

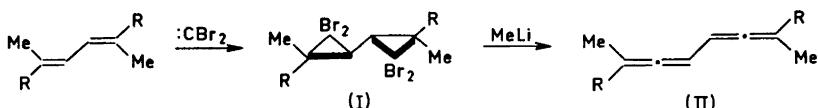
Crystal Structure of 2,7-Diphenyl-2,3,5,6-octatetraene

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The crystals belong to the monoclinic system with space group $P2_1/c$ and cell dimensions $a=15.653 \text{ \AA}$, $b=7.457 \text{ \AA}$, $c=6.454 \text{ \AA}$, $\beta=98.20^\circ$. There are two molecules in the unit cell. The phase problem was solved by direct methods, and the R -value arrived at by full-matrix least squares refinement was 6.0 % for 1049 observed reflections. The cumulated double bonds are 1.310 \AA and 1.306 \AA , respectively, and the single bond length 1.470 \AA . The torsion angle about the bond to the phenyl ring (of length 1.491 \AA) is 20.6°.

When properly substituted, a diallene can exist in a *meso*-configuration or as a racemic modification. The synthesis and some thermal reactions of conjugated *meso*-diallenes have been described by Kleveland and Skattebøl.¹ The two-step synthesis consists of the addition of dibromocarbene to the conjugated dienes followed by treatment of the diadducts with methylolithium ($\text{R}=\text{phenyl}$ and *tert*-butyl):



Structural assignments of intermediates (I) and products (II) were based on spectroscopic evidence and crystallographic space group determinations:

Compound	Space group	Number of molecules in the unit cell
(I)($\text{R}=\text{C}_6\text{H}_5$)	$Pbca$	4
(I)($\text{R}=\text{C}(\text{CH}_3)_3$)	$P2_1/c$	2
(II)($\text{R}=\text{C}_6\text{H}_5$)	$P2_1/c$	2
(II)($\text{R}=\text{C}(\text{CH}_3)_3$)	$P2_1/c$	2

Table 1. Final fractional coordinates and anisotropic temperature factors with estimated standard deviations (multiplied by 10^5 for carbon and 10^4 for hydrogen).

ATOM	X	Y	Z	B	B11	B22	B33	B12	B13	B23
C1	24727(16)	58171(33)	29575(48)		376(12)	1116(58)	1982(74)	150(39)	59(40)	247(198)
C2	31544(19)	58791(49)	42239(49)		402(14)	1046(60)	2281(73)	131(49)	53(59)	424(132)
C3	39574(18)	58776(41)	37405(53)		367(14)	2171(69)	3460(118)	79(49)	81(64)	117(141)
C4	41559(23)	49835(43)	28194(53)		383(14)	2419(74)	3131(103)	173(64)	86(68)	686(154)
C5	35127(28)	41228(43)	7361(53)		523(14)	2172(73)	2688(97)	78(57)	453(68)	275(146)
C6	26769(18)	41441(37)	11948(44)		428(14)	1688(58)	2288(98)	+34(47)	188(59)	+194(128)
C7	15743(16)	49915(33)	34765(42)		362(12)	1292(58)	2878(75)	135(42)	33(52)	84(107)
C8	91444(16)	46392(35)	28454(46)		333(11)	1824(64)	2375(88)	+56(44)	334(55)	138(119)
C9	24866(16)	43887(41)	6355(45)		345(13)	1971(79)	2558(94)	+72(45)	198(56)	+411(126)
C10	14384(24)	53869(51)	56922(53)		445(16)	2583(88)	2788(181)	217(61)	421(65)	+744(154)
H2	3814(15)	6592(35)	5228(44)	3,7(,7)						
H3	4469(17)	6491(36)	4896(46)	5,1(,7)						
H4	4738(17)	4926(38)	1655(44)	5,2(,7)						
H5	3411(14)	3421(38)	4616(51)	5,(,7)						
H6	2223(14)	3488(37)	1611(51)	3,5(,5)						
H9	37(16)	3052(41)	358(46)	5,2(,7)						
H181	841(21)	5229(36)	5811(49)	5,(,7)						
H182	1594(28)	6553(51)	6318(57)	0,4(1,1)						
H183	1838(21)	4582(44)	6869(57)	7,8(,9)						

Under the assumption of an ordered crystal structure, this crystallographic information demands the molecules to be centrosymmetrical, a requirement which is only satisfied by the *meso*-configuration. In order to eliminate the possibility of disorder, a crystal structure determination of (II)(R = C₆H₅) has been carried out.

