

Crystal Structure of Bisethylene-1,2-bis(methylphenylphosphine)palladium(II) Chloride

P. GROTH

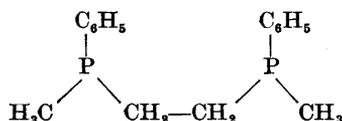
Department of Chemistry, University of Oslo, Oslo 3, Norway

The crystals belong to the orthorhombic system with space group *Pccn* and cell dimensions

$$a = 19.09, \text{ \AA}, b = 15.71, \text{ \AA}, c = 11.84, \text{ \AA}$$

The cell contains four formula units. 1509 reflections were registered on an automatic four circle diffractometer. The structure was solved by the heavy atom method and refined by full-matrix least squares technique to $R = 7.9\%$ ($R_w = 6.0\%$). The complex crystallizes with half a molecule of ethanol per asymmetric unit. The Pd-P distances are 2.33, \AA and the mean value of the six P-C distances 1.81, \AA. The complex may be described in terms of a MESO-*anti-trans* form, and the Cl⁻ ions are situated at distances 4.09, \AA from Pd.

The two stereoisomeric complexes formed by ruthenium(III) chloride with *meso*-ethylene-1,2-bis-(methyl-phenylphosphine):



have been studied by Horner *et al.*¹ A series of analogous complexes with other metals (Co, Pt, Pd, Ni) have been synthesized by Müller.^{2,3}

Since the structural information, that can be obtained by conventional methods (IR, PMR, *etc.*), is somewhat limited, an X-ray analysis of the Pd-complex:



has been undertaken in order to clarify some of the stereochemical problems concerning these complexes.

Table 1. Fractional atomic coordinates with estimated standard deviations (multiplied by 10^5).^a Isotopic temperature factors for oxygen, carbon and hydrogen.^b

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>B</i> (Å ²)
Pd	50000	0	50000	
	0	0	0	
Cl	35090	-00134	74789	
	14	24	28	
P ₁	38933	01896	42028	
	15	17	29	
P ₂	49377	14430	54105	
	18	17	25	
C ₁	39624	03624	27041	2.79
	63	63	97	25
C ₂	45993	04287	22215	2.81
	59	66	102	26
C ₃	46523	06054	10937	3.38
	61	70	110	27
C ₄	40712	07075	04169	4.33
	69	75	110	31
C ₅	34132	06273	08933	4.12
	67	75	113	31
C ₆	33516	04568	20615	3.19
	61	68	108	28
C ₇	35564	11825	47978	2.96
	54	65	95	25
C ₈	41332	18439	47221	3.31
	58	69	101	28
C ₉	48192	16651	68974	4.07
	58	73	105	31
C ₁₀	31911	-05521	44199	3.60
	59	70	106	29
C ₁₁	56313	21189	49054	2.35
	50	62	104	23
C ₁₂	59364	19468	38706	2.98
	58	66	103	27
C ₁₃	64712	24741	34448	4.14
	61	82	104	30
C ₁₄	66787	31701	40869	4.20
	62	76	112	31
C ₁₅	63864	33448	51107	4.18
	60	73	120	28
C ₁₆	58655	28200	55060	3.34
	60	69	103	27
C _e	28605	22850	75328	4.89
	59	75	113	34
O _e	29532	17651	84111	4.41
	77	97	142	39
H ₂	51072	04924	25844	4.0
H ₃	51168	07085	05166	3.4
H ₄	40716	07812	-03161	4.0
H ₅	30391	07406	04921	5.6
H ₆	29210	03554	23922	2.6
H ₁₂	58054	14549	34637	2.2
H ₁₃	66169	23167	29249	3.8
H ₁₄	69389	34376	37938	4.3
H ₁₆	64872	37860	55868	6.4
H ₁₆	57022	29056	61876	3.0
H _{7,1}	32357	13599	44491	3.2

Table 1. Continued.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>B</i> (Å ²)
H _{7,3}	35412	09910	54823	3.0
H _{8,1}	40703	22484	51809	4.7
H _{8,2}	43224	19716	39254	2.5
H _{8,3}	46148	12972	71656	3.3
H _{9,1}	46577	22469	69083	2.4
H _{9,2}	53135	15683	72544	2.8
H _{9,3}	31698	-06694	51060	1.0
H _{10,1}	32952	-10622	40561	1.4
H _{10,2}	27821	-03030	41047	1.1

^a For numbering of the atoms, see Fig. 1.

^b The hydrogen atoms H_m (or H_{m,n}) are bonded to C_m. Estimated standard deviations are about 10 times those of the carbon atoms.

Table 2. (a) Anisotropic thermal vibration parameters for Pd, P, and Cl with estimated standard deviations (multiplied by 10⁴).

Atom	<i>B</i> ₁₁	<i>B</i> ₂₂	<i>B</i> ₃₃	<i>B</i> ₁₂	<i>B</i> ₁₃	<i>B</i> ₂₃
Pd	151	264	671	3	-123	-36
	3	5	11	11	16	17
Cl	221	450	773	100	-9	100
	9	14	30	29	31	47
P ₁	109	286	668	1	-72	34
	9	16	32	20	30	37
P ₂	174	250	612	2	-138	-80
	10	13	32	24	43	30

(b) The principal axes of the thermal vibration ellipsoids given by the components of a unit vector in fractional coordinates *e*_{*x*}, *e*_{*y*}, *e*_{*z*}; the corresponding r.m.s. amplitudes, and the *B*-values.

Atom	<i>e</i> _{<i>x</i>}	<i>e</i> _{<i>y</i>}	<i>e</i> _{<i>z</i>}	(\bar{u}^2) ^{1/2} (Å)	<i>B</i> (Å ²)
Pd	.016	.006	-.080	.244	3.96
	-.003	.063	.007	.181	2.60
	.050	.002	.026	.160	2.02
Cl	.015	.051	.044	.249	4.89
	.016	.028	-.071	.230	4.16
	.048	-.025	.011	.194	2.96
P ₁	-.008	.008	.083	.220	3.81
	.002	.063	-.010	.189	2.81
	.052	-.001	.013	.140	1.54
P ₂	.023	.013	-.074	.220	3.81
	.025	-.056	.003	.177	2.48
	.040	.028	.040	.165	2.16

Table 3. Observed and calculated structure factors on ten times absolute scale. The columns are h, k, l, F_o and F_c .

