

The Conformation of a Nine-membered Ring

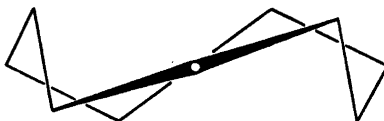
Crystal Structure of the 1 : 1 Addition Compound Mercuric Chloride-Cyclononanone

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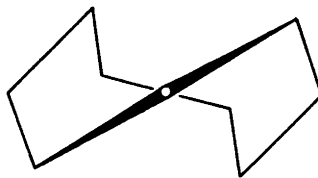
The crystals are monoclinic with space group $P2_1/c$ and cell dimensions $a = 7.58$, Å, $b = 12.21$, Å, $c = 25.14$, Å, and $\beta = 96.2$,°. The unit cell contains eight formula units. The structure was solved by the heavy atom method and refined by full-matrix least squares technique ($R = 11.6$ %, and $R_w = 11.1$ % for 1550 reflections recorded on an automatic four circle diffractometer). Although the standard deviations of light atom parameters are large, it may be stated that the nine-membered ring conformation corresponds to that of a "twisted chair boat" with approximately C_2 symmetry. The charge transfer bonding system consists of infinite chains of alternating $HgCl_2$ and cyclononanone molecules in the [100]-direction.

Strain-energy minimization calculations^{1,2} for the saturated nine-membered ring point out the "twisted boat chair" (TBC), with symmetry D_3 :



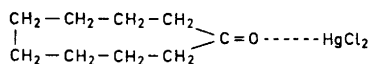
as the energetically favourable conformation. This form has indeed been observed in the crystals of trimeric acetone peroxide,³ where the ring skeleton consists of six oxygens and three carbon atoms.

The only crystal structure analysis reported of a saturated nine-membered ring is that of cyclononylammonium bromide,⁴ where the ring conformation roughly may be described in terms of a "twisted chair boat" (TCB), with a two-fold axis of symmetry:



This conformation has no similarity at all to the TBC-form, and the difference in strain energies amounts to about 3 kcal/mol.

The crystal structure of the 1:1 addition compound mercuric chloride-cyclononanonone has been carried out in order to settle the conformational problem of the cyclic ketone:



The crystals are monoclinic with space group $P2_1/c$. The cell dimensions, determined on a manual four circle diffractometer, with estimated standard deviations ^{5*} are:

$$a = 7.588(2) \text{ \AA}, \quad b = 13.214(3) \text{ \AA}, \quad c = 25.148(6) \text{ \AA}, \quad \beta = 96.26(2)^\circ$$

The unit cell contains eight molecules ($\rho_{\text{calc}} = 2.2 \text{ g/cm}^3$, $\rho_{\text{obs}} \sim 2.0 \text{ g/cm}^3$).

The crystals decompose quickly in air. When exposed to X-rays, the decomposition is rapid even when the crystals are kept in a sealed capillary tube. Out of 14 specimens mounted, three lasted long enough to allow intensity measurements to be carried out. The intensities were recorded on an automatic four circle diffractometer. Using an observed-unobserved cut-off at $2.0 \sigma(I)$, a total number of 1550 reflections were observed. The intensities were corrected for absorption effects.

The structure was solved by the heavy atom method and refined by full-matrix least squares technique. Hydrogen positions were calculated assuming C-H bond lengths of 1.08 Å and included in structure factor calculations, but not refined. Anisotropic temperature factors were introduced for mercury and chloride atoms. The weights in least squares were calculated from the standard deviations in intensities, $\sigma(I)$, taken as

$$\sigma(I) = (C_T + (0.02 C_N)^2)^{\frac{1}{2}}$$

where C_T is the total number of counts, and C_N the net count (peak minus background). Decomposition of the crystals during intensity measurements is probably the main reason for the fact that the weighted R -value arrived at was as large as $R_w = 11.1 \%$ (conventional value, $R = 11.6 \%$) for 1550 observed reflections. The form factors used were those of Hanson *et al.*⁶

Final fractional coordinates and thermal parameters with estimated standard deviations are given in Tables 1 and 2. A comparison between observed and calculated structure factors is presented in Table 3. The principal axes

* All programs used are included in this reference.