The crystals belong to the monoclinic system and the systematic absences lead to the space group P2₁/c. The cell parameters measured by means of a four circle diffractometer, and their estimated standard deviations, are: *a* = 15.653(3) Å, *b* = 7.457(2) Å, *c* = 6.454(1) Å, β = 98.20(2) $^\circ$. Since there are two molecules in the unit cell ($\rho_{\text{obs}} = 1.15 \text{ g cm}^{-3}$, $\rho_{\text{calc}} = 1.14 \text{ g cm}^{-3}$), the molecules themselves must be centrosymmetrical.

With $2\theta(\text{max}) = 50^\circ$ and MoK α -radiation, 1445 reflections were measured on an automatic four-circle diffractometer. 1049 were recorded as observed

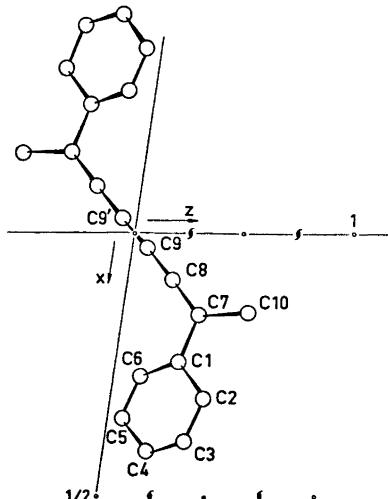


Fig. 1. The molecule viewed along [010].

Table 2. Continued.