0	0	2	2473	3112	1	2	2	618	676	2	1	4	560	-577	2	13	4	413	-472			
0	0	4	3581	3506	1	2	4	273	237	2	1	7	211	279	2	14	4	490	499			
0	0	6	384	-345	1	2	4	492	483	2	1	8	273	-278	2	14	1	342	-423			
0	0	8	894	975	1	2	6	275	-272	2	1	9	569	-740	2	14	2	1084	1100			
0	0	10	551	489	1	2	7	540	-517	2	1	11	346	-129	2	14	4	538	540			
0	1	2	189	166	1	2	11	260	-271	2	2	1	1320	1254	2	14	6	264	263			
0	1	4	618	-546	1	2	1	1516	-1545	2	2	1	509	-512	2	14	4	520	569			
0	1	6	1108	-1007	1	2	1	435	652	2	2	2	808	459	2	14	7	302	257			
0	2	0	3061	3083	1	2	1	480	-419	2	2	2	840	-839	2	14	8	331	223			
0	2	2	235	162	1	2	1	1221	1230	2	2	4	907	496	2	14	1	294	-276			
0	2	4	2117	2054	1	2	4	195	169	2	2	5	306	-282	2	14	7	399	225			
0	2	6	959	-874	1	2	5	285	246	2	2	6	833	840	2	14	7	401	395			
0	2	8	779	742	1	2	7	444	611	2	2	8	804	846	2	14	1	307	122			
0	2	10	544	444	1	2	9	693	616	2	2	10	678	565	2	14	4	333	252			
0	2	12	257	616	1	2	11	800	691	2	2	12	353	421	2	14	4	334	197			
0	2	14	455	306	1	2	13	690	672	2	2	1	551	538	2	14	6	463	349			
0	2	16	1628	-1645	1	2	14	1432	1436	2	2	3	1554	1465	2	20	0	498	376			
0	2	18	413	-295	1	2	1	4	53	600	2	2	4	263	-232	3	0	2	208	-208		
0	2	20	216	-227	1	2	1	4	507	497	2	2	5	41	-115	3	0	4	840	-875		
0	2	22	253	-176	1	2	1	4	367	-311	2	2	7	245	280	3	0	6	339	319		
0	2	24	323	3358	1	2	1	4	267	-230	2	2	9	637	-630	3	0	8	217	-191		
0	2	26	768	-691	1	2	1	4	272	-251	2	2	11	588	624	3	0	10	651	545		
0	2	28	1584	1599	1	2	1	4	768	-778	2	2	13	1275	1265	3	1	1	1768	1728		
0	2	30	335	-370	1	2	1	1	2203	2258	2	2	15	1824	1791	3	1	3	301	-374		
0	2	32	1219	1153	1	2	1	4	364	353	2	2	17	542	561	3	1	5	1400	1421		
0	2	34	1051	407	1	2	1	4	1239	1225	2	2	19	665	671	3	1	7	210	-138		
0	2	36	12	1047	865	1	2	1	4	189	-182	2	2	21	254	242	3	1	9	1572	1656	
0	2	38	323	-363	1	2	1	4	897	861	2	2	23	1190	1136	3	1	11	939	1053		
0	2	40	888	366	1	2	1	4	635	607	2	2	25	864	798	3	1	13	218	63		
0	2	42	312	343	1	2	1	4	7	663	670	2	2	27	925	873	3	1	15	323	461	
0	2	44	382	3132	1	2	1	4	836	782	2	2	29	530	481	3	1	17	337	406		
0	2	46	1020	1060	1	2	1	4	286	-251	2	2	31	844	-833	3	1	19	441	-461		
0	2	48	1691	1746	1	2	1	4	1	678	586	2	2	33	624	602	3	1	21	554	523	
0	2	50	358	335	1	2	1	11	490	347	2	2	35	862	864	3	1	23	306	286		
0	2	52	818	785	1	2	1	1	1316	1311	2	2	37	541	500	3	1	25	840	-875		
0	2	54	349	332	1	2	1	1	4	532	541	2	2	39	306	-359	3	1	27	237	-253	
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0	2	62	282	-264	1	2	1	7	1	384	-394	2	2	47	1837	1919	3	1	35	250	-2496	
0	2	64	270	-228	1	2	1	7	2	2401	2499	2	2	49	333	319	3	1	37	451	-401	
0	2	66	1021	1075	1	2	1	7	2	444	493	2	2	51	1598	1656	3	1	39	460	-435	
0	2	68	489	554	1	2	1	7	7	1279	1358	2	2	53	1358	1344	3	1	41	1149	1145	
0	2	70	1507	1668	1	2	1	7	4	244	237	2	2	55	429	450	3	1	43	360	-295	
0	2	72	433	418	1	2	1	7	4	804	753	2	2	57	1626	967	3	1	45	977	1026	
0	2	74	728	687	1	2	1	7	8	292	200	2	2	59	262	277	3	1	47	624	601	
0	2	76	585	500	1	2	1	7	11	310	113	2	2	61	414	349	3	1	49	1009	1245	
0	2	78	232	-280	1	2	1	8	1	194	197	2	2	63	386	390	3	1	51	720	775	
0	2	80	371	-337	1	2	1	8	2	353	339	2	2	65	621	-601	3	1	53	143	-118	
0	2	82	558	595	1	2	1	8	1	458	-446	2	2	67	820	810	3	1	55	620	564	
0	2	84	659	681	1	2	1	8	4	311	422	2	2	69	4	246	311	3	1	57	601	543
0	2	86	811	801	1	2	1	8	7	601	-610	2	2	71	419	-389	3	1	59	184	130	
0	2	88	369	325	1	2	1	8	8	302	-340	2	2	73	268	222	3	1	61	391	419	
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0	2	92	425	432	1	2	1	8	9	419	-423	2	2	77	964	1011	3	1	65	246	258	
0	2	94	723	582	1	2	1	8	1	1392	1440	2	2	79	611	-595	3	1	67	257	-298	
0	2	96	709	781	1	2	1	8	1	906	932	2	2	81	1526	1615	3	1	69	792	-766	
0	2	98	4	735	791	1	2	1	8	6	645	661	2	2	83	250	-241	3	1	71	1654	1701
0	2	100	409	421	1	2	1	8	6	438	396	2	2	85	777	788	3	1	73	1336	1347	
0	2	102	265	-116	1	2	1	8	7	418	629	2	2	87	211	220	3	1	75	218	213	
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0	2	108	674	617	1	2	1	10	1	371	365	2	2	93	410	313	3	1	81	1138	1131	
0	2	110	485	481	1	2	1	10	1	352	340	2	2	95	870	-906	3	1	83	708	657	
0	2	112	850	519	1	2	1	10	1	265	-239	2	2	97	172	-171	3	1	85	509	508	
0	2	114	474	406	1	2	1	11	1	292	-276	2	2	99	491	491	3	1	87	434	414	
0	2	116	781	823	1	2	1	11	1	611	618	2	2	101	281	-299	3	1	89	476	456	
0	2	118	657	649	1	2	1	11	3	346	382	2	2	103	633	-660	3	1	91	391	419	
0	2	120	281	312	1	2	1	11	3	518	510	2	2	105	424	-410	3	1	93	267	-255	
0	2	122	474	406	1	2	1	11	7	529	414	2	2	107	284	296	3	1	95	206	153	
0	2	124	781	823	1	2	1	11	7	553	509	2	2	109	244	-298	3	1	97	357	392	
0	2	126	657	649	1	2	1	11	9	493	493	2	2	111	298	-265	3	1	99	403	420	
0	2	128	281	312	1	2	1	11	11	444	379	2	2	113	425	421	3	1	101	654	669	
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0	2	136	349	250	1	2	1	11	11	617	577	2	2	121	428	411	3	1	109	489	495	
0	2	138	358	383	1	2	1	11	11	472	530	2	2	123	810	-773	3	1	111	838	779	
0	2	140	620	685	1	2	1	11	11	662	679	2	2	125	386	366	3	1	113	332	273	
0																						