Table 1. Fractional atomic coordinates and anisotropic thermal parameters ($\exp[-(B_{11}h^2 + B_{22}k^2 + B_{33}l^2 + B_{12}hk + B_{13}hl + B_{23}kl)]$) for Hg and Cl atoms, with estimated standard deviations.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>B</i> ₁₁	<i>B</i> ₂₂	<i>B</i> ₃₃	<i>B</i> ₁₂	<i>B</i> ₁₃	<i>B</i> ₂₃
Hg ₁	.0357 3	.1413 1	.1778 1	.0130 9	.0067 2	.0019 1	.0014 5	-.0002 4	-.0002 2
Hg ₂	.5347 3	.1289 1	.1747 1	.0169 9	.0064 2	.0024 1	-.0001 5	.0000 5	.0013 2
Cl ₁	.224 2	.196 1	.2486 6	.018 5	.007 1	.0023 4	.000 3	.002 2	-.0009 9
Cl ₂	-.152 2	.094 1	.1052 6	.015 4	.011 1	.0021 4	.003 3	-.001 2	-.002 1
Cl ₃	.717 2	.064 1	.2431 6	.021 5	.010 1	.0027 5	.011 4	.003 2	.005 1
Cl ₄	.345 2	.188 1	.1050 6	.017 4	.009 1	.0023 4	.009 4	-.002 2	.001 1

of the thermal vibration ellipsoids for Hg and Cl atoms were calculated from the thermal parameters of Table 1. Root mean square amplitudes, the corresponding *B*-values for the atomic anisotropic thermal vibration along the principal axes, as well as their components along the crystal axes are given in Table 4.

Table 2. Fractional atomic coordinates and isotropic temperature factors for oxygen and carbon atoms, with estimated standard deviations.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>B</i>
O ₁	.765 (5)	.292 (2)	.188 (1)	4.5 (0.8)
O ₂	.735 (5)	.471 (2)	.318 (2)	5.8 (0.8)
C ₁	.757 (6)	.383 (3)	.186 (2)	1.9 (0.9)
C ₂	.596 (7)	.433 (3)	.180 (2)	4.0 (1.2)
C ₃	.572 (7)	.518 (3)	.133 (2)	5.0 (1.3)
C ₄	.523 (7)	.470 (3)	.079 (2)	4.0 (1.2)
C ₅	.667 (9)	.398 (4)	.056 (2)	6.4 (1.5)
C ₆	.846 (9)	.447 (4)	.047 (3)	7.5 (1.7)
C ₇	.972 (9)	.487 (4)	.088 (3)	7.5 (1.7)
C ₈	1.026 (8)	.424 (4)	.137 (3)	6.9 (1.5)
C ₉	.940 (7)	.436 (3)	.189 (2)	3.8 (1.1)
C ₁₀	.743 (9)	.380 (4)	.332 (2)	6.1 (1.5)
C ₁₁	.585 (8)	.317 (4)	.326 (2)	5.8 (1.4)
C ₁₂	.458 (8)	.342 (3)	.370 (2)	4.8 (1.3)
C ₁₃	.525 (9)	.296 (4)	.428 (3)	8.1 (1.8)
C ₁₄	.623 (9)	.363 (4)	.469 (3)	7.4 (1.6)
C ₁₅	.812 (9)	.405 (4)	.451 (2)	6.9 (1.5)
C ₁₆	.938 (9)	.334 (4)	.451 (2)	7.4 (1.6)
C ₁₇	.949 (9)	.273 (4)	.393 (3)	8.5 (1.8)
C ₁₈	.931 (8)	.333 (4)	.338 (2)	5.5 (1.3)

Table 3. Observed and calculated structure factors on ten times absolute scale.

(a)

