

## Crystal and Molecular Structure of 5-(Bromomethylene)- 10,11-dihydro-5H-dibenzo[a,d]-cycloheptene

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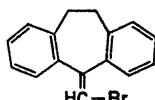
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5-(Bromomethylene)-10,11-dihydro-5H-dibenzo[a,d]-cycloheptene has been studied in order to determine the molecular conformation of the pharmacologically active dibenzocycloheptene derivatives. The crystals are orthorhombic with  $a=6.270$  Å,  $b=9.288$  Å,  $c=22.46$  Å. The unit cell contains four molecules, and the space group is  $P2_12_12_1$ . The seven-membered ring exists in a boat conformation, the long sides of which are formed by the bonds common with the benzene rings. The planes of the two benzene rings form a dihedral angle of 76°. A possible mechanism of action on the neuronal membrane is discussed.

Numerous tricyclic compounds are used as psychoactive agents. Certain dibenzocycloheptene derivatives are utilized in the therapy of depressions, whereas phenothiazines of various kinds are used as tranquillizers. The molecules of these two types of tricyclic compounds, which are relatively similar, can thus exhibit opposite pharmacological effects. This has been a reason for extensive speculations on their molecular geometry.

A crystal structure study was undertaken by the present author in order to provide information on the detailed molecular conformation in these active tricyclic compounds. As will be discussed below it is believed that drugs of this type act on the biological membranes, and knowledge on the structure of the drug as well as of the membrane is needed in order to understand their function on a molecular level.

The structural formula of 5-(bromomethylene)-10,11-dihydro-5H-dibenzo [a,d]cycloheptene is:



If the bromine atom is replaced by a  $-\text{CH}_2-\text{CH}_2-\text{N}(\text{CH}_3)_2$  group a wellknown antidepressive agent called amitriptylin is obtained.

## PREPARATION OF CRYSTALS

5-(Bromomethylene)-10,11-dihydro-5H-dibenzo[a,d]-cycloheptene was prepared according to a method described by Ebnöther *et al.*,<sup>1</sup> and crystals for X-ray work were grown from ethanol at room temperature. The crystals form needles parallel to the *a*-axis with a cross-section of hexagonal appearance (edges parallel to the *b*-axis and two *bc*-diagonals).

## X-RAY DATA

Rotation and Weissenberg photographs were taken about the *a*- and *b*-axis with a calibrated camera using  $\text{CuK}\alpha$  radiation. The following X-ray data were obtained:

Unit cell: orthorhombic

$a = 6.270 \pm 0.009 \text{ \AA}$ ,  $b = 9.288 \pm 0.012 \text{ \AA}$ ,  $c = 22.46 \pm 0.03 \text{ \AA}$ .

Space group:  $P2_12_12_1$ .

Four molecules per unit cell.

Density calculated:  $1.45 \text{ g.cm}^{-3}$ .

Density measured:  $1.43 \text{ g.cm}^{-3}$ .

The reflection intensities were measured using an on-line automatic scanner developed by Abrahamsson.<sup>2</sup> The corresponding operation programs produce a paper tape with indices and integrated intensity values of the spots on the films which can be used directly as input to the data reduction program. The film factors were plotted against the intensity values for each film pack, and from this plot the most reliable intensity interval was estimated. The final intensity values were then selected from these intervals.

The reflection intensities were corrected for the Lorentz and polarization factors but not for absorption. Absolute values were later obtained by comparison with calculated structure factors.

## STRUCTURE DETERMINATION

The position of the bromine atom in the asymmetric unit was derived from the Patterson function, which was sharpened to correspond to point atoms at rest. The carbon atoms were then located by successive cycles of structure factors calculations and Fourier syntheses.