Table 3. Continued.

3 10	κ	265	- 231	4 11	4	253	- 306	5 19	1	520	445	7 2	7	339	- 447	
3 10	κ	315	155	4 11	κ	304	290	6 0	κ	3267	3300	7 2	ν	531	645	
3 11	1	632	620	4 11	κ	291	337	6 0	κ	693	709	7 2	1	1041	1058	
3 11	3	541	494	4 12	2	704	815	6 0	κ	1833	1985	7 2	2	343	378	
3 11	κ	591	548	4 12	4	423	446	6 0	κ	711	752	7 2	3	1341	1319	
3 11	7	629	613	4 12	κ	748	727	6 0	κ	426	459	7 2	κ	818	825	
3 11	κ	280	114	4 12	κ	392	248	6 0	1	317	307	7 2	7	767	934	
3 11	κ	585	504	4 12	κ	500	446	6 0	2	459	- 422	7 2	κ	604	439	
3 11	11	421	395	4 12	10	298	- 201	6 0	3	1031	- 991	7 2	10	219	- 204	
3 12	7	315	348	4 12	κ	873	864	6 0	4	344	- 326	7 2	11	379	605	
3 12	κ	345	- 338	4 14	κ	1047	1029	6 0	5	411	- 457	7 2	κ	374	363	
3 12	1	1024	1041	4 14	4	555	517	6 0	κ	2355	2404	7 2	κ	735	702	
3 12	2	284	- 254	4 14	κ	567	576	6 0	2	1	706	674	7 2	κ	283	- 286
3 12	3	935	917	4 14	κ	351	249	6 0	2	1037	1014	7 2	κ	390	387	
3 12	κ	742	700	4 14	κ	426	- 292	6 0	2	264	242	7 2	κ	249	335	
3 12	7	426	423	4 14	κ	574	490	6 0	2	1944	1980	7 2	κ	18	169	
3 12	7	298	15	4 14	κ	355	338	6 0	2	597	639	7 2	κ	568	556	
3 12	7	505	617	4 14	κ	326	- 247	6 0	2	1054	1154	7 2	κ	505	624	
3 12	κ	622	606	4 14	κ	457	- 416	6 0	2	1 ^a	327	370	7 2	κ	494	477
3 12	κ	268	294	5 0	κ	459	- 484	6 0	3	794	799	7 2	κ	440	441	
3 12	κ	444	356	5 0	κ	851	- 874	6 0	3	991	- 927	7 2	κ	413	192	
3 12	κ	732	254	5 0	1	1395	1390	6 0	3	1133	-1082	7 2	κ	279	169	
3 12	κ	327	191	5 0	1	433	417	6 0	3	294	441	7 2	κ	7	406	422
3 12	κ	278	- 301	5 0	1	2284	2251	6 0	3	294	441	7 2	κ	409	332	
3 12	κ	338	- 295	5 0	1	4210	199	6 0	3	422	745	7 2	κ	11	339	303
3 12	κ	375	- 344	5 0	1	1578	1700	6 0	3	456	- 513	7 2	κ	11	221	244
3 12	κ	339	- 109	5 0	1	7111	1204	6 0	3	593	- 480	7 2	κ	11	445	- 432
4 0	κ	1149	-1128	5 0	2	1216	-1132	6 0	3	478	- 420	7 2	κ	11	201	188
4 0	κ	951	996	5 0	2	697	- 693	6 0	3	1166	1142	7 2	κ	11	483	- 414
4 0	κ	1942	1889	5 0	2	1158	-1119	6 0	3	351	- 362	7 2	κ	7	7	307
4 0	κ	471	376	5 0	2	326	- 323	6 0	3	422	745	7 2	κ	7	1761	1180
4 0	κ	488	424	5 0	2	748	- 623	6 0	3	456	- 513	7 2	κ	7	1303	1292
4 0	1	736	654	5 0	2	314	326	6 0	3	661	- 678	7 2	κ	7	971	950
4 0	1	1121	-1095	5 0	2	639	642	6 0	3	777	415	7 2	κ	7	419	- 429
4 0	1	895	- 889	5 0	2	322	- 306	6 0	3	607	751	7 2	κ	7	528	564
4 0	1	187	- 191	5 0	2	1047	1037	6 0	3	751	751	7 2	κ	7	389	470
4 0	1	349	349	5 0	2	200	- 161	6 0	3	332	345	7 2	κ	7	426	- 374
4 0	1	333	- 342	5 0	2	934	975	6 0	3	1713	1778	7 2	κ	7	244	- 223
4 0	1	442	- 513	5 0	2	314	326	6 0	3	857	869	7 2	κ	7	310	262
4 0	1	253	- 203	5 0	2	899	854	6 0	3	1912	1881	7 2	κ	7	1080	1077
4 0	2	189	- 86	5 0	2	308	309	6 0	3	673	683	7 2	κ	7	915	915
4 0	2	250	245	5 0	2	273	219	6 0	3	735	- 385	7 2	κ	7	689	619
4 0	2	313	241	5 0	2	185	- 173	6 0	3	680	734	7 2	κ	7	323	- 380
4 0	2	360	399	5 0	2	215	206	6 0	3	383	141	7 2	κ	7	518	442
4 0	2	184	- 138	5 0	2	501	507	6 0	3	641	- 624	7 2	κ	7	425	544
4 0	2	2018	2112	5 0	2	1045	1081	6 0	3	974	- 940	7 2	κ	7	742	298