<i>h</i>	<i>k</i>	<i>l</i>	<i>F</i> _o	<i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	<i>F</i> _o	<i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	<i>F</i> _o	<i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	<i>F</i> _o	<i>F</i> _c
0	0	2	4886	-4480	0	6	7	3573	3723	1	10	-2	290	213	1	5	0	773	715
0	0	4	2243	-2667	0	6	8	1029	-1126	1	10	-4	376	375	1	5	1	1085	-1110
0	0	6	3822	4027	0	6	9	2174	-2241	1	10	-5	434	477	1	5	2	288	-316
0	0	8	3794	-4089	0	6	11	722	-803	1	10	-6	622	-665	1	5	3	391	-441
0	0	12	3304	3342	0	6	12	744	761	1	10	-7	487	-397	1	5	5	578	673
0	0	14	4166	-4335	0	6	13	1900	1962	1	10	-8	331	-428	1	5	7	1270	1330
0	0	16	1562	1585	0	6	14	712	-696	1	10	-10	355	205	1	5	9	764	-804
0	0	18	1188	840	0	6	15	1234	-1158	1	10	-11	347	325	1	5	12	483	572
0	0	20	1567	-1334	0	6	16	335	293	1	10	-12	420	-495	1	5	13	402	-393
0	0	22	1056	947	0	6	19	1135	947	1	9	-15	334	294	1	5	14	442	-416
0	1	18	543	-1286	0	9	20	419	-526	1	9	-18	304	-175	1	5	18	362	333
0	1	17	923	1637	0	7	19	828	-808	1	9	-11	573	-490	1	4	19	366	188
0	1	16	1484	1879	0	7	17	1385	1283	1	9	-9	805	922	1	4	17	348	436
0	1	15	1561	-1781	0	7	16	475	-429	1	9	-8	292	212	1	4	15	564	-564
0	1	13	346	-601	0	7	15	782	-781	1	9	-7	295	-320	1	4	9	621	594
0	1	12	2277	-2393	0	7	13	674	-655	1	9	-5	307	-347	1	4	8	377	296
0	1	11	2170	2438	0	7	12	454	380	1	9	-3	496	525	1	4	7	685	697
0	1	10	2754	3046	0	7	11	2099	2119	1	9	0	284	108	1	4	6	398	511
0	1	9	1831	-2125	0	7	10	924	-925	1	9	1	326	-304	1	4	5	654	-756
0	1	8	821	-1011	0	7	9	2443	-2540	1	9	3	935	973	1	4	3	1272	-1212
0	1	7	475	569	0	7	8	776	871	1	9	4	260	245	1	4	2	263	-257
0	1	6	4166	-4335	0	7	7	265	333	1	9	5	761	-786	1	4	1	1278	1323
0	1	5	2345	2524	0	7	5	2640	2751	1	9	9	549	568	1	4	2	911	243
0	1	4	4166	4303	0	7	4	1247	-1296	1	9	11	314	-448	1	4	1	711	-860
0	1	3	3684	-3891	0	7	3	3103	-3290	1	9	15	504	400	1	4	0	255	-104
0	1	2	3852	-3816	0	7	2	1050	1067	1	8	15	444	396	1	4	-1	413	400
0	2	1	4814	-4701	0	7	1	1304	1377	1	8	13	482	-379	1	4	-2	717	296
0	2	2	484	546	0	8	0	1847	1840	1	8	9	294	268	1	4	-11	711	669
0	2	3	1840	1800	0	8	1	1012	-1072	1	8	7	790	-747	1	4	-12	447	458
0	2	5	2901	3060	0	8	2	1085	-1056	1	8	6	713	-763	1	4	-13	438	-401
0	2	7	2068	-2250	0	8	3	432	377	1	8	5	755	785	1	4	-16	321	-430
0	2	9	1624	1765	0	8	4	876	-875	1	8	2	446	-450	1	4	-20	402	260
0	2	11	1123	1265	0	8	5	831	904	1	8	1	729	-748	1	4	-21	374	457
0	2	13	3479	-3645	0	8	6	2460	2556	1	8	0	369	-339	1	3	-21	419	-289
0	2	15	2162	2119	0	8	7	1486	-1508	1	8	-1	545	544	1	3	-19	392	274
0	2	19	1294	-1169	0	8	8	2042	-2081	1	8	-5	252	-300	1	3	-17	399	-358
0	2	21	1030	873	0	8	9	826	760	1	8	-6	364	-384	1	3	-16	402	260
0	3	22	427	342	0	8	11	245	210	1	8	-7	635	615	1	3	-15	392	460
0	3	21	345	342	0	8	12	1049	1059	1	8	-8	358	262	1	3	-14	1066	-1112
