

## The Crystal Structure of HfAl

LARS-ERIK EDHAMMAR

*Institute of Inorganic and Physical Chemistry, University of Stockholm, Stockholm, Sweden*

The crystal structure of HfAl is of the  $B_f$  (CrB) structure type. The orthorhombic unit cell, space-group  $Cmcm$  (No. 63), has the dimensions  $a = 3.25_3$  Å,  $b = 10.83_1$  Å and  $c = 4.28_2$  Å. A comparison between HfAl and other  $B_f$  structures is given.

Studies on arc-melted hafnium-aluminium alloys by means of X-ray powder methods have revealed the existence of the phases  $Hf_3Al_2$ , HfAl,  $Hf_2Al_3$ ,  $HfAl_2$  and  $HfAl_3$ . The structure types of  $Hf_3Al_2$ ,  $HfAl_2$  and  $HfAl_3$  were reported in a previous note<sup>1,2</sup>. In the course of further studies on this system the crystal structure of HfAl has been derived.

### EXPERIMENTAL

The alloy was prepared by arc-melting an equimolecular mixture of hafnium metal (98.5 % from I.C.I) and high-purity aluminium in an argon atmosphere. Crystals of HfAl were obtained by crushing the melt. They formed thin plates that could be cut down to dimensions suitable for single-crystal studies. A complete set of data was registered in a Weissenberg camera using  $CuK\alpha$  radiation and rotating the crystal around an axis situated in the plane of the crystal ( $a$  axis). The intensities of the reflexions were estimated visually.

Accurate unit cell dimensions were obtained from photographs taken in a Guinier focusing camera using strictly monochromatized  $CuK\alpha_1$  radiation and applying a procedure described in a previous communication from this Institute<sup>3</sup>.

### DERIVATION OF THE STRUCTURE

The single crystal data obtained showed the structure to be orthorhombic. The unit cell dimensions derived from the Guinier powder photograph were:

$$a = 3.25_3 \text{ Å}, \quad b = 10.83_1 \text{ Å}, \quad c = 4.28_2 \text{ Å}.$$

The observed density of 8.97 indicated that the unit cell contains 4 formula units of HfAl, the calculated density being 9.07. The powder pattern of HfAl is given in Table 1.

Table 1. The Guinier powder pattern of HfAl (CuK $\alpha_1$  radiation)

<i>hkl</i>	$\sin^2\theta_{\text{obs}}$	$\sin^2\theta_{\text{calc}}$	$I_{\text{obs}}$	$I_{\text{calc}}$
020	0.02021	0.02023	m	7.0
021	0.05257	0.05259	vst	24.4
110	0.06109	0.06111	mw	4.8
040	0.08091	0.08091	w	3.1
111	0.09351	0.09347	vvst	30.3
130	0.10150	0.10156	st	16.9
041	0.11324	0.11327	st	12.6
002	0.12945	0.12944	m	7.5
131	0.13398	0.13392	vw	0.6
022	0.14970	0.14967	vw	1.6
060 } 150 }	0.1823	0.18205 } 0.18248 }	m	3.0 2.4
112	0.19056	0.19055	w	2.2
042	0.21040	0.21035	w	2.1
061	—	0.21441	—	0.1
151	0.21486	0.21484	m	7.9
200	0.22418	0.22420	w	3.8
132	0.23101	0.23100	st	12.2
220	0.24444	0.24444	vw	0.9
221	0.27673	0.27679	mw	5.7
170	0.30400	0.30384	w	2.2
240	0.30504	0.30511	vw	1.5
023 } 062 } 152 }	0.3115	0.31147 } 0.31149 } 0.31192 }	mst	2.6 3.4 2.7
080	0.32358	0.32365	vw	0.6
171 } 241 }	0.3370	0.33620 } 0.33747 }	mw m	5.7 8.3
113 } 202 }	0.3530	0.35235 } 0.35364 }	mw mw	7.3 5.5
081	0.35606	0.35601	w	4.1
043	0.37209	0.37215	w	4.1
133	—	0.39280	—	0.2
260	0.40626	0.40625	w	3.4
172 } 242 }	0.4335	0.43328 } 0.43455 }	m	4.5 3.1
082	0.45322	0.45309	vw	1.4
190	0.46557	0.46567	mw	4.5
063 } 153 }	0.4735	0.47329 } 0.47372 }	mw	0.1 7.0

Reflexions  $hkl$  were observed only for  $h + k = 2n$ ,  $0kl$  for  $k = 2n$ ,  $h0l$  for  $h = 2n$  and  $l = 2n$  and  $hk0$  for  $h + k = 2n$ . The probable space-groups thus are  $Cmcm$  (No. 63),  $Cmc2_1$  (No. 36) and  $Ama2$  (No. 40). The symmetry, the unit-cell dimensions and the appearance of the Patterson projection along [100] suggested that the phase is isomorphous with CrB. Both the hafnium and aluminium atoms are thus in the point positions 4(c) of the space-group  $Cmcm$ :

$$\pm (0, y, \frac{1}{4}); \quad \pm (\frac{1}{2}, \frac{1}{2} + y, \frac{1}{4})$$

Approximate values of the  $y$  parameters were obtained from the Patterson projection and the signs of the  $F(0kl)$  could be calculated. The projection of