4 0	2	702	809	5 0	2	877	987	6 0	3	334	- 350	7 2	κ	10	300	- 335
4 0	2	464	736	5 0	2	411	474	6 0	3	220	- 216	7 2	κ	11	496	512
4 0	2	1088	1038	5 0	2	263	177	6 0	3	1645	1645	7 2	κ	11	236	321
4 0	2	462	- 450	5 0	2	448	- 443	6 0	3	297	298	7 2	κ	11	310	293
4 0	2	604	571	5 0	2	742	- 268	6 0	3	962	978	7 2	κ	11	425	566
4 0	2	777	- 371	5 0	2	443	- 419	6 0	3	1559	1543	7 2	κ	11	263	- 301
4 0	2	407	- 519	5 0	2	224	- 172	6 0	3	230	- 232	7 2	κ	11	472	482
4 0	2	788	- 747	5 0	2	226	- 126	6 0	3	781	799	7 2	κ	11	597	479
4 0	2	351	- 339	5 0	2	355	342	6 0	3	242	- 296	7 2	κ	11	442	408
4 0	2	1746	1713	5 0	2	300	178	6 0	3	813	883	7 2	κ	11	518	442
4 0	2	926	918	5 0	2	205	- 268	6 0	3	280	- 285	7 2	κ	11	425	544
4 0	2	741	319	5 0	2	1302	1347	6 0	3	400	- 445	7 2	κ	12	308	297
4 0	2	1645	1751	5 0	2	396	- 368	6 0	3	449	- 478	7 2	κ	12	344	373
4 0	2	898	910	5 0	2	1164	1177	6 0	3	400	- 445	7 2	κ	12	537	565
4 0	2	785	903	5 0	2	358	- 347	6 0	3	449	- 478	7 2	κ	12	893	874
4 0	2	618	618	5 0	2	1288	1230	6 0	3	373	341	7 2	κ	12	609	609
4 0	2	693	- 667	5 0	2	843	491	6 0	3	860	860	7 2	κ	12	590	509
4 0	2	458	511	5 0	2	468	602	6 0	3	349	355	7 2	κ	12	361	- 398
4 0	2	712	683	5 0	2	337	- 341	6 0	3	324	- 300	7 2	κ	12	351	263
4 0	2	483	527	5 0	2	590	- 619	6 0	3	740	746	7 2	κ	12	253	13
4 0	2	244	189	5 0	2	168	- 110	6 0	3	505	461	7 2	κ	12	621	555
4 0	2	810	817	5 0	2	393	- 370	6 0	3	368	- 329	7 2	κ	12	661	617
4 0	2	363	- 373	5 0	2	294	70	6 0	3	289	244	7 2	κ	12	385	463
4 0	2	1716	1766	5 0	2	595	635	6 0	3	1048	1063	7 2	κ	12	332	221
4 0	2	781	777	5 0	2	174	165	6 0	3	304	335	7 2	κ	12	2047	2145
4 0	2	1722	1722	5 0	2	578	545	6 0	3	778	754	7 2	κ	12	1467	1570
4 0	2	256	246	5 0	2	716	701	6 0	3	632	632	7 2	κ	12	1520	1485
4 0	2	246	138	5 0	2	720	652	6 0	3	302	221	7 2	κ	12	547	540
4 0	2	399	340	5 0	2	459	533	6 0	3	863	853	7 2	κ	12	375	442
4 0	2	376	389	5 0	2	384	233	6 0	3	740	746	7 2	κ	12	502	- 608
4 0	2	537	- 513	5 0	2	181	- 199	6 0	3	529	533	7 2	κ	12	360	- 330
4 0	2	391	- 373	5 0	2	346	- 350	6 0	3	791	740	7 2	κ	12	421	447
4 0	2	301	342	5 0	2	406	- 433	6 0	3	312	169	7 2	κ	12	297	- 310
4 0	2	211	- 123	5 0	2	624	620	6 0	3	392	- 370	7 2	κ	12	977	- 945
4 0	2	353	- 232	5 0	2	400	406	6 0	3	291	- 133	7 2	κ	12	355	- 364
4 0	2	433	461	5 0	2	376	363	6 0	3	658	687	7 2	κ	12	309	- 322
4 0	2	693	733	5 0	2	803	796	6 0	3	322	296	7 2	κ	12	2840	2075
4 0	2	370	- 339	5 0	2	581	583	6 0	3	435	444	7 2	κ	12	422	414
4 0	2	1772	1821	5 0	2	665	653	6 0	3	373	- 230	7 2	κ	12	1278	1289
4 0	2	871	854	5 0	2	341	372	6 0	3	463	466	7 2	κ	12	324	- 264
4 0	2	1059	1021	5 0	2	284	291	6 0	3	1440	-1412	7 2	κ	12	1221	1149
4 0	2	406	336	5 0	2	440	449	6 0	3	356	- 360	7 2	κ	12	229	- 208
4 0	2	430	442	5 0	2	386	366	6 0	3	605	- 615	7 2	κ	12	327	340
4 0	2	354	370	5 0	2	1122	1135	6 0	3	548	- 574	7 2	κ	12	298	484
4 0	2	658	- 668	5 0	2	739	756	6 0	3	212	- 217	7 2	κ	12	329	502
4 0	2	260	- 278	5 0	2	645	651	6 0	3	1801	1841	7 2	κ	12	492	

PALLADIUM (II) COMPLEXES

Table 3. Continued.