0	3	19	642	659	0	8	13	603	-579	1	8	-9	289	-232	1	3	-12	635	819
0	3	18	959	-944	0	8	14	1133	-1009	1	8	-11	362	-331	1	3	-11	367	-254
0	3	17	1639	-1645	0	8	15	594	611	1	8	-12	434	-359	1	3	-10	365	448
0	3	16	1284	1308	0	8	16	331	257	1	8	-13	593	442	1	3	-9	776	356
0	3	15	1394	1509	0	9	14	350	-323	1	7	-17	397	248	1	3	-7	682	710
0	3	14	347	-459	0	9	12	930	-789	1	7	-16	357	-298	1	3	-6	1205	-1221
0	3	13	632	716	0	9	11	436	318	1	7	-14	532	463	1	3	-5	1082	-1144
0	3	12	1096	-1285	0	9	10	1881	1924	1	7	-11	299	208	1	3	-4	873	750
0	3	11	2594	-2879	0	9	9	529	-417	1	7	-10	635	-652	1	3	-3	419	459
0	3	10	2120	2296	0	9	8	1043	-1078	1	7	-8	616	529	1	3	-2	342	-302
0	3	9	2483	2786	0	9	7	315	180	1	7	-6	428	436	1	3	-1	901	-961
0	3	8	777	-829	0	9	6	1025	-1012	1	7	-5	290	258	1	3	0	2198	2078
0	3	7	496	-414	0	9	5	321	240	1	7	-4	582	-643	1	3	1	1242	1187
0	3	6	748	-847	0	9	4	2107	2130	1	7	-3	331	-307	1	3	2	653	-569
0	3	5	1284	1308	0	9	3	521	-450	1	7	-2	578	669	1	3	3	333	-369
0	3	4	2425	2519	0	9	2	1463	-1503	1	7	0	676	-693	1	3	6	221	-132
0	3	3	4651	4683	0	9	1	351	217	1	7	2	464	-567	1	3	8	747	803
0	3	2	3431	-3405	0	10	0	891	-877	1	7	3	330	-288	1	3	9	532	529
0	3	1	3814	-3709	0	10	1	1215	-1298	1	7	4	1136	1181	1	3	10	922	-1040
0	4	0	5057	-5352	0	10	2	719	657	1	7	5	354	333	1	3	11	449	-457
0	4	1	1284	1282	0	10	3	272	264	1	7	6	683	-803	1	3	12	632	763
0	4	2	2508	2666	0	10	5	1159	1268	1	7	9	321	-277	1	3	14	456	-292
0	4	4	1208	1408	0	10	6	1229	-1185	1	7	10	586	574	1	3	15	366	-334
0	4	5	484	-485	0	10	7	1588	-1644	1	7	12	311	-245	1	3	18	359	163
0	4	6	3959	-4321	0	10	8	1023	999	1	7	13	351	-328	1	2	21	371	185
0	4	7	1075	1097	0	10	9	597	565	1	7	16	385	327	1	2	17	325	-450
0	4	8	3082	3323	0	10	11	239	471	1	7	18	496	-352	1	2	16	470	-496
0	4	9	859	-760	0	10	12	439	-451	1	6	16	507	501	1	2	14	355	343
0	4	12	1948	-2155	0	10	13	864	-724	1	6	14	488	-450	1	2	11	627	-577
0	4	13	727	830	0	11	9	1109	1118	1	6	11	340	427	1	2	10	588	548
0	4	14	2619	2646	0	11	5	1274	-1313	1	6	8	668	-709	1	2	9	396	554
0	4	15	809	-667	0	11	4	359	71	1	6	7	520	-412	1	2	7	346	322
0	4	16	752	-787	0	11	3	1442	1513	1	6	6	728	784	1	2	6	668	-758
0	4	18	1015	-930	0	11	1	636	-579	1	6	5	381	455	1	2	5	346	-393
0	4	20	1396	1220	0	12	0	1045	-1077	1	6	4	396	320	1	2	2	1290	1210
0	5	18	1316	1132	0	12	1	750	722	1	6	3	326	255	1	2	4	707	-577
0	5	17	608	-563	0	12	2	667	685	1	6	2	1136	-1173	1	2	5	243	127
0	5	16	1435	-1451	0	12	3	366	-289	1	6	1	374	-418	1	2	7	293	247
0	5	15	716	656	1	12	0	375	393	1	6	0	348	446	1	2	8	544	558
0	5	14	323	315	1	12	0	465	542	1	6	-6	692	752	1	2	-9	437	-498
0	5	12	1925	1941	1	12	-1	286	-342	1	6	-7	258	333	1	2	-12	393	-323
0	5	11	1147	-1174	1	11	-8	808	-797	1	6	-8	676	-945	1	2	-13	312	-315
0	5	10	2868	-2907	1	11	-6	899	262	1	6	-10	861	-807	1	2	-15	622	-622
0																			

