

Hydrolysis of Palatinose (Isomaltulose) by Pig Intestinal Glycosidases

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Homogenates of intestinal mucosa from adult pigs hydrolyze palatinose (6-(α -D-glucopyranosyl)-D-fructose).

The palatinase activity is caused by a mixture of three separate enzymes, namely *specific isomaltase*, *maltase II* and *maltase III*.

These three enzymes also hydrolyze isomaltose, but the relative rates of hydrolysis of isomaltose and palatinose during the conditions used for glycosidase activity determination are different for the three enzymes.

Maltase III alone exerts about 50 % of the total palatinase activity of crude homogenates of pig intestinal mucosa.

The isolation of a new disaccharide, *palatinose* (isomaltulose, (6-(α -D-glucopyranosyl)-D-fructose), formed by a bacterium isolated from sugar beets, was recently described by Weidenhagen and Lorenz¹⁻³. This disaccharide is hydrolysed by some species of yeast. The hydrolysis seems not, however, to be caused by maltase (α -glucosidase), as might have been expected, but by a specific *palatinase*⁴.

It is not known whether the glycosidases of the *intestinal mucosa* are able to hydrolyze palatinose or not. This question is of interest as the effect of orally administered palatinose on intestinal flora is currently under study⁵. It is also important for the further characterization of the specificity of the different pig intestinal α -glucosidases recently separated⁶. Professor Weidenhagen kindly supplied the author with a sample of palatinose for the study of the action of pig intestinal glycosidases on this sugar.

MATERIALS AND METHODS

Intestinal glycosidase preparations

The mucosa and the contents of the small intestine of adult pigs were obtained by pressing the intestine between two rollers immediately after the slaughter. The whole length of the small intestine was used. In the first experiments the preparation obtained was homogenized in an Ultra-Turrax homogenizer for 2 min⁷, and after centrifugation

in an ordinary laboratory centrifuge for 5 min the opalescent supernatant was analyzed immediately. Later preparations were used which had been solubilized and thereafter purified by ethanol precipitation as described earlier ⁶.

Substrates

Sucrose (2-(α -D-glucopyranosyl)- β -D-fructofuranoside) (cryst.) was obtained from Baker Co. (U.S.A.) and *maltose* (4-(α -D-glucopyranosyl)-D-glucose) (cryst., monohydrate) from Merck A.G. (Germany). *Isomaltose* (6-(α -D-glucopyranosyl)-D-glucose) and *phenyl- α -D-glucopyranoside* were prepared by the methods described earlier ^{8,9}. *Palatinose* (isomaltulose, 6-(α -D-glucopyranosyl)-D-fructose) (cryst., monohydrate) was kindly supplied by Professor R. Weidenhagen (Neuoffstein, Germany).

Glycosidase activity determinations

Methods for the determination of *invertase*, *maltase*, *isomaltase* and *phenyl- α -D-glucopyranosidase* activities have been described in a previous paper ¹⁰.

Palatinase activity was determined by the use of glucose oxidase in the same way as has been described earlier for *turanase* and *lactase* activity determinations ¹⁰. The substrate solution was prepared by dissolving 1.90 g of palatinose in 0.10 M maleate buffer pH 6.5 to a final volume of 100 ml* (cf. Table 5 in Ref.¹⁰). In accordance with the other glycosidase activities one unit of palatinase activity is defined as the activity causing 5 % of hydrolysis in 2.0 ml of reaction mixture in 60 min at 37°C.

Protein determinations

The determination of protein was performed with the method of Lowry *et al.*¹¹ using the modified reagent B introduced by Eggstein and Kreutz ¹². A standard curve was prepared with human serum albumin, kindly supplied by A. B. Kabi.

Mutual displacement chromatography

The mutual displacement technique as applied to the chromatography of glycosidases upon TEAE-cellulose columns has been described in a previous paper ¹³. The TEAE-cellulose was prepared in the laboratory from Solkaflor cellulose powder SW 40 A ¹³. The ion exchanger was stored and used in its bromide form. Commercial TEAE-cellulose (from Serva Entwicklungslabor, Germany) was found unsuitable for columns of the dimensions used (1 \times 100 cm) since the flow rate became too low.