R	4	A	219	273	0	7	7	293	410	11	1	7	1076	1032	12	7	1	358	375
R	4	7	372	-340	0	7	0	276	143	11	1	7	798	745	12	7	1	272	-277
R	4	9	574	649	0	9	7	468	-384	11	1	7	220	-30	12	7	7	357	-587
R	4	0	318	360	0	0	1	608	634	11	1	7	419	465	12	9	0	626	607
R	4	10	308	486	0	0	2	216	-276	11	1	0	767	508	12	9	0	366	-339
R	5	1	500	-592	0	0	3	772	764	11	2	2	465	459	12	9	2	500	549
R	5	2	1008	1005	0	0	4	336	-251	11	2	3	366	-331	12	9	4	706	746
R	5	3	354	361	0	0	5	788	307	11	2	4	395	396	12	9	6	638	651
R	5	4	547	564	0	0	7	630	494	11	2	5	374	-319	12	9	7	396	377
R	5	5	507	-585	0	10	2	780	291	11	2	7	764	-304	12	9	9	522	501
R	5	6	570	648	0	10	0	722	-244	11	2	8	328	-433	12	9	1	422	411
R	5	7	2059	2126	0	11	1	698	570	11	2	10	316	-244	12	9	3	399	-331
R	5	8	799	704	0	11	2	240	-167	11	2	11	193	125	12	9	5	280	-249
R	5	9	1340	1404	0	11	3	585	595	11	2	12	437	577	12	9	7	416	-433
R	5	10	1394	1356	0	11	4	720	278	11	2	13	613	571	12	10	0	499	488
R	5	11	250	215	0	11	7	501	533	11	2	14	588	542	12	10	1	416	-431
R	5	12	496	379	0	11	8	387	357	11	2	15	250	277	12	10	2	391	308
R	5	13	745	-439	0	11	9	402	388	11	2	16	449	439	12	10	3	533	470
R	5	14	367	434	0	12	3	523	585	11	2	17	412	617	12	10	4	421	372
R	5	15	292	304	0	12	4	404	427	11	2	18	412	617	12	10	5	568	608
R	5	16	1085	-1082	0	12	5	470	466	11	2	19	377	378	12	10	6	377	355
R	5	17	232	270	0	12	6	637	670	11	2	20	215	120	12	11	5	291	280
R	5	18	749	-723	0	12	7	757	741	11	2	21	272	279	12	11	6	496	483
R	5	19	424	457	0	12	8	421	427	11	2	22	249	247	12	11	7	324	376
R	5	20	292	-322	0	12	9	388	459	11	2	23	245	-219	12	11	8	525	535
R	5	21	1975	2055	0	12	10	537	489	11	2	24	1089	1124	12	11	9	374	336
R	5	22	1292	1266	0	12	11	473	461	11	2	25	940	918	12	11	10	333	389
R	5	23	1012	956	0	12	12	502	359	11	2	26	836	801	12	11	11	711	727
R	5	24	347	393	0	12	13	1514	1574	11	2	27	651	601	12	11	12	357	-249
R	5	25	780	-285	0	12	14	2342	2354	11	2	28	434	566	12	11	13	403	589
R	5	26	339	407	0	12	15	563	531	11	2	29	286	217	12	11	14	536	551
R	5	27	424	-581	0	12	16	656	673	11	2	30	316	330	12	11	15	362	297
R	5	28	326	-336	0	12	17	340	-364	11	2	31	366	417	12	11	16	463	428
R	5	29	473	584	0	12	18	485	461	11	2	32	293	288	12	11	17	443	311
R	5	30	273	292	0	12	19	739	-393	11	2	33	560	-561	12	11	18	532	-292
R	5	31	315	289	0	12	20	424	-427	11	2	34	1113	1338	12	11	19	712	736
R	5	32	518	518	0	12	21	681	705	11	2	35	1203	1206	12	11	20	248	-301
R	5	33	436	449	0	12	22	503	-506	11	2	36	855	844	12	11	21	1009	931
R	5	34	360	373	0	12	23	2617	1976	11	2	37	242	266	12	11	22	349	-309
R	5	35	470	-455	0	12	24	208	-211	11	2	38	496	459	12	11	23	1101	1124
R	5	36	292	322	0	12	25	171	348	11	2	39	256	-238	12	11	24	325	-303
R	5	37	315	314	0	12	26	976	1023	11	2	40	279	-325	12	11	25	959	950
R	5	38	255	305	0	12	27	502	668	11	2	41	247	-144	12	11	26	433	512
R	5	39	359	-447	0	12	28	463	497	11	2	42	557	-577	12	11	27	256	249
R	5	40	993	977	0	12	29	355	317	11	2	43	238	266	12	11	28	236	-270
R	5	41	578	581	0	12	30	416	422	11	2	44	444	429	12	11	29	399	-261
R	5	42	264	212	0	12	31	242	-236	11	2	45	412	329	12	11	30	310	-331
R	5	43	734	715	0	12	32	202	258	11	2	46	372	380	12	11	31	505	451
R	5	44	536	464	0	12	33	496	540	11	2	47	567	445	12	11	32	822	782
R	5	45	302	-144	0	12	34	354	-359	11	2	48	405	349	12	11	33	295	317
R	5	46	423	424	0	12	35	1624	1587	11	2	49	312	-277	12	11	34	776	759
R	5	47	397	371	0	12	36	299	322	11	2	50	485	458	12	11	35	542	615
R	5	48	328	-339	0	12	37	613	625	11	2	51	528	498	12	11	36	342	345
R	5	49	268	65	0	12	38	308	332	11	2	52	419	459	12	11	37	316	252
R	5	50	307	322	0	12	39	509	793	11	2	53	362	349	12	11	38	269	234
R	5	51	275	-192	0	12	40	196	19	11	2	54	600	571	12	11	39	297	145
R	5	52	755	721	0	12	41	459	412	11	2	55	394	361	12	11	40	426	441
R	5	53	620	544	0	12	42	381	376	11	2	56	298	318	12	11	41	209	168
R	5	54	560	545	0	12	43	371	391	11	2	57	307	-305	12	11	42	676	654
R	5	55	397	268	0	12	44	485	433	11	2	58	654	604	12	11	43	326	265
R	5	56	353	217	0	12	45	335	-242	11	2	59	602	642	12	11	44	879	914
R	5	57	319	-222	0	12	46	1318	1353	11	2	60	533	467	12	11	45	724	684
R	5	58	395	313	0	12	47	1628	1631	11	2	61	454	408	12	11	46	415	484
R	5	59	452	419	0	12	48	245	171	11	2	62	428	-395	12	11	47	406	376
R	5	60	424	359	0	12	49	801	894	11	2	63	478	499	12	11	48	271	-299
R	5	61	232	-77	0	12	50	589	505	11	2	64	395	395	12	11	49	876	907
R	5	62	868	-825	0	12	51	288	-226	11	2	65	172	173	12	11	50	915	926
R	5	63	382	-419	0	12	52	613	591	12	0	2	1585	1641	12	7	5	674	836
R	5	64	272	-305	0	12	53	406	-355	12	0	3	1150	1158	12	7	7	500	476
R	5	65	2273	2234	0	12	54	1332	1347	12	0	4	716	728	12	7	9	232	-280
R	5	66	243	-225	0	12	55	323	347	12	0	5	434	347	12	7	11	244	250
R	5	67	882	906	0	12	56	625	541	12	0	6	84	826	12	7	13	495	-465
R	5	68	638	584	0	12	57	313	285	12	0	7	198	-167	12	7	15	376	-432
R	5	69	202	220	0	12	58	422	568	12	0	8	259	-247	12	7	17	408	-363
R	5	70	516	614	0	12	59	440	381	12	0	9	687	650	12	7	19	487	481
R	5	71	407	320	0	12	60	224	-64	12	0	10	325	-382	12	7	21	650	621
R	5	72	304	-269	0	12	61	285	-628	12	0	11	590	-610	12	7	23	663	657
R	5	73	249	-247	0	12	62	326	337	12	0	12	561	600	12	7	25	531	513
R	5	74	341	-449	0	12	63	281	-244	12	0	13	415	392	12	7	27	272	-297
R	5	75	290	-247	0	12	64	707	694	12	0	14	1041	1082	12	7	29	290	-231
R	5	76	468	-473	0	12	65	438	615	12	0	15	868	825	12	7	31	276	312
R	5	77	1295	1239	0	12	66	545	600	12	0	16	432	654	12	7	33	409	362
R	5	78	711	693	0	12	67	763	344	12	0	17	341	316	12	7	35	531	461
R																			