Table 3. Continued.

<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>
1	1	4	307	366															
1	1	6	563	530															
1	1	7	1548	-1645															
1	1	8	860	-917															
1	1	9	527	534															
1	1	12	402	-306															
1	1	15	409	388															
1	1	21	479	308															
1	0	22	404	440															
1	0	16	404	441															
1	0	14	450	-576															
1	0	12	503	-390															
1	0	10	486	680															
1	0	8	566	-589															
1	0	-6	290	-298															
1	0	-10	279	343															
1	0	-12	287	-145															
0	-22	490	386																
0	-20	201	-186																
2	0	-18	1895	1850															
2	0	-14	1495	-1609															
2	0	-12	1907	1881															
2	0	-10	1183	-1121															
2	0	-8	2070	-2125															
2	0	-6	5590	5633															
2	0	-4	5734	-5053															
2	0	-2	376	-393															
2	0	0	4912	4304															
2	0	2	3178	-3000															
2	0	4	352	3450															
2	0	6	5174	-5178															
2	0	10	2142	2119															
2	0	12	1673	1732															
2	0	14	1798	-1758															
2	0	16	1247	1193															
2	0	20	1614	-1393															
2	1	18	872	-1134															
2	1	17	707	988															
2	1	16	771	774															
2	1	13	799	-1160															
2	1	12	2251	-2240															
2	1	11	2491	2517															
2	1	10	3088	2934															
2	1	9	1885	-1927															
2	1	8	410	-474															
2	1	7	1057	1082															
2	1	6	2731	-2855															
2	1	5	2740	2863															
2	1	4	3034	2944															
2	1	3	1554	-1632															
2	1	2	655	-660															
2	1	1	264	-233															
2	1	0	978	-1258															
2	1	-1	1811	2031															
2	1	-2	3511	3581															
2	1	-3	3628	-3527															
2	1	-4	4416	-3977															
2	1	-5	1998	1912															
2	1	-6	669	-712															
2	1	-7	2084	2038															
2	1	-8	3309	3281															
2	1	-9	2671	-2634															
2	1	-10	2184	-2093															
2	1	-11	1121	1017															
2	1	-14	1153	1131															
2	1	-15	1426	-1325															
2	1	-16	1899	-1789															
2	1	-17	1153	1321															
2	1	-18	783	1055															
2	1	-21	634	-692															
2	1	-22	741	-938															
2	1	-21	1221	-980															
2	1	-19	2179	1970															
2	1	-17	1005	-916															
2	1	-15	1252	-1149															
2	1	-14	350	381															
2	1	-13	2552	2393															
2	1	-11	2177	-2150															
2	1	-10	282	313															
2	1	-9	337	-495															
2	1	-7	4702	4648															
2	1	-6	850	-860															
2	1	-5	5716	-5834															
2	1	-4	576	269															
2	1	-3	802	1060															
2	1	-1	2596	3063															
2	2	0	468	-463															
2	2	1	3086	-3657															
2	2	2	329	268															
2	2	3	2397	2554															
2	2	5	905	953															
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2	2	-18	1416	-1327															
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2	2	-7	898	-865															
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2	2	12	1162	-1191															
2	2	14	2162	2021															
2	2	16	1526	-1457															
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2	2	19	453	496															
2	2	18	1488	1314															
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2	5	-12	719	-706															
2	5	-13	418	-263															
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2	5	-7	1802	-1901															
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2	5	-3	767	-661															
2	5	-1	1802	-1901															
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2	6	8	1107	-1122															
2	6	9	1776	-1852															
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2	6	13	1705	1684															
2	6	14	699	-649															
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2	6	-6	306	-257															
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2	7	-14	391	-364															
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2	7	-13	1105	1048															
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2	7	-11	612	-605															
2	7	-10	768	-781															
2	7	-8	718	-704															
2	7	-7	803	793															
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2	7	-5	893	-941															
2	7	-4	1039	-1118															
2	7	-3	325	260															
2	7	-2	475	-522															
2	7	-1	741	624															
2	7	0	2296	2420															
2	7	1	1533	-1543															
2	7	2	2369	-2438															
2	7	3	921	963															
2	7	4	483	480															
2	7	5	384	364															
2	7	6	1120	1139															
2	7	7	842	-773															
2	7	8	1375	-1366															
2	7	9	751	657															
2	7	10	606	636															
2	7	12	648	633															

Table 3. Continued.