L ^a	29	1	4	67	42	8	52	51	8	26	20	0	32	46	4	28	28	+	3	38	32	+	1	81	84				
+13	32	29	6	64	84	+	9	31	25	+	1	24	17	2	115	117	0	43	40	3	190	222	2	49	41				
-12	33	35	7	71	67	+	3	22	20	0	55	51	3	28	28	1	29	27	8	68	61	6	36	28					
-11	23	28	8	67	62	+	2	42	39	2	28	29	4	68	66	2	85	86	7	18	17	1	19	19					
-10	26	26	9	107	99	+	1	18	7	3	38	33	6	71	69	3	26	26	L ^b	+	33	33	+	10	82	80			
+ 8	59	59	10	15	14	0	12	60	L ^b	5, K ^c	1	14	14	1	34	34	0	30	30	+	10	82	80	+	8	36	32		
- 6	19	20	10	49	46	0	12	60	+	4	14	14	1	22	23	5	50	50	+	8	84	83	+	9	37	33			
- 5	43	43	L ^b	4, K ^c	3	2	72	71	+	13	56	58	9	28	17	6	17	16	+	8	19	16	+	3	33	33			
- 4	54	51	14	43	43	3	159	155	+	12	74	74	10	22	11	6	26	26	+	7	94	91	2	0	0	49			
+ 3	40	47	13	54	52	4	80	43	+	10	163	169	L ^b	5, K ^c	3	1	47	47	44	2	44	44	2	44	39	2	44	39	
+ 1	21	19	12	44	41	5	15	12	+	8	166	164	+	14	24	21	11	46	42	1	22	24	3	88	85	2	44	39	
0	63	64	11	42	42	6	41	40	0	7	81	77	+	12	41	39	9	31	24	4	28	26	L ^b	6, K ^c	0	3	33	32	
- 1	21	18	10	42	39	7	22	20	+	6	43	40	11	38	38	+	7	50	50	L ^b	6, K ^c	2	0	0	33	32			
- 2	16	13	9	87	88	9	42	44	5	35	34	10	35	35	0	58	58	+	11	19	19	1	33	33	0	33	32		
- 3	132	111	1	19	10	10	44	33	2	48	43	9	24	23	4	38	35	+	11	19	19	2	28	24	1	28	24		
- 6	66	70	7	19	11	11	33	32	3	41	44	8	69	69	+	3	18	14	14	L ^b	7, K ^c	1	1	1	1	1	1		
- 10	16	12	6	57	57	L ^b	4, K ^c	5	+	2	29	18	+	6	17	15	1	22	17	+	9	56	59	+	7	29	22		
- 12	34	38	8	58	57	+	13	26	27	1	48	47	8	46	49	1	41	38	+	7	83	85	+	8	43	34			
- 14	L ^b	4, K ^c	2	4	69	71	0	34	36	0	33	30	+	4	40	40	40	2	63	5	89	30	+	4	41	44			
- 15	41	38	5	77	76	5	40	39	1	30	30	32	3	26	28	3	18	13	+	4	23	22	2	2	37	38			
- 16	20	18	+	1	37	33	4	49	48	2	156	155	+	1	32	33	9	53	49	+	3	76	76	0	0	0	38		
- 14	18	10	0	77	77	3	43	45	3	21	21	1	22	22	+	8	57	57	L ^b	5, K ^c	0	0	0	0	0	0			
- 13	37	37	1	52	51	+	1	26	22	0	203	227	1	36	36	0	33	33	0	25	18	8	88	87	0	0	0		
- 14	49	49	0	32	27	8	26	26	8	45	44	2	119	118	+	7	30	29	2	33	27	4	84	59	0	0	0		
- 10	33	34	226	218	3	105	96	6	51	53	4	123	115	+	3	21	18	3	158	146	L ^b	7, K ^c	2	0	0	0	0	0	
- 9	244	263	4	26	23	4	47	42	7	19	23	5	73	67	0	25	19	5	61	55	+	8	30	27	0	0	0		
- 8	47	83	5	30	24	4	40	41	9	19	22	6	35	29	1	41	35	6	17	16	+	7	27	35	0	0	0		
- 7	20	22	6	108	103	L ^b	4, K ^c	6	10	33	35	7	39	34	4	25	24	7	81	81	14	+	3	83	80	0	0	0	
- 6	49	49	8	22	22	+	5	24	22	11	67	67	1	21	18	6, K ^c	0	21	21	+	2	38	37	0	0	0	27		
- 5	61	61	9	19	23	+	1	101	104	12	31	35	10	30	29	+	11	35	35	+	11	35	35	0	0	0	21		
- 14	10	10	16	15	16	0	25	27	L ^b	4, K ^c	1	11	11	1	35	35	+	11	35	35	+	11	35	35	0	0	0	21	
- 3	74	71	1	65	59	59	9	37	37	+	10	22	9	10	37	39	+	9	58	66	66	5	1	30	21				
- 2	138	127	127	L ^b	4, K ^c	4	1	74	77	+	8	82	77	+	9	38	37	+	9	70	77	+	8	38	31	3	17	19	
- 1	85	84	13	26	25	2	52	54	6	27	33	8	80	76	7	15	22	2	7	46	39	L ^b	7, K ^c	3	0	0	0	0	0
- 0	73	70	12	43	38	3	116	115	5	41	39	7	18	14	5	30	30	6	81	73	+	4	25	28	0	0	0		
- 1	83	83	0	11	21	24	4	39	35	+	2	22	17	2	27	29	4	17	20	3	23	24	+	1	83	18	0	0	0
- 2	46	50	10	28	28	6	26	22	+	1	116	105	+	6	28	31	+	3	69	76	+	2	82	83	1	1	41	37	
- 3	104	101	9	66	69																								

when an observed-unobserved cutoff at $2.5 \sigma(I)$ was used. No corrections have been made for absorption or secondary extinction effects.