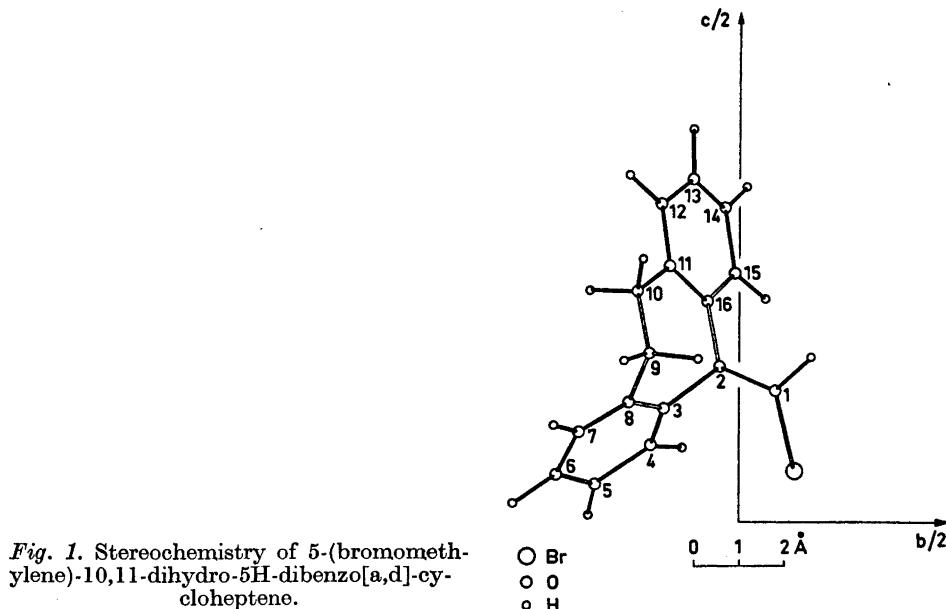
The structure was refined by block-diagonal and full matrix least-squares treatment using anisotropic temperature factors. When an *R* value of 0.20 was reached a difference synthesis was calculated. All hydrogen atoms could then be located. The hydrogen atoms were included in the following least-squares refinement with calculated positions, as the corresponding peaks in the Fourier maps were too broad to give positions of reasonable accuracy. The hydrogen positions were not refined but recalculated after each round of least-squares refinement. Isotropic temperature factors were used for the hydrogen atoms calculated from the mean-square amplitude tensor of the corresponding hydrogen-carrying atom. When all shifts were less than one-third of the standard deviations the refinement was stopped. The final *R* value for the 611 observed reflections was 0.12.

The scattering curves given in the *International Tables for X-ray Crystallography*, Vol. III (1962), were used. The calculations were performed on the Dataaab D21 computer using a program system developed at this Institute.<sup>3</sup> The weighting scheme applied in the refinement was:

$$w = \frac{1}{1 + [(|F_o| - 8|F_{\min}|)/5|F_{\min}|]^2}$$

### RESULTS AND DISCUSSION

Observed and calculated structure factors are listed in Table 1. Final atomic positions are given in Table 2. The vibration tensor elements for non-hydrogen atoms are given in Table 3. Isotropic temperature factors were used for hydrogen atoms averaged from the mean-square amplitude tensors of the corresponding hydrogen-carrying atom.



The stereochemistry of the molecule is shown in Fig. 1. The seven-membered ring exists in a boat form although irregular as will be described below, and the long sides are formed by the bonds which are common with the benzene rings. Bond distances and angles are given in Tables 4 and 5. The  $sp^3$ -carbon atoms in the seven-membered ring have an average bond angle of  $114.5^\circ$ . Enlarged angles in seven-membered rings have been discussed by Asher and Sim.<sup>4</sup> The planes of the two benzene rings and adjacent bonds form a dihedral angle of  $76^\circ$ . The geometry of the seven-membered ring is

Table 1. Observed and calculating structure factors ( $\times 100$ ) and phase angles as fraction of one revolution.

