

## The Geometrical Structure of 3- $\beta$ -Indolylacrylic Acid

CHRISTOFFER RAPPE

Chemical Institute, University of Uppsala,  
Uppsala, Sweden

NMR-spectroscopy has proved to be a convenient tool in many fields of organic chemistry. One example is the determination of geometrical structures. *Trans*-compounds have larger spin coupling constants than *cis*-compounds; mean values of 17–18 cycles/sec for *trans*- and of 10–11 cycles/sec for *cis*-compounds are given.<sup>1</sup>

In connection with studies on plant growth regulators it was of interest to investigate the geometrical structure of 3- $\beta$ -indolylacrylic acid.<sup>2</sup> This acid was obtained by condensation of 3-formylindole with malonic acid.<sup>3</sup> Condensation with malonic acid usually give *trans*-isomers,<sup>4</sup> but in the present case no geometrical structure had been assigned to the product.

The NMR-spectrum of 3- $\beta$ -indolylacrylic acid was recorded and compared with several known *cis*- and *trans*-isomers of  $\alpha,\beta$ -unsaturated acids. The results are collected in Table 1. Both aliphatic and aromatic acids were investigated. Besides the spin coupling constants, aromatic acids show another difference between *cis*- and *trans*-isomers. The  $\beta$ -hydrogens of *trans*-acids have higher  $\delta$ -values than the

aromatic hydrogens, while the *cis*-acids have lower  $\delta$ -values.

Both these features are in accordance with a *trans* configuration for the known 3- $\beta$ -indolylacrylic acid. The latter has a coupling constant of 17.5 cycles/sec and the  $\beta$ -hydrogen doublet has a higher  $\delta$ -value than the aromatic hydrogens.

*Experimental.* The NMR-spectra were recorded on a Varian Associates Model A 60 spectrometer. 3- $\beta$ -Indolylacrylic acid. Commercial sample, Sigma Chemical Company, m.p. 192–194°C.

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Table 1.

Acid	Spin coupling constants cycles/sec	
	<i>cis</i> -	<i>trans</i> -
CH <sub>3</sub> -CH=CH-COOH	12	16.5
C <sub>6</sub> H <sub>5</sub> -CH=CH-COOH	12	—
<i>n</i> -C <sub>3</sub> H <sub>7</sub> -CH=CH-COOH	12	17
<i>iso</i> -C <sub>3</sub> H <sub>7</sub> -CH=CH-COOH	12	—
BrCH=CH-COOH	8.5	14
ClCH=CH-COOH	8.5	—
C <sub>6</sub> H <sub>5</sub> -CH=CH-COOH	13	17
(4)ClC <sub>6</sub> H <sub>4</sub> -CH=CH-COOH	13	17
(3,4)Cl <sub>2</sub> C <sub>6</sub> H <sub>3</sub> -CH=CH-COOH	13	—
(3,5)Cl <sub>2</sub> C <sub>6</sub> H <sub>3</sub> -CH=CH-COOH	13	17
C <sub>8</sub> H <sub>7</sub> N-CH=CH-COOH	—	17

## The Effect of Sodium Salicylate on Hexosamine Synthesis in Evi- scerated Mouse Fetuses

BERNARD JACOBSON, HARRY  
BOSTRÖM and K. SUNE LARSSON

Department of Metabolic Research, Wenner-  
Gren Institute, University of Stockholm and  
Department of Anatomy, Karolinska Institutet,  
Stockholm, Sweden

Several anti-inflammatory drugs have been found to depress the synthesis of acid mucopolysaccharides in various tissues<sup>1,2</sup> and in addition, to produce cleft-palate, as well as other skeletal and vascular malformations in mouse em-

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