Table 3. Continued.

14	3	A	299	-	81	15	E	7	541	539	16	12	A	394	308	19	1	1	P20	845
14	4	A	333	353	15	E	9	379	361	16	14	A	509	486	19	1	3	560	612	
14	4	1	530	524	15	E	11	378	257	16	14	2	483	521	19	1	5	373	305	
14	4	A	850	838	15	A	1	762	-362	16	14	4	346	379	19	1	9	354	301	
14	4	A	671	575	15	A	2	277	-236	17	0	4	315	-287	19	1	1	431	546	
14	4	A	771	928	15	A	4	302	-266	17	1	1	1111	1198	19	2	9	389	328	
14	E	1	262	-315	15	A	7	330	297	17	1	3	764	777	19	4	3	384	245	
14	E	A	308	255	15	7	1	460	558	17	1	E	364	312	19	4	3	296	187	
14	E	A	385	262	15	7	E	645	619	17	2	4	312	-268	19	4	E	426	301	
14	A	A	913	955	15	7	7	402	467	17	2	E	288	-320	19	E	3	422	497	
14	A	2	568	581	15	8	1	305	-304	17	3	1	293	254	19	E	E	336	251	
14	A	A	1170	1125	15	9	1	439	450	17	3	1	759	747	19	7	1	764	778	
14	A	E	266	-144	15	9	3	708	648	17	3	3	445	428	19	7	1	537	588	
14	A	A	602	602	15	9	E	644	638	17	3	E	293	170	19	7	1	402	427	
14	A	E	518	619	15	9	7	594	570	17	3	9	413	378	19	7	1	499	547	
14	7	1	349	-321	15	10	A	351	169	17	E	0	759	750	19	9	3	350	348	
14	7	E	353	-395	15	11	0	423	397	17	E	1	736	760	19	9	E	406	262	
14	P	0	1195	1204	15	11	1	477	-307	17	E	5	334	665	20	0	0	1053	1117	
14	P	2	430	399	15	11	7	372	326	17	E	5	514	472	20	0	2	487	427	
14	P	A	929	914	15	11	0	382	335	17	E	7	342	336	20	0	A	765	736	
14	P	A	499	348	15	11	1	499	536	17	A	4	297	-247	20	0	A	416	470	
14	P	1	284	-244	15	11	3	460	458	17	A	A	328	-206	20	2	2	738	770	
14	Q	2	278	-290	15	11	4	422	292	17	A	2	274	-277	20	2	A	709	683	
14	Q	E	382	-645	15	15	3	384	398	17	7	3	571	549	20	2	A	415	426	
14	Q	E	303	-239	16	0	2	1406	1407	17	7	E	341	334	20	A	4	397	444	
14	Q	A	385	-232	16	0	A	614	596	17	7	7	389	269	20	A	0	377	421	
14	10	0	539	610	16	0	A	855	405	17	7	7	322	-340	20	A	0	700	752	
14	10	A	756	743	16	1	1	814	-821	17	7	0	434	-296	20	A	2	345	344	
14	10	0	560	536	16	1	3	292	-258	17	9	1	484	449	20	A	2	587	588	
14	11	2	274	-286	16	1	E	653	-650	17	9	3	456	436	20	A	0	728	729	
14	11	3	277	240	16	1	7	310	165	17	11	3	311	228	20	A	1	413	-364	
14	12	A	688	700	16	2	0	406	383	17	11	7	362	315	20	A	2	445	351	
14	12	3	280	81	16	2	1	305	-347	17	13	1	641	588	20	A	4	425	533	
14	12	A	548	556	16	2	2	958	929	17	13	3	487	487	20	A	10	411	332	
14	12	A	418	379	16	2	3	325	-266	18	0	0	970	966	21	1	1	538	479	
14	14	1	806	759	16	2	A	328	239	18	0	2	804	869	21	1	3	434	444	
14	14	2	378	332	16	2	E	335	-267	18	0	A	515	404	21	1	3	503	487	
14	14	A	616	626	16	2	E	627	639	18	0	A	884	956	21	1	7	490	369	
15	0	2	653	-630	16	3	1	378	-386	18	2	2	815	787	21	1	7	35	-73	
15	0	A	390	-409	16	3	2	302	161	18	2	E	361	301	21	3	3	391	491	
15	0	A	304	-142	16	3	A	345	-109	18	4	0	696	718	21	3	7	409	358	
15	1	0	265	-263	16	3	E	394	-392	18	4	2	641	661	21	E	1	320	294	
15	1	1	926	924	16	4	2	899	891	18	4	A	403	339	21	7	1	349	452	
15	1	2	260	-165	16	4	A	334	255	18	4	A	291	-51	21	7	3	616	526	
15	1	3	837	833	16	4	A	685	703	18	4	A	304	287	21	7	E	472	466	
15	1	A	351	-290	16	4	10	340	513	18	E	2	300	249	21	7	0	378	-310	
15	1	E	878	836	16	E	1	468	-518	18	E	4	459	401	21	9	1	357	264	
15	1	7	560	534	16	E	2	340	292	18	E	A	362	210	22	0	0	415	354	
15	1	9	458	399	16	E	3	708	-392	18	E	A	783	785	22	0	2	546	569	
15	2	1	209	-313	16	E	4	466	697	18	A	2	872	854	22	A	A	554	553	
15	2	2	367	-347	16	A	1	318	324	18	A	3	299	215	22	A	2	481	531	
15	2	3	242	-229	16	A	2	987	980	18	A	4	319	266	22	A	A	481	531	
15	2	4	546	-560	16	A	4	485	417	18	7	3	314	-329	22	2	2	362	262	
15	3	1	398	334	16	A	A	618	572	18	7	A	301	194	22	2	4	534	450	
15	3	2	387	-278	16	7	1	285	-282	18	8	0	594	652	22	2	A	469	540	
15	3	3	348	314	16	7	E	362	-340	18	8	1	312	-174	22	2	A	402	241	
15	3	E	730	741	16	8	0	706	714	18	8	2	735	752	22	2	A	383	396	
15	3	7	640	643	16	8	2	844	861	18	8	A	390	323	22	A	A	415	420	
15	3	9	518	501	16	8	A	387	276	18	8	A	456	296	22	A	0	357	217	
15	4	1	276	-292	16	8	E	312	-288	18	9	0	381	-94	22	A	2	438	291	
15	4	2	227	-237	16	8	A	431	418	18	10	2	429	423	22	1	3	439	422	
15	4	7	289	217	16	10	2	315	340	18	10	A	357	313	23	1	E	464	412	
15	E	0	562	512	16	10	A	517	454	18	11	2	316	214	23	3	3	394	314	
15	E	1	411	407	16	12	0	391	302	18	12	0	337	219	24	0	2	490	421	
15	E	3	647	673	16	12	1	309	373	19	0	0	396	447						
15	E	E	722	706	16	12	2	372	389	19	0	A	337	-361						