RESULTS AND DISCUSSION

Palatinase activity of crude intestinal preparations

Fresh intestinal mucosa homogenates from adult pigs hydrolyzed palatinose with a rate that was 0.3–0.5 times the rate for the hydrolysis of isomaltose; 0.1–0.2 times that for sucrose; and 0.02–0.05 times that for maltose during the conditions used for glycosidase activity determination.

* Since palatinose did not decrease in weight on storage *in vacuo* over P₂O₅ at 50°C it was believed to be anhydrous. The sugar contains, however, one mole of crystal water which is difficultly removed ⁵. The substrate solution therefore should contain 2.0 g of palatinose per 100 ml ¹⁰. The activity measured at the somewhat lower substrate concentration used here is, however, only about 2 % too low.

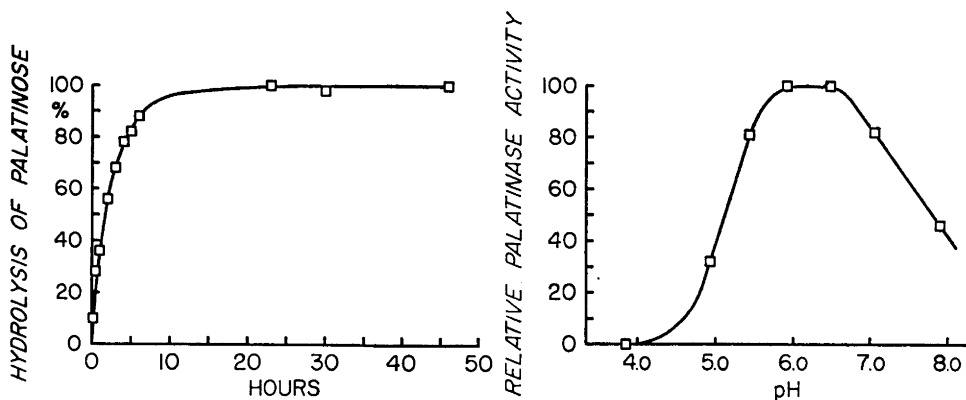


Fig. 1. Hydrolysis of palatinose by a crude extract of pig intestinal mucosa at 0.027 M substrate concentration in 0.05 M maleate buffer pH 6.5 at 37°C. Toluene was used as a preservative. The hydrolysis proceeded to completion.

Fig. 2. Influence of pH on the palatinase activity of a crude extract of pig intestinal mucosa. Substrate concentration 0.027 M. Temperature 37°C. Buffers: 0.05 M acetate buffer pH 4–5, 0.05 M maleate buffer pH 5.5–6.5, and 0.025 M veronal buffer pH 7–8.

Solubilization of the preparation by the method described earlier⁶, and precipitation of the solubilized preparation with ethanol^{6,7}, did not alter the ratio of palatinase activity to the isomaltase, invertase or maltase activity. Like these activities, therefore, the palatinase activity is undamaged by solubilization and ethanol precipitation, and for the experiments described below preparations treated in this way were used.

The hydrolysis of palatinose by crude intestinal glycosidase preparations proceeded to completion (Fig. 1). The palatinase activity had its optimum pH at 6.0–6.5 (Fig. 2).

Experiments were then performed to determine whether the intestinal palatinase activity was exerted by a specific palatinase, as in yeast⁴, or by one or several of the intestinal α -glucosidases previously separated⁶.

Separation of palatinase from trehalase

The fact that the palatinase activity was undamaged by ethanol precipitation seemed to indicate that intestinal *trehalase* was without palatinase activity, since this enzyme is inactivated by ethanol⁷. This also became evident from the analysis of a purified intestinal trehalase preparation, obtained as described elsewhere¹⁴. Intestinal trehalase had no detectable palatinase activity under conditions such that hydrolysis of palatinose with a rate only 1/250 of that of trehalose would have been detected.

The remaining intestinal α -glucosidases to be considered were: *invertase* (=maltase I), *maltase II*, *maltase III*, *specific isomaltase* and *specific phenyl- α -D-glucopyranosidase*^{6,9}.

