one ethylene acetal (9). 3 β ,28-Diacetoxy-18,19-secolupan-18,19-dione 19-ethylene acetal (8) (1.0 g) in EtOH (50 ml) and $NaBH_4$ (0.5 g) was stirred for 1 h. Benzene (50 ml) was added and the solution washed with water, dried, and the solvent evaporated. The reaction mixture consisting of several compounds was chromatographed on silica plates ($CHCl_3$ + 2 % EtOAc eluent). The least polar main component 3 β ,28-diacetoxy-18 α -hydroxy-18,19-secolupan-19-one ethylene acetal (9) crystallised from EtOH

(0.2 g), m.p. 163 °C, $[\alpha]_D +10.8^\circ$ (c 1.02); M^+ absent, M^+ –43 561; ν 3500, 1730, 1240; δ 4.40 (1 H, m, 3 α H), 4.1–3.7 (6 H, m, C-28 and 19-ethylene acetal protons), 3.33 (1 H, br. s, half-height width 4 Hz, C-18 β H). Anal. $C_{38}H_{58}O_7$: C, H.

3 β ,28-Diacetoxy-18,19-secolup-13(18)-en-19-one (5a) via dehydration. 3 β ,28-Diacetoxy-18 α -hydroxy-18,19-secolupan-19-one ethylene acetal (9) (0.2 g) in hexamethylphosphoric triamide (3 ml) was heated for 10 min at 240 °C. To the cooled mixture was added benzene (30 ml), washed with water and 1 M H_2SO_4 . Benzene was evaporated and the partially hydrolysed acetal was dissolved in acetone and 1 M H_2SO_4 added. The mixture was stirred for 2 h, worked up, and chromatographed on silica plates ($CHCl_3$ eluent). Crystallisation from Et_2O -MeOH gave 3 β ,28-diacetoxy-18,19-secolup-13(18)-en-19-one (5a) (0.1 g) identical with the product from the BF_3 -rearrangement above.

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