The Phase Diagram SiO₂-CaF₂ LARS HILLERT

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The phase diagram for the system SiO_2 — CaF_2 has been constructed by the heat treatment, at various temperatures and oven atmospheres, of powder mixtures of compositions ranging from 100 % SiO_2 to 100 % CaF_2 and by subsequent cooling

help of a Guinier camera were also examined. The stable system has a eutectic point at 1240° corresponding to a composition of $47~\%~\mathrm{SiO_2}~+~53~\%~\mathrm{CaF_2}$, and a two-liquid area between 1420° and 1290° and the compositions $43~\%~\mathrm{SiO_2}~+~57~\%$ $\mathrm{CaF_2}$ and $23~\%~\mathrm{SiO_2}~+~77~\%~\mathrm{CaF_2}$. In the meta-stable system the compounds $\mathrm{SiO_2} \cdot 2\mathrm{CaF_2}$ and $\mathrm{SiO_2} \cdot 2\mathrm{CaF_2}$ appear below 1230° and 1210° , respectively, together with a glass-phase whose composition lies between $50~\%~\mathrm{SiO_2}~+~50~\%~\mathrm{CaF_2}$ and $47~\%~\mathrm{SiO_2}~+~53~\%~\mathrm{CaF_2}$. O CaF_2 and $47~\%~\mathrm{SiO_2}~+~53~\%~\mathrm{CaF_2}$. Other $\mathrm{CaF_2}$ -containing systems such as $\mathrm{TiO_3}-\mathrm{CaF_2}$, $\mathrm{Al_2O_3}-\mathrm{CaF_2}$, $\mathrm{SiO_2}-\mathrm{TiO_2}-\mathrm{CaF_2}$, $\mathrm{SiO_2}-\mathrm{Al_2O_3}-\mathrm{CaF_2}$, $\mathrm{TiO_2}-\mathrm{Al_2O_3}-\mathrm{CaF_2}$, $\mathrm{SiO_2}-\mathrm{Al_2O_3}-\mathrm{CaF_2}$, $\mathrm{TiO_2}-\mathrm{Al_2O_3}-\mathrm{CaF_2}$,

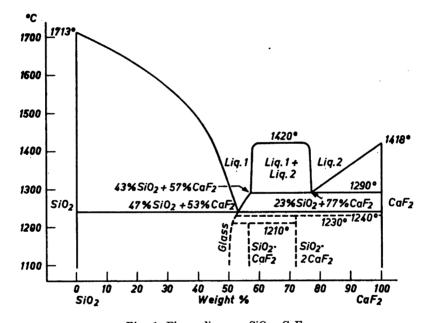


Fig. 1. Phase diagram SiO₂-CaF₂.

of the samples in different ways. The experiments were performed in a specially constructed apparatus, and the flow properties and microscopic appearance of the products obtained studied. X-Ray powder photographs obtained with the

TiO₂—CaO—CaF₂ and SiO₂—CaO—CaF₃ are now being investigated. Within certain parts of these systems it is already possible to state the preliminary appearance of the phase diagrams.

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