<i>h</i>	<i>k</i>	<i>l</i>	<i>F</i> ₀	<i>F</i> _C	<i>h</i>	<i>k</i>	<i>l</i>	<i>F</i> ₀	<i>F</i> _C	<i>h</i>	<i>k</i>	<i>l</i>	<i>F</i> ₀	<i>F</i> _C	<i>h</i>	<i>k</i>	<i>l</i>	<i>F</i> ₀	<i>F</i> _C
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3	3	-11	636	-631	3	5	4	333	-354	3	6	0	703	749	3	8	-4	469	457
3	3	-12	418	-390	3	5	3	1307	-1265	3	6	1	421	-456	3	8	-3	604	588
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3	4	-20	385	348	3	5	-1	932	917	3	6	10	765	827	3	8	1	806	-821
3	4	-13	332	-347	3	5	-2	308	-375	3	6	14	542	-580	3	8	2	463	448
3	4	-10	282	136	3	5	-5	883	-879	3	6	16	340	243	3	8	3	445	494
3	4	-9	555	651	3	5	-6	269	328	3	7	12	498	-473	3	8	9	406	416
3	4	-5	493	600	3	5	-7	523	533	3	7	8	279	93	3	8	12	358	-332
3	4	-4	332	-304	3	5	-8	321	-275	3	7	4	722	784	3	9	11	563	-590
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3	4	-1	552	-572	3	5	-10	615	626	3	7	2	634	-620	3	9	5	570	-518
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3	4	7	340	423	3	6	-17	388	-380	3	7	-6	491	-575	3	9	-9	569	607
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3	4	18	429	434	3	6	-15	359	375	3	7	-10	342	-376	3	10	-6	403	-387
3	5	17	547	559	3	6	-14	412	-395	3	7	-13	364	-137	3	10	0	550	-508
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3	5	8	504	-543	3	6	-4	926	-946	3	8	-11	526	-482	3	11	-2	605	-619
3	5	7	361	-278	3	6	-3	506	611										

(c)

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4	2	-4	1304	227	4	3	3	1045	-815	4	5	5	1472	1517	5	3	5	263	305
4	2	-3	1632	1738															

Table 3. Continued.

<i>h</i>	<i>k</i>	<i>l</i>	<i>F</i> ₀	<i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	<i>F</i> ₀	<i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	<i>F</i> ₀	<i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	<i>F</i> ₀	<i>F</i> _c	
5	4	-8	410	362	6	2	4	854	-252	1	8	19	445	-367	4	10	6	639	693	
5	4	-5	379	490	6	2	5	859	-951	1	8	18	389	-296	4	10	9	824	928	
5	4	-2	305	-210	6	2	7	682	-632	1	6	-21	411	77	4	11	3	691	517	
5	4	-1	421	-463	6	2	9	1628	1523	1	5	21	454	-464	4	11	1	333	427	
5	4	0	307	382	6	3	9	621	-576	1	3	-23	407	-119	4	11	0	411	-85	
5	4	2	266	112	6	3	8	876	679	1	1	-25	379	-159	4	11	-1	125	-1224	
5	4	3	522	-574	6	3	7	1495	1608	2	0	-24	434	591	4	11	-2	486	129	
5	4	5	718	743	6	3	6	1125	-947	2	0	22	1120	1366	4	11	-3	801	1085	
5	5	11	548	581	6	3	5	1287	-1321	2	2	23	653	-863	4	11	-4	287	-93	
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6	1	1	1413	-1394	0	9	18	894	-880	4	6	-17	602	881	6	7	4	540	-430	
6	1	0	1839	-1978	0	9	16	818	851	4	6	15	1025	-1105	6	7	2	291	-137	
6	1	-1	1874	1978	0	10	14	490	534	4	6	17	712	774	6	7	1	947	-1093	
6	1	-2	1743	1798	0	10	15	540	555	4	7	13	1173	-1424	6	7	0	605	438	
6	1	-3	994	-853	0	10	16	417	-332	4	7	-15	735	-1103	6	7	-1	1109	1195	
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6	1	-5	976	-669	0	12	5	498	510	4	7	-17	506	762	6	7	-3	443	-354	
6	1	-6	1437	-1336	0	12	6	788	-843	4	8	-14	327	-445	6	7	-4	457	-183	
6	1	-7	1353	1237	0	12	7	879	965	4	8	-13	445	505	6	7	-5	712	-735	
6	1	-8	1189	1138	0	12	8	702	736	4	8	11	683	-662	6	7	-6	686	494	
6	1	-9	815	-646	0	12	9	405	-508	4	8	14	640	-850	6	7	-7	1337	1430	
6	1	-10	848	-817	0	13	5	359	85	4	9	12	1							