<i>h</i>	<i>k</i>	<i>l</i>	Fobs	Fcalc	Fi	<i>h</i>	<i>k</i>	<i>l</i>	Fobs	Fcalc	Fi	<i>h</i>	<i>k</i>	<i>l</i>	Fobs	Fcalc	Fi	
0	0	4	1834	1883	1.0000	0	6	6	2240	2110	0.5000	1	4	2	6254	5509	0.0332	
0	6	5680	5916	0.5000		0	6	8	5023	4791	1.0000	1	4	3	2149	1783	0.6574	
0	8	7411	7569	0.5000		0	6	10	2362	2153	0.5000	1	4	4	7308	8293	0.9823	
0	10	10073	1058	0.5000		0	6	12	2265	2266	0.5000	1	4	5	4466	3509	0.0626	
0	12	5374	5310	0.5000		0	6	14	2350	2365	0.5000	1	4	6	6882	7841	0.9936	
0	14	3078	3058	0.5000		0	6	16	2342	3479	0.5000	1	4	7	3299	3127	0.6257	
0	16	2798	1964	1.0000		0	6	18	2342	2342	0.5000	1	4	8	3870	4194	0.9975	
0	18	4370	3899	1.0000		0	6	20	2161	2489	0.5000	1	4	9	3770	3629	0.5020	
0	20	4262	3516	1.0000		0	7	1	2381	2652	0.7500	1	4	10	1955	1197	0.4554	
1	1	4229	4871	0.2500		0	7	2	1987	1901	0.2500	1	4	11	1203	4395	0.5064	
1	2	5118	5118	0.5000		0	7	3	14	1633	0.7500	1	4	12	2614	2513	0.2486	
1	3	5348	5295	0.2500		0	7	4	6198	6000	0.2500	1	4	13	2625	2561	0.5321	
1	4	4013	3659	0.2500		0	7	6	4337	5096	0.2500	1	4	14	10	150	0.0583	
1	5	2076	2379	0.2500		0	7	7	1720	158	0.7500	1	5	15	2103	3669	0.7500	
1	6	2041	2120	0.7500		0	7	8	3183	2870	0.2500	1	5	16	2417	2058	0.3138	
1	7	2635	2528	0.2500		0	7	9	2021	2125	0.2500	1	5	17	2680	2385	0.6827	
1	8	5094	0.7500			0	7	10	2489	2335	0.2500	1	5	18	6439	6118	0.2522	
1	9	2701	2041	0.2500		0	7	11	2489	2335	0.2500	1	5	19	2119	1404	0.8762	
1	10	3551	4230	0.7500		0	7	12	2470	2300	0.7500	1	5	20	600	600	0.0066	
1	11	6556	6370	0.7500		0	7	13	4028	4179	0.7500	1	5	21	602	5815	0.2774	
1	12	3978	4339	0.7500		0	7	14	3644	3696	0.7500	1	5	22	1787	1483	0.2452	
1	13	3660	3424	0.7500		0	7	15	3469	3613	1.0000	1	5	23	1937	1521	0.1936	
1	14	2349	2181	0.2500		0	7	16	2259	2370	0.5000	1	5	24	3983	3636	0.2929	
1	15	3730	3730	0.5000		0	7	17	1954	1794	1.0000	1	5	25	2609	2558	0.7830	
1	16	3591	3630	0.2500		0	7	18	3514	3593	0.7500	1	5	26	2725	2575	0.2725	
1	17	1409	1496	0.7500		0	7	19	1087	1821	0.5000	1	5	27	2614	2593	0.5011	
1	18	2981	2440	0.2500		0	7	20	2878	3312	0.5000	1	5	28	2802	2454	0.7526	
1	19	2051	1324	0.2500		0	7	21	3466	3956	0.2500	1	5	29	3705	3480	0.7878	
1	20	4058	4577	0.2500		0	7	22	2010	2271	0.5000	1	5	30	3550	3317	0.0498	
1	21	4730	3929	0.5000		0	7	23	2010	2271	0.5000	1	5	31	3771	3554	0.5436	
2	2	3369	2194	0.5000		0	7	24	1256	1340	0.5000	1	5	32	2968	2755	0.2552	
2	3	15104	15634	0.5000		0	7	25	1569	1043	0.7499	1	5	33	4749	4598	0.4006	
2	4	3007	2199	1.0000		0	7	26	15971	16631	0.5000	1	5	34	5195	4850	0.4674	
2	5	13825	13852	0.5000		0	7	27	3023	3319	0.2500	1	5	35	1979	2007	0.9621	
2	6	2621	2197	0.5000		0	7	28	8431	8849	0.5000	1	5	36	3161	2954	0.9854	
2	7	8087	8087	0.5000		0	7	29	3622	3622	0.5000	1	5	37	1707	1899	0.9950	
2	8	1118	1383	1.0000		0	7	30	10	1899	4497	0.5000	1	5	38	5233	5096	0.2550
2	9	3451	3694	0.5000		0	7	31	12	6189	5757	1.0000	1	5	39	2124	1911	0.2577
2	10	1519	990	1.0000		0	7	32	5810	6709	1.0000	1	5	40	5478	4842	0.2070	
2	11	7373	7129	1.0000		0	7	33	3788	4765	1.0000	1	5	41	2968	2407	0.1909	