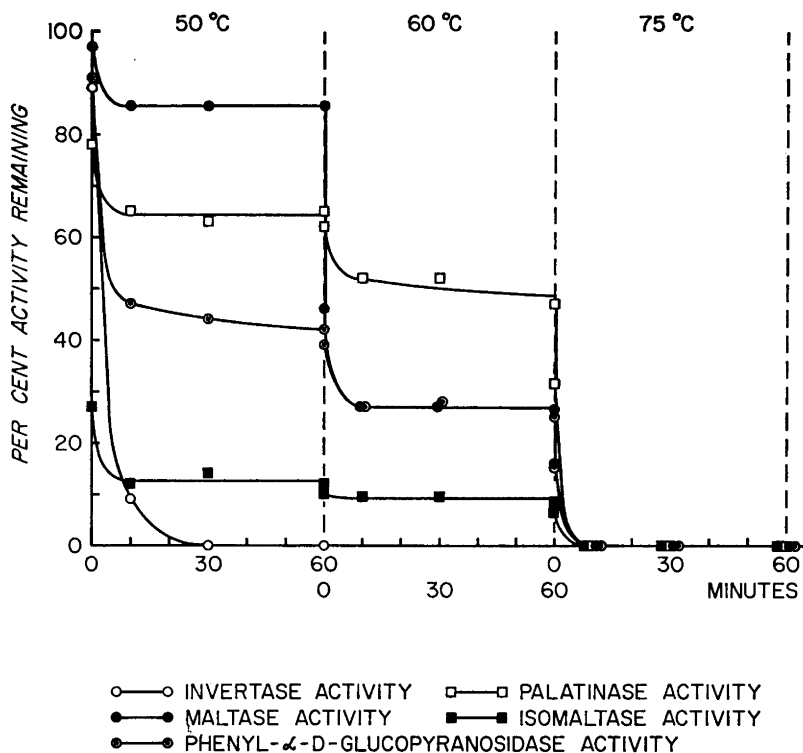


Fig. 3. Heat inactivation of intestinal α -glucosidases in three separate steps. The enzyme preparation contained 2.0 mg/ml of protein in 0.01 M phosphate buffer pH 7.0.

Step-wise heat inactivation of a crude intestinal glycosidase preparation

It has been demonstrated earlier that the differences between pig intestinal invertase (= maltase I), maltase II and maltase III, with regard to their sensitivity to heat, is so great that these enzymes may be heat-inactivated in three separated steps under appropriate conditions¹³. The results of such an experiment, in which also the palatinase, isomaltase, and phenyl- α -D-glucopyranosidase activities were followed are seen in Fig. 3.

The solubilized crude intestinal glycosidase preparation used for this experiment contained 2.0 mg of protein per ml in 0.01 M phosphate buffer pH 7.0. The α -glucosidase activities of the preparation were as follows; invertase 93 units/ml, maltase (total) 350 units/ml, palatinase 10.8 units/ml, isomaltase (total) 30 units/ml and phenyl- α -D-glucopyranosidase (total) 13.5 units/ml. The technique used for heat inactivation has been described earlier^{13,15}.

The heat inactivation experiment illustrated in Fig. 3 was performed at pH 7.0 at all temperatures. In the experiments previously reported, the pH

was lowered to 6.0 before heating at 60°C¹³. Later this precaution was found to be unnecessary.

A. Inactivation at 50°C. During heating at 50°C the invertase activity (93 units/ml) was completely abolished, together with 60 units/ml of maltase activity (this fraction amounting to 17 % of the total maltase activity of the preparation), (Fig. 3). This fraction of the maltase activity is known to be exerted by the invertase (=maltase I)¹⁶. The maltase/invertase activity quotient for this enzyme as calculated from the present experiment is 0.65, which agrees well with the figure reported earlier¹⁰.

At this temperature 26 units/ml (87 % of the total) of the isomaltase activity and 7.5 units/ml (56 %) of the phenyl-*α*-D-glucopyranosidase activity were inactivated (Fig. 3). Since purified intestinal invertase (= maltase I) previously has been obtained which exerted neither isomaltase nor phenyl-*α*-D-glucopyranosidase activity⁸, these activities are caused by one or two other enzymes (specific isomaltase and specific phenyl-*α*-D-glucopyranosidase^{6,9}).