Table 4. The principal axes of the thermal vibration ellipsoids given by the components of a unit vector in fractional coordinates e_x , e_y , e_z , the corresponding r.m.s. amplitudes, and the B -values.

Atom	e_x	e_y	e_z	$(\bar{u}^2)^{\dagger}$ (Å)	B (Å)
Hg ₁	.030	.046	.029	.252	5.00
	-.003	.059	.025	.239	4.52
	.129	-.010	.012	.190	2.85
Hg ₂	-.020	.030	.036	.288	6.53
	.056	.065	-.010	.229	4.13
	.119	-.026	.015	.217	3.72
Cl ₁	.011	-.040	.034	.276	6.00
	.007	.064	.021	.244	4.71
	.132	.000	.000	.228	4.10
Cl ₂	.019	.067	-.018	.323	8.25
	-.019	.036	.034	.245	4.75
	.130	-.004	.012	.200	3.16
Cl ₃	.043	.055	.026	.363	10.38
	.085	.023	-.026	.255	5.13
	.092	-.047	.018	.198	3.08
Cl ₄	.069	.060	-.011	.299	7.04
	-.008	.032	.036	.281	6.24
	.113	-.034	.014	.181	2.58

Bond distances and angles may be found in Figs. 1 and 2. Estimated standard deviations in bond distances and angles involving the light atoms only are about 0.07 Å and 5°, respectively. The asymmetric unit contains two independent cyclononane molecules with enantiomeric conformations of the TCB-form. The approximate two-fold axes of symmetry are indicated in Fig. 1. In view of the large standard deviations, the considerable variations in bond distances and angles of the nine-membered rings are in fact insignificant and will not be discussed further. The C—C bond distances and the C—C—C bond angles average to 1.533 Å and 115.7°, the corresponding mean values for cyclononylammonium bromide⁴ being 1.532 Å and 116.7°, respectively.

The carbonyl groups occupy positions corresponding to the shortest transannular H···H contacts in the TBC-form of cyclononane. In the present compound, such contacts range from 1.9 to 2.7 Å (1.8 to 2.4 Å in cyclononylammonium bromide).

Fig. 3 shows dihedral angles (averaged for the two isomers) of the present compound (A) and cyclononylammonium bromide (B), compared with those calculated by Hendrickson² (C), and Bixon and Lifson¹ (D) for the TCB-conformation. Apart from a somewhat larger eclipsing at the carbonyl group, the agreement is satisfactory. Standard deviations in dihedral angles are about 6°.

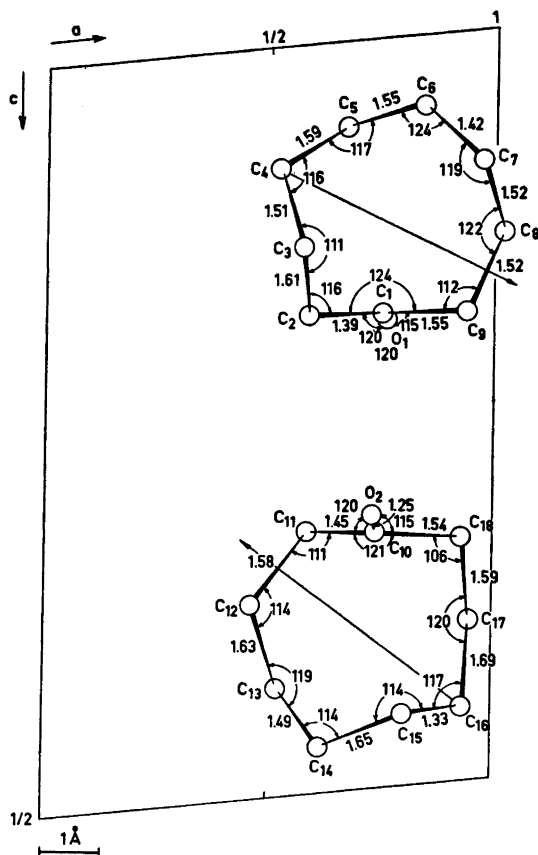


Fig. 1. Schematical drawing (viewed along [010]) showing bond distances and angles of the cyclononane molecules.