2	12	5578	5589	1.0000		0	7	34	12122	13141	0.2500	1	5	42	2417	1972	0.7480	
2	13	4154	3192	0.5000		0	7	35	2511	2148	0.1917	1	5	43	4733	4282	0.7549	
2	14	4188	3672	1.0000		0	7	36	1477	1777	0.2500	1	5	44	1035	1952	0.2759	
2	15	2136	1704	0.5000		0	7	37	1603	682	0.237	1	5	45	5295	5052	0.2563	
2	16	2205	1934	1.0000		0	7	38	3023	3093	0.9499	1	5	46	2665	2400	0.7646	
2	17	1310	1112	1.0000		0	7	39	5604	6579	0.6760	1	5	47	2110	2095	0.7945	
2	18	3282	3812	0.5000		0	7	40	2842	1883	0.7764	1	5	48	2124	1859	0.7067	
2	19	7153	7153	0.5000		0	7	41	6129	7161	0.6797	1	5	49	1840	2084	0.3004	
2	20	6077	5796	0.7500		0	7	42	2674	2674	0.5000	1	5	50	2119	2681	0.2681	
2	21	1216	213	0.7467		0	7	43	3190	2607	0.7905	1	5	51	1911	1916	0.1616	
2	22	9731	9616	0.7500		0	7	44	1811	5749	0.7847	1	5	52	2517	2562	0.5353	
2	23	1524	1531	0.7500		0	7	45	3984	3617	0.2963	1	5	53	4057	3874	0.5053	
2	24	6505	6572	0.7500		0	7	46	3940	3666	0.7644	1	5	54	2830	3074	0.0396	
2	25	2873	3451	0.7500		0	7	47	5505	5598	0.1987	1	5	55	1957	2346	0.5165	
2	26	1644	972	0.2500		0	7	48	2420	1869	0.1287	1	5	56	2377	2343	0.0073	
2	27	4348	4057	0.7500		0	7	49	3165	3819	0.2388	1	5	57	2539	2209	0.9778	
2	28	3664	3507	0.2500		0	7	50	3149	2858	0.2255	1	5	58	2002	2277	0.2500	
2	29	2462	2587	0.7500		0	7	51	3064	2646	0.2714	1	5	59	2238	1622	0.8026	
2	30	5573	5573	0.2500		0	7	52	2826	2693	0.2714	1	5	60	2460	2141	0.3838	
2	31	3330	3490	0.2500		0	7	53	2567	2567	0.2500	1	5	61	2903	3493	0.7827	
2	32	1992	1408	0.2500		0	7	54	15171	15180	0.4109	1	5	62	4607	4798	0.7374	
2	33	1473	1180	0.2500		0	7	55	1778	1925	0.7674	1	5	63	3068	3137	0.7322	
2	34	9138	8731	0.5000		0	7	56	6001	6749	0.0890	1	5	64	2018	2522	0.2681	
2	35	10908	10032	0.5000		0	7	57	2525	1544	0.2445	1	5	65	1931	2624	0.4734	
2	36	2569	2300	0.5000		0	7	58	5257	1044	0.2767	1	5	66	1598	2324	0.0315	
2	37	1419	1308	1.0000		0	7	59	4623	4807	0.9416	1	5	67	2227	1608	0.2500	
2	38	2965	2390	1.0000		0	7	60	2800	1904	0.4150	1	5	68	3504	3435	0.2500	
2	39	2572	3636	0.2500		0	7	61	10583	2057	0.2872	1	5	69	1158	1696	0.5000	
2	40	7245	6888	0.5000		0	7	62	6161	6266	0.2266	1	5	70	2112	2634	0.7500	
2	41	997	868	0.5000		0	7	63	2563	1664	0.7650	1	5	71	5725	5699	1.0000	
2	42	4264	4262	0.7500		0	7	64	5449	5055	0.0280	1	5	72	2456	2492	0.7500	
2	43	4774	4296	0.2500		0	7	65	3688	4147	0.2445	1	5	73	3226	3646	0.7383	
2	44	2231	2209	0.7500		0	7	66	3176	2593	0.0281	1	5	74	2066	1335	0.4982	
2	45	2881	2521	0.7500		0	7	67	5393	5912	0.2708	1	5	75	4174	3176	0.8194	
2	46	4004	3410	0.5000		0	7	68	1411	994	0.7813	1	5	76	6551	6053	0.0297	
2	47	8254	7920	0.7500		0	7	69	5423	5789	0.1592	1	5	77	2446	2263	0.6018	
2	48	1114	782	0.7469		0	7	70	1957	1957	0.2267	1	5	78	4931	4242	0.2422	
2	49	1790	2129	0.7500		0	7	71	3407	3060	0.2243	1	5	79	5264	4865	0.0052	
2	50	2406	1998	1.0000		0	7	72	2447	2461	0.2445	1	5	80	3182	4165	0.1779	
2	51	4647	4647	1.0000		0	7	73	5322	4940	0.2937	1	5	81	3224	3128	0.0050	
2	52	1639	1691	0.5000		0	7	74	5536									