Of the palatinase activity 3.8 units/ml (35 %) was inactivated at 50°C (Fig. 3). This fraction of the palatinase activity may be caused by any of the different enzymes inactivated at this temperature.

B. Inactivation at 60°C. During heating at 60°C the maltase activity decreased by further 198 units/ml (57 % of the original maltase activity of the preparation), this fraction corresponding to the maltase II present in the preparation^{6,8,13}. In the same time the isomaltase activity decreased by 1.5 units/ml (5 % of the original value) and the phenyl-*α*-D-glucopyranosidase by 2.5 units/ml (19 %). These activities are known to be exerted by the maltase II *per se*⁸. The isomaltase/maltase and phenyl-*α*-D-glucopyranosidase/maltase activity quotients for this enzyme were both calculated to be about 0.01. As was to be expected¹⁰ these quotients are somewhat lower than those earlier obtained with the use of more concentrated substrate⁸.

The palatinase activity decreased by 2.0 units/ml (19 % of the original value) during heating at 60°C. There seems to be reason to suppose that this part of the palatinase activity is caused by the maltase II *per se*. Under this assumption the palatinase/maltase activity quotient for maltase II may be calculated to be 0.01, *i.e.* palatinose and isomaltose, which are structurally very similar to each other, are hydrolyzed at about the same rate by this enzyme during the conditions used for glycosidase activity determination.

C. Inactivation at 75°C. All the glycosidase activities hitherto remaining were completely abolished during heating at 75°C. The fraction of the maltase activity which was inactivated in this step amounted to 92 units/ml (26 % of the original activity); isomaltase, 2.5 units/ml (8 %); phenyl-*α*-D-glucopyranosidase, 3.5 units/ml (26 %), and palatinase 5.0 units/ml (46 %).

The fractions of the maltase, isomaltase and phenyl-*α*-D-glucopyranosidase activities inactivated in this step are known to be exerted by one single enzyme, maltase III^{8,17}. The isomaltase/maltase activity quotient (0.03) and the phenyl-*α*-D-glucopyranosidase/maltase activity quotient (0.04) calculated from the present experiment are comparable with the figures earlier obtained¹⁰.

Using the assumption that the fraction of the palatinase activity inactivated in this step is that caused by maltase III *per se*, (for further evidence for this assumption, see below), the palatinase/maltase activity quotient for this enzyme may be calculated to be 0.05.

Palatinase activity of purified
invertase and maltase III preparations

Purified invertase (= maltase I) and maltase III were obtained by mutual displacement chromatography on TEAE-cellulose columns of a solubilized intestinal glycosidase preparation as described in a previous paper¹³. The concentrated phosphate buffer present in the fractions obtained was removed by dialysis, before heat inactivation experiments were performed.

A. Purified invertase (= maltase I). The invertase preparation obtained by mutual displacement chromatography contained 930 units/ml of invertase and 6.1 mg/ml of protein. It had maltase/invertase activity quotient 0.7, and thus contained no maltase II or maltase III. The preparation had palatinase activity, 72 units/ml. It had, however, also isomaltase activity, 352 units/ml, and phenyl-*α*-D-glucopyranosidase activity, 124 units/ml, *i.e.* the preparation contained, in addition to invertase, specific isomaltase and specific phenyl-*α*-D-glucopyranosidase.

On earlier occasions invertase preparations have been obtained by this method which had neither isomaltase nor phenyl-*α*-D-glucopyranosidase activities⁸. It has also been observed that the specific isomaltase is rapidly inactivated during adsorption to anion exchange cellulose⁹. In the present experiment it was therefore somewhat astonishing to find that the specific isomaltase activity had survived the mutual displacement chromatography which involved adsorption of the enzymes to TEAE-cellulose at room temperature (20°C) for about 24 h. As far as the author is aware the only difference between this experiment and that described earlier is that in the present case a solubilized intestinal glycosidase preparation was used, which had been stored in the frozen state (-16°C) for several months, while in the earlier case a fresh preparation was used. Storage in the frozen state is known to affect the specific isomaltase in a complicated way, causing an initial decrease, followed by a slower increase of the activity of this enzyme⁹, and may also alter the stability of the enzyme during adsorption to TEAE-cellulose.