where C_T is the total number of counts and C_N the net count (peak minus background).

After six cycles the R -value converged at 12.8 %, and the corresponding Fourier map contained additional peaks with heights up to 4.8 e.Å⁻³. The two largest were situated very close to the two-fold axis of symmetry and could only be interpreted in terms of statistically distributed atoms, which indeed is required by the space group symmetry with four molecules of crystal ethanol per unit cell. The distance between the two symmetry related peaks closest to the rotation axis (Fig. 1) were about 1.54 Å. This peak was therefore interpreted as corresponding to half a methylene carbon atom of one ethanol molecule and half a methyl carbon of another molecule. In the least squares refinement it was treated as a normal carbon atom. Since the final R -value was reasonable (4.89 Å²), refinement of individual half-atoms at this position

Table 4. Bond distances and angles with estimated standard deviations.

Distance	(Å)	Distance	(Å) ^a
Pd—P ₁	2.334(3)	C ₂ —H ₂	1.07
Pd—P ₂	2.333(3)	C ₃ —H ₃	1.13
P ₁ —C ₇	1.829(11)	C ₄ —H ₄	0.88
P ₂ —C ₈	1.849(12)	C ₅ —H ₅	0.89
P ₂ —C ₉	1.809(13)	C ₆ —H ₆	0.92
P ₁ —C ₁₀	1.800(12)	C ₁₂ —H ₁₂	0.94
P ₁ —C ₁	1.800(12)	C ₁₃ —H ₁₃	0.72
P ₂ —C ₁₁	1.800(10)	C ₁₄ —H ₁₄	0.74
C ₇ —C ₈	1.517(13)	C ₁₅ —H ₁₅	0.91
C ₁ —C ₂	1.348(14)	C ₁₆ —H ₁₆	0.88
C ₂ —C ₃	1.368(15)	C ₇ —H _{7,1}	0.79
C ₃ —C ₄	1.378(15)	C ₇ —H _{7,2}	0.87
C ₄ —C ₅	1.383(15)	C ₈ —H _{8,1}	0.85
C ₅ —C ₆	1.414(16)	C ₈ —H _{8,2}	1.03
C ₆ —C ₁	1.401(14)	C ₉ —H _{9,1}	0.77
C ₁₁ —C ₁₂	1.384(14)	C ₉ —H _{9,2}	0.97
C ₁₂ —C ₁₃	1.409(14)	C ₉ —H _{9,3}	1.05
C ₁₃ —C ₁₄	1.390(15)	C ₁₀ —H _{10,1}	0.83
C ₁₄ —C ₁₅	1.363(16)	C ₁₀ —H _{10,2}	0.93
C ₁₅ —C ₁₆	1.375(14)	C ₁₀ —H _{10,3}	0.95
C ₁₆ —C ₁₁	1.386(13)		
C _c —C _{c'}	1.534(18)		
C _c —O _c	1.335(17)		