Table 1. Continued.

<i>n</i>	<i>k</i>	<i>l</i>	Fobs	Fcalc	<i>F</i> <sub>1</sub>	<i>n</i>	<i>k</i>	<i>l</i>	Fobs	Fcalc	<i>F</i> <sub>1</sub>	<i>n</i>	<i>k</i>	<i>l</i>	Fobs	Fcalc	<i>F</i> <sub>1</sub>			
2	2	2108	1356	0.4667		3	1	12	3673	3531	0.8226	4	0	12	4198	3854	0.5000			
		6048	7257	0.9915				13	3624	3604	0.8169			2404	2084	1.0000				
		2460	2150	0.8654			1	15	2757	2438	0.7278			1	1	3415	2458	0.2920		
		6556	7563	0.0028			2	1	2597	2992	0.7500			2	1	2977	2454	0.1877		
		2132	2365	0.6761			2	2	4307	5225	0.0754			1	1	4258	3711	0.1445		
		7011	7197	0.0098			2	3	2467	1921	0.8633			1	1	3451	3172	0.9207		
		4549	4453	0.2959			3	4	501	4761	0.9775			1	1	3884	2943	0.3641		
		10	1784	871	0.8101			4	5	1897	1747	0.5523			1	1	4200	3750	0.7518	
		11	3302	2662	0.5111			5	6	2677	2344	0.2091			1	1	2675	2192	0.1021	
		12	5702	5144	0.5310			7	7	4310	3712	0.5431			1	1	2198	1962	0.7971	
		13	2418	2314	0.4054			9	8	4812	4808	0.4950			1	1	2594	3028	0.7278	
		14	5551	5070	0.1513			10	9	1682	1954	0.2290			1	1	3224	2794	0.5954	
		15	2644	2561	0.0426			11	10	4477	4477	0.75			1	1	1589	1397	1.0000	
		16	2112	2323	0.5174			12	13	3311	3075	0.5416			1	1	1616	1416	0.4162	
		17	3482	3592	0.8093			13	14	2123	1448	0.3040			1	1	1390	1180	0.2579	
		18	4781	4684	0.3174			14	15	2140	2194	0.9271			1	1	3739	3301	0.5170	
		19	8650	5835	0.8501			15	16	2185	1787	0.9802			1	1	2191	2249	0.2226	
		20	6050	9103	0.0747			16	17	2124	1785	0.0220			1	1	4541	4230	0.5332	
		21	4121	2906	0.2765			17	18	2071	1627	0.45			1	1	2799	3093	0.4678	
		22	5725	5209	0.2026			18	19	17351	7142	0.2500			1	1	1796	1974	0.1944	
		23	5551	3724	0.1176			19	20	2069	2023	0.1379			1	1	2269	2314	0.9297	
		24	4120	3834	0.3806			20	21	4801	4604	0.2458			1	1	3282	2991	0.8927	
		25	2019	1735	0.0916			21	22	2348	2432	0.1486			1	1	2219	2559	0.6560	
		26	1565	2205	0.0920			22	23	2006	2322	0.1048			1	1	2229	1725	0.7165	
		27	3137	3853	0.7347			23	24	1974	2048	0.9242			1	1	1120	1070	0.2523	
		28	2762	2219	0.3020			24	25	2034	1844	0.5625			1	1	1737	2263	0.5686	
		29	4184	3273	0.7280			25	26	2532	2381	0.2515			1	1	1796	2621	0.7507	
		30	4474	4997	0.7444			26	27	3432	3436	0.8064			1	1	1814	1922	0.4857	
		31	9304	9769	1.0000			27	28	1637	1501	0.4111			1	1	4176	3610	0.7338	
		32	5053	5194	0.0716			28	29	3693	3717	0.7657			1	1	1032	1456	0.4794	
		33	4271	3573	0.0422			29	30	1792	1743	0.2997			1	1	2815	3285	0.3285	
		34	1686	1472	0.4797			30	31	5366	4233	0.4928			1	1	1879	2181	0.5000	
		35	1372	961	0.5533			31	32	1563	1095	0.2838			1	1	1333	2142	0.1921	
		36	5273	5153	0.5533			32	33	5841	1658	0.4756			1	1	1390	4708	0.5251	
		37	1844	1604	0.5625			33	34	2060	2320	0.6849			1	1	1710	2383	0.8810	
		38	1885	2618	0.0214			34	35	2254	2021	0.3779			1	1	1956	3001	0.0115	
		39	1177	1633	1.0000			35	36	2213	2072	0.8907			1	1	2291	2608	0.5699	
		40	6577	6428	0.3367			36	37	2200	2355	0.3117			1	1	1489	2354	0.5000	
		41	3228	2930	0.0716			37	38	2265	2265	0.0492			1	1	3105	3247	0.7037	