The structure was solved by direct methods and refined by full-matrix least squares technique. Hydrogen atom positions were calculated; for the methyl group with one H in the plane defined by the three nearest carbon atoms. Anisotropic temperature factors were introduced for the carbons, and weights in least squares were calculated from the standard deviations in intensities, $\sigma(I)$, taken as

$$\sigma(I) = (C_T + (0.02 C_N)^2)^{\frac{1}{2}}$$

where C_T is the total number of counts and C_N is the net count (peak minus background). The conventional R -value arrived at was 6.0 % (weighted value $R_w = 6.4$ %) for 1049 observed reflections. The form factors used were those of Hanson *et al.*² for C-atoms and those of Stewart *et al.*³ for hydrogens.

Table 3. Interatomic distances and angles with estimated standard deviations.

DISTANCE	(Å)	DISTANCE	(Å)	DISTANCE	(Å)
C1 - C2	1.383(4)	C2 - C3	1.381(4)	C3 - C4	1.378(4)
C4 - C5	1.378(4)	C5 - C6	1.379(4)	C6 - C7	1.388(4)
C1 - C7	1.491(3)	C7 - C10	1.585(4)	C7 - C9	1.318(3)
C8 - C9	1.388(3)	C9 - C9	1.478(6)		

ANGLE	(°)	ANGLE	(°)	ANGLE	(°)
C1 - C2 - C3	121.3(3)	C2 - C3 - C4	126.1(3)	C3 - C4 - C5	137.0(4)
C3 - C4 - C5	119.8(3)	C4 - C5 - C6	120.3(3)	C5 - C6 - C7	117.5(3)
C6 - C7 - C1	121.1(3)	C6 - C1 - C2	117.5(3)	C7 - C10 - C10	120.9(2)
C2 - C1 - C7	121.6(2)	C6 - C1 - C7	120.9(2)	C8 - C7 - C10	120.3(3)
C1 - C7 - C10	119.5(2)	C6 - C7 - C10	119.5(2)	C7 - C10 - C9	120.9(3)
C1 - C7 - C8	121.2(2)	C7 - C8 - C9	124.3(3)	C7 - C10 - C9	120.9(3)
C8 - C9 - C9	124.3(3)				

Final fractional coordinates and thermal vibration parameters are given in Table 1. The expression for anisotropic vibration is:

$$\exp [- (B_{11} h^2 + B_{22} k^2 + B_{33} l^2 + B_{12} hk + B_{13} hl + B_{23} kl)]$$

The maximum r.m.s. amplitudes obtained by thermal analysis range from 0.22 Å to 0.28 Å. Rigid-body analysis of translational and librational motion⁴ gave relatively large r.m.s. discrepancy between atomic vibration tensor components calculated from the thermal parameters of Table 1, and those calculated from the rigid-body parameters. By including all atoms, the value obtained was 0.0040 Å². This number does not support the assumption of regarding the molecule as an oscillating rigid-body, and the coordinates were therefore not corrected for librational motion.

A comparison between observed and calculated structure factors is presented in Table 2. Bond distances and angles are listed in Table 3. The numbering of atoms is shown in Fig. 1 where the molecule is viewed along [010].

The bond distances C7=C8, C8=C9, C9-C9', and the angle C8=C9-C9' agree closely with the corresponding values obtained by a recent electron diffraction investigation of the free molecule of 1,2,4,5-hexatetraene:⁵ 1.316 Å, 1.313 Å, 1.471 Å, 124.0°. The three carbon atoms C7, C8, C9 are, within error limits, situated on a straight line, and the bond lengths C1-C7(1.491 Å) and C7-C10(1.505 Å) indicate some degree of conjugation in this part of the molecule.

The phenyl ring is planar to within 0.007 Å, and the torsion angle of 20.6° about C1-C7 gives contacts between H2 and two methyl hydrogens of 2.42 Å and 2.67 Å, respectively. The ring angles have normal values except for C2-C1-C6 which is about 3° smaller. Owing to the electron donating characteristics of the system attached to C1, the effect is what should be expected.⁶

No short inter-molecular contacts are observed.

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