To reveal which enzyme in the preparation was responsible for the palatinase activity, attempts were made to separate the activities by heat inactivation. At pH 7.0 (in 0.01 M phosphate buffer) the palatinase and isomaltase activities were found to be more sensitive to heat than the invertase and phenyl-*α*-D-glucopyranosidase activities. The course of the inactivation at 32°C is seen in Fig. 4. At this temperature the palatinase and isomaltase activities were slowly inactivated, and the inactivation of these activities ran parallel, while the invertase and phenyl-*α*-D-glucopyranosidase activities were only very slightly affected. It seems justifiable, therefore, to suppose that the most heat-labile fraction of the palatinase activity (*i.e.* the fraction inactivated at 50°C in the experiment with the crude preparation described above) is exerted by the specific isomaltase. The palatinase/isomaltase activity quotient for this enzyme may be calculated to be 0.2. The experiment also shows that specific isomaltase and specific phenyl-*α*-D-glucopyranosidase are two separate enzymes.

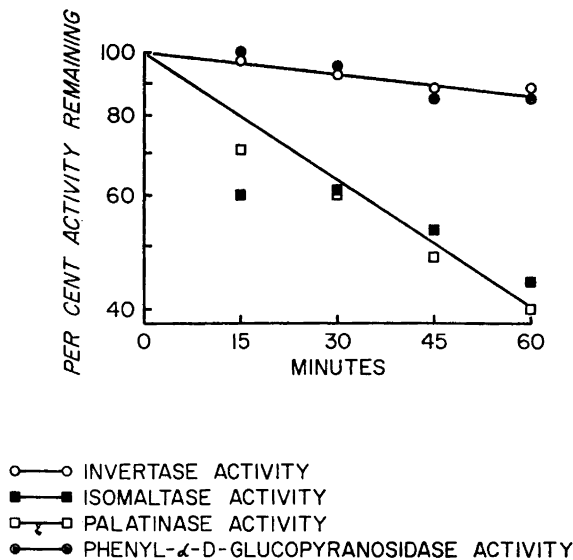


Fig. 4. Heat inactivation of a purified invertase preparation (obtained by mutual displacement chromatography) at 32°C in 0.01 M phosphate buffer pH 7.0. The preparation contained specific isomaltase and specific phenyl- α -D-glucopyranosidase as contaminants (see text). The enzyme solution used for heat inactivation contained 0.2 mg/ml of protein, 37 units/ml of invertase, 12.4 units/ml of isomaltase, 4.0 units/ml of phenyl- α -D-glucopyranosidase and 2.5 units/ml of palatinase.

Inactivation at so low temperature as 32°C occurs only with purified preparations at low salt concentration. When a homogenate of intestinal mucosa was heated at 37°C for 2 h, no inactivation of the glycosidase activities investigated was observed. There is therefore no reason to suppose any heat inactivation of intestinal glycosidases to occur *in vivo* at body temperature. Nor does inactivation occur during the conditions for glycosidase activity determinations (at 37°C) even with purified preparations. Experiments with varying incubation time showed the amount of monohexoses formed during the last half hour of the incubation to be the same as the amount formed during the first half hour.

B. Purified maltase III. The purified maltase III preparation obtained by mutual displacement chromatography contained 304 units/ml of maltase and 0.8 mg/ml of protein. It had no invertase activity. The preparation contained 13 units/ml of isomaltase, 15 units/ml of phenyl- α -D-glucopyranosidase and 22 units/ml of palatinase. The isomaltase and phenyl- α -D-glucopyranosidase activities of the preparation are known to be caused by the maltase III *per se* ^{8,10,17}.

During heating of the preparation at 71°C (in 0.01 M phosphate buffer pH 6.0) the maltase and palatinase activities of this preparation were inactivated in a parallel manner, indicating the palatinase activity of this preparation to be also caused by the maltase III *per se* (Fig. 5). The palatinase/maltase activity quotient for this enzyme was calculated to be 0.06.