		42	2525	2414	0.1591			38	39	2120	1369	0.7055			1	1	1498	1455	0.6349	
		43	1964	1675	0.1687			40	41	2265	2265	0.0529			1	1	1510	2202	0.7432	
		44	2950	3019	0.3151			41	42	3605	4218	0.9837			1	1	2807	2789	0.2196	
		45	4616	4156	0.8162			42	43	1720	2047	0.5465			1	1	2843	3212	0.7627	
		46	6882	6882	0.7746			43	44	2220	2662	0.9997			1	1	3697	3550	0.5000	
		47	4018	4029	0.8002			44	45	2200	2577	0.1416			1	1	523	106	0.2502	
		48	1914	1755	0.8002			45	46	2129	2463	0.5823			1	1	2452	2910	0.5000	
		49	1671	1740	0.7204			46	47	2265	2625	0.0492			1	1	2024	1640	0.2500	
		50	1948	1935	0.5000			47	48	2120	1369	0.7055			1	1	1650	1890	0.2500	
		51	1619	2330	0.3758			48	49	2265	2265	0.0529			1	1	1512	1265	0.3095	
		52	2527	2527	0.3725			50	51	3605	5158	0.7424			1	1	1965	1190	0.0557	
		53	2654	3144	0.1574			51	52	1600	1573	0.7777			1	1	2807	2817	0.5195	
		54	4016	3763	0.5360			52	53	2017	1726	0.2220			1	1	2843	3212	0.7627	
		55	3781	4066	0.5020			53	54	1631	1628	0.8500			1	1	2024	1812	0.0011	
		56	1463	1558	0.5383			54	55	2006	2546	0.6581			1	1	2452	2910	0.5000	
		57	2789	3737	0.9342			55	56	2267	2708	0.3738			1	1	2024	1640	0.2500	
		58	2501	3050	0.9792			56	57	1465	1210	0.7422			1	1	1654	2061	0.5851	
		59	5077	4441	1.0000			57	58	2265	2258	0.2224			1	1	1762	2715	0.7049	
		60	6121	5483	1.0000			58	59	5929	5264	0.7899			1	1	2279	2418	0.3812	
		61	1403	5084	0.7426			59	60	2365	2365	0.6259			1	1	1571	1504	0.4122	
		62	3277	4010	0.5713			60	61	1374	1668	0.3500			1	1	2843	2877	0.2670	
		63	2405	3995	0.7078			61	62	1720	4174	0.4285			1	1	1924	1812	0.0011	
		64	3605	3605	0.7078			62	63	2336	2557	0.6515			1	1	2098	1624	0.2230	
		65	3064	3064	0.7500			63	64	2579	2579	0.0168			1	1	1442	1624	0.5100	
		66	3064	3064	0.7500			64	65	989	111	0.0069			1	1	1948	2049	0.9582	
		67	3064	3709	1.0000			65	66	1262	1863	0.2484			1	1	1441	1135	0.8328	
		68	2604	2730	0.7500			66	67	1924	2152	0.3488			1	1	1627	2030	0.3657	
		69	5077	2971	0.2500			67	68	2870	2870	0.0316			1	1	1714	1977	0.6647	
		70	2659	2681	0.5000			68	69	117	1995	0.9681			1	1	2604	2250	0.7118	
		71	2645	2681	0.5000			69	70	2411	2411	0.3000			1	1	1571	1978	0.7291	
		72	11	2900	2131	0.3901			70	71	1949	1900	0.7068			1	1	1055	1323	0.4768
		73	2492	3054	0.7360			71	72	3046	4214	0.7593			1	1	1998	1250	0.2500	
		74	2512	1962	0.0187			72	73	2209	2495	0.6584			1	1	646	1040	0.3773	
		75	2493	2823	0.5000			73	74	1808	1962	0.7422			1	1	2248	2590	0.3512	
		76	2502	2680	0.7500			74	75	1924	2152	0.3488			1	1	1199	1730	0.0014	
		77	4451	4451	0.0000			75	76	2874	3754	0.2997			1	1	2770	2653	0.2678	
		78	2509	2971	0.2500			76	77	1241	1241	0.3777			1	1	1746	2044	0.1622	
		79	2645	2681	0.5000			77	78	1943	1995	0.2870			1	1	1862	1470	0.0434	
		80	1	2900	2131	0.3901			78	79	2476	3775	0.0191			1	1	1201	1201	0.5000
		81	2492	3054	0.7360			79	80	1614	1515	0.9381			1	1	1304	861	0.6615	
		82	2512	1962	0.0187			80	81	108	108	1								