When oxygen acts as a donor in charge-transfer complexes, only one of the lone pairs is usually engaged in the bond. Among the exceptions is the 1 : 1 complex between acetone and bromine,⁷ where the $\text{Br} \cdots \text{O} \cdots \text{Br}$ angle is 110° . From Fig. 2, showing the charge-transfer binding system, it is clear that the present structure constitutes another example, the $\text{Hg}_1 \cdots \text{O}_1 \cdots \text{Hg}_2$ and $\text{Hg}_1 \cdots \text{O}_2 \cdots \text{Hg}_2$ angles being 84.1° and 82.0° , respectively. Each mercury atom is complexed with two independent cyclononane molecules, thus forming infinite chains in the [100]-direction. The mean value of the four charge-transfer bonds is 2.87 Å. The arrangement around the Hg-atoms is widely different from what has been observed for the complex $\text{HgCl}_2 \cdot 2(\text{C}_6\text{H}_5)_3\text{AsO}$ ⁸ and the 1 : 1 addition compound between mercuric chloride and cyclohexane-1,4-dione,⁹ but resembles that of $\text{HgCl}_2 \cdot 1,4$ -dioxane.¹⁰

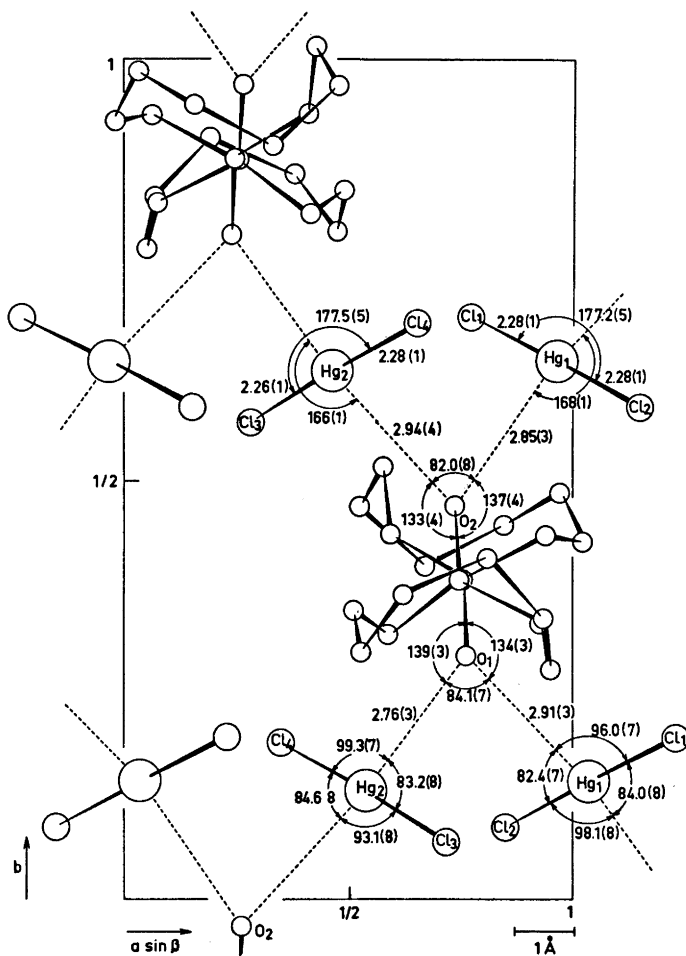


Fig. 2. Schematic drawing (viewed along [001]) showing the charge-transfer bonding system.

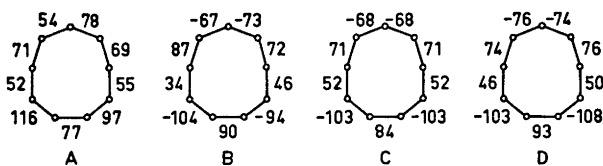


Fig. 3. Dihedral angles; of the present compound (A), cyclononylammonium bromide (B), and calculated² (C) and¹ (D).

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