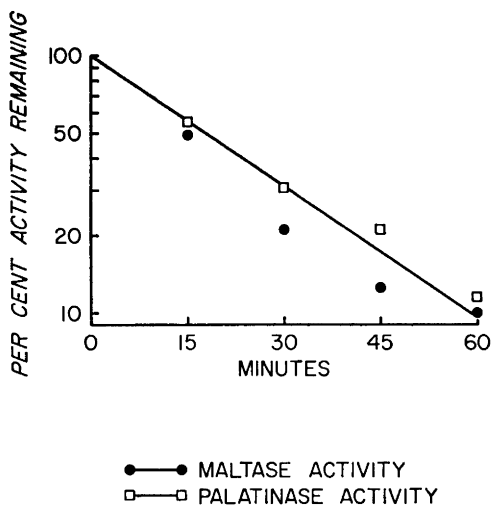


Fig. 5. Heat inactivation of a purified maltase III preparation at 71°C in 0.01 M phosphate buffer pH 6.0. The enzyme solution used for heat inactivation contained 0.15 mg/ml of protein, 55 units/ml of maltase and 2.6 units/ml of palatinase.

Relations between palatinase and isomaltase activity

From the experiments described above, it appears that palatinose is hydrolyzed by the same intestinal glycosidases as hydrolyze isomaltose, *i.e.* *specific isomaltase*, *maltase II* and *maltase III*. This is in harmony with the structural similarity between the two substrates. It is apparent that none of the enzymes is absolutely specific for the structure around the 1st and 2nd carbon atom of the *aglycon* of the substrate.

Nevertheless the structural difference between isomaltose and palatinose causes marked differences in the relative rates of hydrolysis of the two substrates by the different enzymes. During the conditions used for glycosidase activity determinations *specific isomaltase* hydrolyzes palatinose with a rate that is only one fifth of the rate for hydrolysis of isomaltose by the same enzyme. *Maltase II* hydrolyzes the two substrates at approximately the same rate, and *maltase III* hydrolyzes palatinose nearly twice as fast as isomaltose.

These differences make maltase III the most important enzyme for the hydrolysis of palatinose by crude extracts of intestinal mucosa, for this enzyme causes about 50 % of the total palatinase activity of such preparations, although it causes only about 10 % of the total isomaltase activity.

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REFERENCES

1. Weidenhagen, R. and Lorenz, S. *Z. Zuckerind.* **7** (1957) 533.
2. Weidenhagen, R. and Lorenz, S. *Angew. Chem.* **69** (1957) 641.
3. Lorenz, S. *Z. Zuckerind.* **8** (1958) 535.
4. Emeis, C. C. and Windisch, S. *Z. Zuckerind.* **10** (1960) 248.
5. Weidenhagen, R. *Personal communication.*
6. Dahlqvist, A. *Hog Intestinal α -Glucosidases.* Diss. Lund 1960.
7. Borgström, B. and Dahlqvist, A. *Acta Chem. Scand.* **12** (1958) 1997.
8. Dahlqvist, A. *Acta Chem. Scand.* **14** (1960) 1.
9. Dahlqvist, A. *Acta Chem. Scand.* **14** (1960) 72.
10. Dahlqvist, A. *Acta Chem. Scand.* **14** (1960) 1797.
11. Lowry, O. H., Rosebrough, N. J., Farr, A. L. and Randall, R. J. *J. Biol. Chem.* **193** (1951) 265.
12. Eggstein, M. and Kreutz, F. H. *Klin. Wochschr.* **33** (1955) 879.
13. Dahlqvist, A. *Acta Chem. Scand.* **13** (1959) 1817.
14. Dahlqvist, A. *Acta Chem. Scand.* **14** (1960) 9.
15. Dahlqvist, A. *Acta Chem. Scand.* **13** (1959) 945.
16. Dahlqvist, A. *Acta Chem. Scand.* **14** (1960) 63.
17. Dahlqvist, A. *Acta Chem. Scand.* **13** (1959) 2156.

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