quite unsymmetrical compared to those found in other compounds (*cf.* Ref. 5). This is a consequence of the steric requirements due to the bond situation in the ring. The carbon atom C2 is situated within 0.02 Å in the intersection of the planes of the two benzene rings, whereas the carbon atoms C9 and C10 are situated in the planes of the benzene rings C3–C8 and C11–C16, respectively, (distances 0.01 Å and 0.04 Å). The carbon atom C9 is located 0.95 Å from the best least-squares plane through the four carbon atoms of the seven-membered ring which are common to the benzene rings, and the corresponding distance for C10 is 0.19 Å.

*Table 2.* Fractional coordinates for the atoms of one molecule. The numbering for non-hydrogen atoms is shown in Fig. 1. The hydrogen atoms are numbered in the same way as their parent atom with the addition of one last digit to indicate the number attached to the same parent atom.

	<i>x</i>	<i>y</i>	<i>z</i>
Br	0.2312	1.1352	0.0505
C(1)	0.2753	1.0849	0.1284
C(2)	0.2529	0.9502	0.1515
C(3)	0.2454	0.8176	0.1113
C(4)	0.0939	0.7873	0.0754
C(5)	0.0995	0.6593	0.0363
C(6)	0.2998	0.5666	0.0442
C(7)	0.4602	0.6159	0.0862
C(8)	0.4312	0.7422	0.1178
C(9)	0.6183	0.7849	0.1653
C(10)	0.5312	0.7556	0.2254
C(11)	0.3426	0.8351	0.2501
C(12)	0.2599	0.8151	0.3108
C(13)	0.0993	0.8915	0.3338
C(14)	-0.0228	0.9701	0.3040
C(15)	0.0144	0.9932	0.2444
C(16)	0.1889	0.9248	0.2169
H(11)	0.326	1.169	0.158
H(41)	-0.043	0.860	0.073
H(51)	-0.026	0.638	0.004
H(61)	0.324	0.465	0.021
H(71)	0.605	0.553	0.092
H(91)	0.758	0.721	0.157
H(92)	0.654	0.899	0.161
H(101)	0.484	0.640	0.225
H(102)	0.661	0.769	0.256
H(121)	0.337	0.733	0.338
H(131)	0.071	0.884	0.381
H(141)	-0.156	1.021	0.325
H(151)	-0.090	1.061	0.218

The C–Br bond seems to be significantly shorter than olefinic carbon–bromine distances in general ( $1.89 \pm 0.01$  Å according to *The International Tables for X-Ray Crystallography*, Vol. III (1962) p. 273). This might be an inductive effect due to the vicinity of the  $\pi$ -electron orbitals of the conjugated systems of the molecule.

Table 3. Vibration parameters. Anisotropic temperature factors were used in the refinement according to  $\exp[-2^2(h^2a^2U_{11} + k^2b^2U_{22} + l^2c^2U_{33} + 2klbcU_{23} + 2hlacU_{13} + 2hkabU_{12})]$ . The tensor elements in  $\text{\AA}^2$  have been multiplied by 1000.

	$U_{11}$	$U_{22}$	$U_{33}$	$U_{23}$	$U_{31}$	$U_{12}$
Br	712	257	417	109	81	17
C(1)	1108	156	401	95	-1064	13
C(2)	820	342	233	-69	-304	663
C(3)	641	227	-68	38	279	21
C(4)	725	491	19	206	-23	34
C(5)	889	219	402	-86	-571	7
C(6)	1335	295	1357	301	1880	457
C(7)	341	572	327	-110	13	-28
C(8)	97	187	534	142	-198	525
C(9)	809	449	486	44	8	-92
C(10)	-350	1226	389	487	149	307
C(11)	1067	362	285	-77	203	-259
C(12)	1207	652	197	7	964	95
C(13)	430	860	751	650	-98	-106
C(14)	-101	978	502	-336	123	112
C(15)	-266	304	334	-199	-174	37
C(16)	1388	271	-16	-11	-517	-94

Table 4. Bond lengths in  $\text{\AA}$  (with standard deviations  $\times 100$ ). Atom numbering according to Fig. 1.