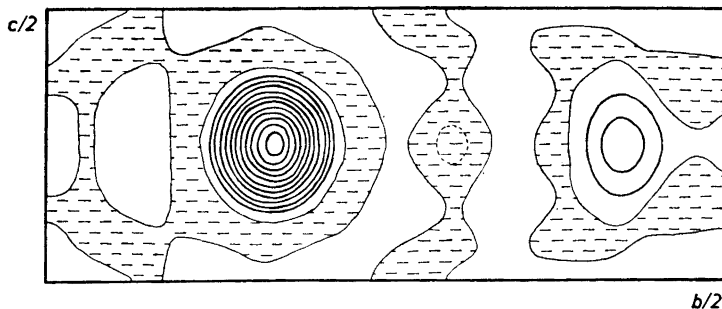


Fig. 1. a). Projection along [100] of the electron density of HfAl.

the electron density along [100] thus calculated (*cf.* Fig. 1a) gave the following parameter values:

$$y_{\text{Hf}} = 0.167 \quad y_{\text{Al}} = 0.42_5$$

The difference synthesis  $\rho (F(0kl)_{\text{obs}} - F(0kl)_{\text{Hf}})$  gave the same value for the  $y_{\text{Al}}$  parameter (*cf.* Fig. 1b). However, the accuracy of the aluminium position is rather low due to the relatively low scattering power of this atom. The index of agreement

$$\frac{\sum || F(0kl)_{\text{obs}} | - | F(0kl)_{\text{calc}} ||}{\sum | F(0kl)_{\text{obs}} |}$$

corresponding to this structure is 0.11. Table 1 gives a comparison between calculated and observed powder intensity data. The interatomic distances are given in Table 2.

#### DISCUSSION

Following Frank and Kasper<sup>4</sup>, the structure of HfAl can be described as a  $3^3 \cdot 4^2$  tessellation in the  $z = \frac{1}{4}$ -plane. This tessellation repeated by the  $2_1$  axes parallel to [001] is demonstrated in Fig. 2. It is interesting to note that the zig-zag aluminium chains thus formed (A—B in Fig. 2) have quite normal

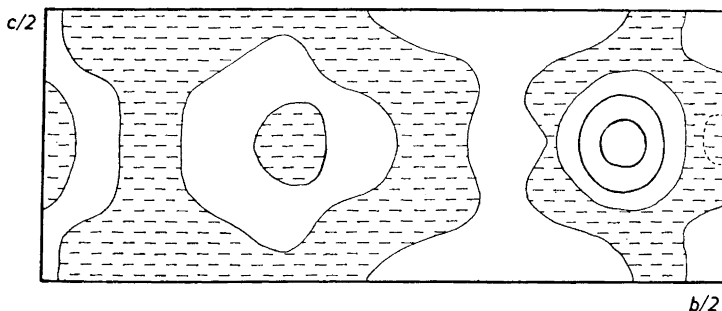


Fig. 1. b). The difference synthesis  $\rho (F(0kl)_{\text{obs}} - F(0kl)_{\text{Hf}})$ . Shaded areas indicate negative values.

Table 2. Interatomic distances in HfAl.

Hf	4 Hf	3.26 Å	Al	4 Hf	2.86 Å
	2 Hf	3.25		2 Hf	3.07
	4 Al	2.86		1 Hf	2.82
	2 Al	3.07		2 Al	3.25
	1 Al	2.82		2 Al	2.69

distances of about 2.69 Å. The corresponding boron-boron distances in CrB determined by Kiessling are extremely short and are probably due to boron-boron bonds of a covalent type<sup>5,6</sup>.

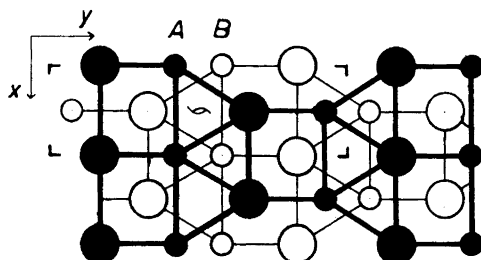


Fig. 2. The  $3^3 \cdot 4^2$  tessellation of HfAl. Large (Hf) and small (Al) black circles represent atoms at  $z = \frac{1}{2}$  and open ones atoms at  $z = \frac{3}{2}$ . The unit cell is indicated by  $\llcorner$ .

The isomorphism between HfAl and ThAl (described by Braun<sup>7</sup> on a different system of axes) is also of considerable interest since the aluminium-aluminium distances along the zig-zag chains of the latter phase are reported to be as short as 2.46 Å. It might be that this is a matter of the relative size of the atoms, the ratios  $r_{\text{Th}}/r_{\text{Al}}$  and  $r_{\text{Cr}}/r_{\text{B}}$  being approximately the same and substantially higher than  $r_{\text{Hf}}/r_{\text{Al}}$ . However, the analogous crystal structure of ThCo does not support this assumption<sup>8</sup> since the cobalt-cobalt distances are 2.77 Å in spite of the atomic radius of cobalt being less than that of aluminium.

The atomic distances within the chains of several other phases of  $B_2$  structure, uncluding CaSi<sup>9</sup>, CaGe<sup>10</sup>, CaSn<sup>10</sup> and YSi<sup>11</sup>, are throughout of the long "normal" type.

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