Angle	(°)	Angle	(°) ^b
P ₁ —Pd—P ₂	85.0(1)	C ₁ —C ₂ —H ₂	131
Pd—P ₁ —C ₁	110.6(4)	H ₂ —C ₂ —C ₃	108
Pd—P ₁ —C ₇	105.8(4)	C ₂ —C ₃ —H ₃	132
Pd—P ₁ —C ₁₀	122.4(4)	H ₃ —C ₃ —C ₄	105
Pd—P ₂ —C ₁₁	118.0(4)	C ₃ —C ₄ —H ₄	126
Pd—P ₂ —C ₈	106.4(4)	H ₄ —C ₄ —C ₅	115
Pd—P ₂ —C ₉	113.5(4)	C ₄ —C ₅ —H ₅	120
P ₁ —C ₇ —C ₈	107.8(7)	H ₅ —C ₅ —C ₆	120
P ₂ —C ₈ —C ₇	110.1(7)	C ₅ —C ₆ —H ₆	121
P ₁ —C ₁ —C ₂	119.7(1.0)	H ₆ —C ₆ —C ₁	120
P ₁ —C ₁ —C ₆	119.4(9)	C ₁₁ —C ₁₂ —H ₁₂	120
C ₁ —C ₂ —C ₃	119.8(1.1)	H ₁₂ —C ₁₂ —C ₁₃	119
C ₂ —C ₃ —C ₄	122.1(1.2)	C ₁₂ —C ₁₃ —H ₁₃	113
C ₃ —C ₄ —C ₅	118.9(1.2)	H ₁₃ —C ₁₃ —C ₁₄	129
C ₄ —C ₅ —C ₆	119.5(1.2)	C ₁₃ —H ₁₄ —C ₁₄	113
C ₅ —C ₆ —C ₁	118.8(1.2)	H ₁₄ —C ₁₄ —C ₁₅	126
C ₆ —C ₁ —C ₂	120.9(1.1)	C ₁₄ —C ₁₅ —H ₁₅	128
P ₂ —C ₁₁ —C ₁₂	119.3(8)	H ₁₅ —C ₁₅ —C ₁₆	113
P ₂ —C ₁₁ —C ₁₃	122.4(9)	C ₁₅ —C ₁₆ —H ₁₆	119
C ₁₁ —C ₁₂ —C ₁₃	120.5(1.1)	H ₁₆ —C ₁₆ —C ₁₁	119
C ₁₂ —C ₁₃ —C ₁₄	118.3(1.2)	P ₁ —C ₇ —H _{7,1}	112
C ₁₃ —C ₁₄ —C ₁₅	121.9(1.2)	P ₁ —C ₇ —H _{7,2}	94
C ₁₄ —C ₁₅ —C ₁₆	118.6(1.2)	H _{7,1} —C ₇ —H _{7,2}	126
C ₁₅ —C ₁₆ —C ₁₁	122.4(1.2)	P ₂ —C ₈ —H _{8,1}	95
C ₁₆ —C ₁₁ —C ₁₂	118.3(1.0)	P ₂ —C ₈ —H _{8,2}	100
C _c —C _{c'} —O _c	112.9(1.3)		

Angle	(°)	Angle	(°)
H _{8,1} —C ₈ —H _{8,2}	120	P ₂ —C ₉ —H _{9,1}	109
P ₁ —C ₁₀ —H _{10,1}	109	P ₂ —C ₉ —H _{9,2}	104
P ₁ —C ₁₀ —H _{10,2}	109	P ₂ —C ₉ —H _{9,3}	105
P ₁ —C ₁₀ —H _{10,3}	107	H _{9,1} —C ₉ —H _{9,2}	123
H _{10,1} —C ₁₀ —H _{10,2}	106	H _{9,1} —C ₉ —H _{9,3}	101
H _{10,1} —C ₁₀ —H _{10,3}	116	H _{9,2} —C ₉ —H _{9,3}	115
H _{10,2} —C ₁₀ —H _{10,3}	110		

^a e.s.d.'s range from 0.1 to 0.2 Å.^b e.s.d.'s of about 4°.

was not attempted. The other peak was interpreted as a half oxygen atom and refined with multiplicity factor 0.5.

Methylene and phenyl hydrogen positions were calculated by assuming C-H bonds of length 1.03 Å. Three methyl hydrogen atoms could be localized in the difference Fourier map while positional parameters of the remaining three were calculated. Hydrogens of the statistically distributed ethanol molecules were not included in the calculations. Coordinates and isotropic thermal parameters were refined for all H-atoms. The final conventional R -value was 7.9 % and the weighted value $R_w=6.0$ % for 1509 observed reflections.

Final positional and thermal parameters are given in Tables 1 and 2. A comparison between observed and calculated structure factors is presented in Table 3.

Fig. 1 shows a schematical drawing of the complex viewed along [001]. It may be described¹ in terms of a MESO-*anti-trans* form, where MESO indicates a centre of symmetry or a mirror plane, and *anti* means that pairs of equal ligands are situated at opposite sides of the plane through the phosphorus atoms (planarity is required by the center of symmetry).

Bond distances and angles may be found in Table 4. The two Pd-P distances (2.334 Å and 2.333 Å) are equal and somewhat shorter than the sum of covalent radii (2.42 Å). The bond angle P₁-Pd-P₂ is 85.0°. The two Pd-P-C angles as well as the two P-C-C angles of the five-membered ring are equal within probable limits of error, with values (105.8°; 106.4°) and (107.8°; 110.1°), respectively. The distances from the P₁-Pd-P₂ plane to C₇ and C₈ are +0.56 Å and -0.20 Å, respectively. The best least squares plane through four ring atoms is:

Atoms defining the plane	Distance from the plane
Pd	-0.013
P ₁	0.032
P ₂	0.050
C ₈	-0.107

C₇ is at a distance of 0.65 Å from this plane. The geometry of the five-membered ring is thus somewhere between an envelope and the conformation with symmetry C₂.

Table 5. Least squares planes of the phenyl rings.

Atoms defining the plane	Distance (Å)	Atoms defining the plane	Distance (Å)
C ₁	.005	C ₁₁	-.001
C ₂	-.007	C ₁₂	.000
C ₃	.003	C ₁₃	.003
C ₄	.004	C ₁₄	-.004
C ₅	-.007	C ₁₅	.003
C ₆	.002	C ₁₆	.000

The six P—C distances are equal within error limits. Their mean value, 1.815 Å, is somewhat smaller than the microwave result (1.841 ± 0.003 Å) reported for trimethylphosphine.⁵ According to the large standard deviations in the present results, the difference is probably not significant.

As shown in Table 5, the phenyl groups are planar, the largest deviation being 0.007 Å.

The larger distance (4.09 Å) between the chlorine ions and palladium shows that there is no bonds between these atoms. On the other hand, several chlorine-hydrogen contacts are somewhat (but not significantly) shorter than van der Waals distance. Also the distance Cl⁻...O_c of 3.19 Å corresponds to van der Waals contact. The angles P₁—Pd...Cl- and P₂—Pd...Cl- are 70.2° and 79.6°, respectively.

The distance C_e—C_e' (1.534 Å) and the angle C_e—C_e'—O_c (112.9°) of the ethanol molecule are normal. The C_e—O_c distance (1.335 Å) indicates, however, a slight disorder of C_e and C_e' with the methylene carbon situated about 0.08 Å further away from the oxygen and the methyl group slightly displaced in the opposite direction. The separation should be no more than 0.15—0.20 Å (in agreement with the thermal parameter value arrived at) and, as mentioned before, refinement under these assumptions was not attempted.

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