Br	-C(1)	1.83(2)
C(1)	-C(2)	1.36(3)
C(2)	-C(3)	1.53(3)
C(2)	-C(16)	1.54(3)
C(3)	-C(4)	1.28(4)
C(4)	-C(5)	1.48(3)
C(5)	-C(6)	1.53(4)
C(6)	-C(7)	1.45(4)
C(7)	-C(8)	1.38(4)
C(3)	-C(8)	1.37(4)
C(8)	-C(9)	1.63(4)
C(9)	-C(10)	1.48(3)
C(10)	-C(11)	1.50(4)
C(11)	-C(12)	1.47(3)
C(12)	-C(13)	1.34(4)
C(13)	-C(14)	1.26(4)
C(14)	-C(15)	1.37(4)
C(15)	-C(16)	1.41(4)
C(11)	-C(16)	1.47(3)

The molecular packing viewed along the  $\alpha$ -axis is shown in Fig. 2. Only forces of van der Waals type determine the intermolecular packing. There are eight hydrogen-hydrogen distances less than 3  $\text{\AA}$  per molecule.

It is believed that the effect of this type of tricyclic compounds is due to their ability to inhibit the uptake in the neuronal membrane of nor-

Table 5. Bond angles in degrees (with standard deviations). Atom numbering according to Fig. 1.

Br—C(1)—C(2)	125.7(1.7)
C(1)—C(2)—C(3)	121.2(1.8)
C(1)—C(2)—C(16)	122.1(2.0)
C(3)—C(2)—C(16)	115.6(1.8)
C(2)—C(3)—C(4)	125.0(2.9)
C(2)—C(3)—C(8)	108.8(2.6)
C(4)—C(3)—C(8)	126.1(2.1)
C(3)—C(4)—C(5)	122.3(2.7)
C(4)—C(5)—C(6)	113.7(2.4)
C(5)—C(6)—C(7)	117.8(2.3)
C(6)—C(7)—C(8)	120.6(2.5)
C(7)—C(8)—C(9)	116.5(2.1)
C(7)—C(8)—C(3)	119.5(2.3)
C(3)—C(8)—C(9)	123.9(2.1)
C(8)—C(9)—C(10)	106.6(2.4)
C(9)—C(10)—C(11)	122.5(2.5)
C(10)—C(11)—C(12)	123.9(2.5)
C(10)—C(11)—C(16)	127.3(2.1)
C(16)—C(11)—C(12)	108.0(2.8)
C(11)—C(12)—C(13)	123.9(2.9)
C(12)—C(13)—C(14)	124.2(2.9)
C(13)—C(14)—C(15)	120.6(2.6)
C(14)—C(15)—C(16)	119.3(2.2)
C(15)—C(16)—C(2)	123.5(2.5)
C(15)—C(16)—C(11)	122.7(2.9)
C(11)—C(16)—C(2)	113.4(2.8)

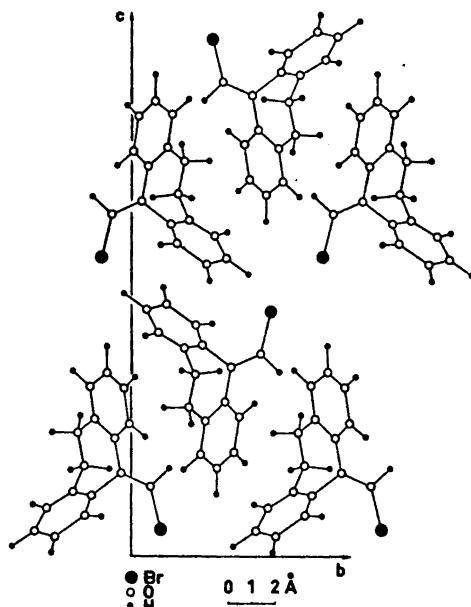


Fig. 2. Molecular packing of 5(bromomethylene)-10,11-dihydro-5H-dibenzo[a,d]-cycloheptene as seen along the  $a$ -axis.

epinephrine (*cf.* Ref. 6). Furthermore a study of penetration of lipid monolayers by psychoactive drugs indicate that the drug molecules are attached to the lipids of natural membranes. On the assumption that the lipids in the membrane form a bimolecular layer where the chains possess a liquid type of disorder a possible mechanism might be that the rigid skeleton of the drug molecules promote the formation of a more crystalline structure of the lipids and thus reduce the membrane permeability. The present author has started measurements of the effect of this compound with known molecular dimensions on the structure of lipid surface films.

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