

Formation of Particulate Ribonucleoprotein with Enzymatic Activities *in vitro*

AGNAR P. NYGAARD

Johan Throne Holst's Institutt for Ernæringsforskning, University of Oslo, Norway

During the purification of lactic dehydrogenase from baker's yeast¹ a particulate ribonucleoprotein was observed to be formed in a soluble preparation of proteins and ribonucleic acid. The solution was obtained from the mechanically extracted cells by acetone precipitation, calcium phosphate adsorption/elution and ammonium sulfate precipitation. The nucleoprotein precipitated on storage at 6°C from the saltfree solution which had been concentrated by lyophilization.

The precipitate contained about equal amounts of nucleic acid and protein. Adenine, guanine, cytosine and uracil were all present. The nucleoprotein (nucleic acid and protein) was insoluble in water and in different salt solutions (*e. g.* 1 M sodium chloride), but dissolved in 0.04 M sodium hydroxide. In some preparations the particles settled easily in a low centrifugal field, in others a clear solution was not obtained even after 30 min centrifugation at $30\,000 \times g$.

The original solution contained 10 % ribonucleic acid, 5–10 % each of lactic-, malic-, alcohol- and triosephosphate dehydrogenase, and had proteolytic and nucleolytic activities. The well-washed particles had malic dehydrogenase activity. None of the other dehydrogenases were present. The particles digested the particulate ribonucleic acid. Thus, on incubation in phosphate buffer purine and pyrimidine material appeared in the supernatant. This material was not liberated by boiling of the particulate ribonucleoprotein.

Malic dehydrogenase was measured spectrophotometrically at 340 m μ , with oxalacetic acid and DPNH at pH 8.4, with malic acid and DPN at pH 9.6. The rate was 40 μ moles oxalacetic acid reduced per mg particulate protein per hour. Highly purified malic dehydrogenase has been obtained from the original ammonium sulfate precipitate by zone electrophoresis on cellulose. This soluble preparation did not contain ribonucleic acid and had an activity of 8 500 μ moles oxalacetic acid reduced per mg protein per hour.

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The results indicate that particulate ribonucleoproteins isolated from bacterial cells may not necessarily represent original particles of the cell. Particles may be formed in a haphazard manner during extraction and purification.

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Standard Deviations of Interatomic Distances in Completely Deuterated Benzene

S. J. CYVIN

Institutt for teoretisk kjemi, Norges tekniske høgskole, Trondheim, Norway

In a previous communication the calculation of standard deviations of interatomic distances in the benzene molecule has been reported¹. The calculated values have been compared with values obtained from electron diffraction data².

Similar calculations from spectroscopic data have been performed for the benzene-d₆ molecule with the result given in Table 1. The experimental fundamental

Table 1. Standard deviations for interatomic distances in benzene-d₆.

Distance	Calculated standard deviations in Å units			
	$T = 0$	273.16	298.16	323.16
C ₁ —D ₁	0.0660	0.0660	0.0660	0.0660
C ₂ —D ₁	0.0889	0.0895	0.0898	0.0902
C ₃ —D ₁	0.0835	0.0845	0.0848	0.0952
C ₄ —D ₁	0.0821	0.0834	0.0839	0.0844
C ₁ —C ₂	0.0455	0.0456	0.0457	0.0458
C ₁ —C ₃	0.0529	0.0539	0.0543	0.0546
C ₁ —C ₄	0.0570	0.0587	0.0593	0.0599
D ₁ —D ₂	0.1277	0.1290	0.1296	0.1303
D ₁ —D ₃	0.1095	0.1107	0.1112	0.1117
D ₁ —D ₄	0.1012	0.1024	0.1028	0.1033

frequencies given by Brodersen and Langseth³ have been applied. To the knowledge of the author, the molecule has not been investigated by electron diffraction.

1. Cyvin, S. J. *Acta Chem. Scand.* **11** (1957) 1499.

2. Bastiansen, O. and Cyvin, S. J. *Nature* **180** (1957) 980.
 3. Brodersen, S. and Langseth, A. *Kgl. Danske Videnskab. Selskab, Mat. Fys. Skrifter* **1** (1956) No. 1.

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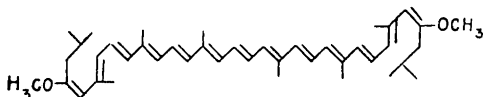
Two New Naturally Occurring Methoxyl-Containing Carotenoids

SYNNØVE LIAAEN JENSEN

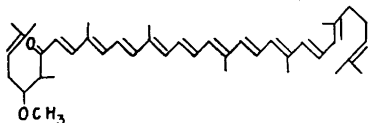
Institutt for Organisk Kjemi, Norges tekniske høgskole, Trondheim, Norway

Methoxyl-containing carotenoids have so far been restricted to the photosynthetic bacteria. Carotenoids hitherto known to contain methoxyl groups are:

1) Spirilloxanthin, isolated by van Niel *et al.*¹ in 1935 from old cultures of *Rhodospirillum rubrum*, and shown by comparison of analytical data² to be identical with rhodoviolascin, isolated the same year by Karrer *et al.*³ from mixed cultures of *Rhodovibrio* species. Karrer *et al.* have proposed the formula for spirilloxanthin⁴.

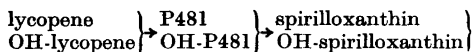


2) Pigment Y (Yellow) and Pigment R (Red), isolated from *Rhodopseudomonas spheroides*, and both shown by Goodwin *et al.*⁵ to contain one methoxyl group. For Pigment R Goodwin *et al.* have proposed the formula



and renamed it spheroidenone⁵.

Kinetic experiments performed by Stanier *et al.*⁶ to establish the route of carotenoid synthesis in the photosynthetic, non-sulphur purple bacterium *Rhodospirillum rubrum*, showed that the following transformation occurs quantitatively during growth:



The structures of the carotenoids P481 and OH-P481 are not known. The designation Pigment 481 was given by Goodwin *et al.*⁷ from the main absorption band in petroleum ether.

During endogenous carotenoid synthesis (anaerobically in light) washed cells of *Rhodospirillum rubrum* previously grown in the presence of diphenylamine, produced considerable amounts of P481 and OH-P481, whereas only a limited synthesis of spirilloxanthin normally was observed. This observation led to the assumption that P481 and OH-P481 were carotenoids with the same C₄₀ skeleton as lycopene, and that lack of intracellular methyl donors limited the endogenous spirilloxanthin synthesis⁸.

In the investigation now reported the presence of methoxyl groups in crystalline, chromatographically pure P481 and OH-P481 was indicated by IR-absorption bands at 1 078 cm⁻¹ (KBr discs).

Quantitative methoxyl determinations were carried out according to the modified method of Inglis⁹, and gave the following results:

Carotenoid	% OCH ₃ found	% OCH ₃ calculated	based on formula
P481	5.51	5.48	C ₄₀ H ₅₅ (OCH ₃)
OH-P481	5.33	5.31	C ₄₀ H ₅₄ (OH)(OCH ₃)

These results establish the presence of one methoxyl group in both P481 and OH-P481.

The fact that P481 and OH-P481 are C₄₁ compounds is not in accordance with our previous suggestion. However, this finding fits well the role that has been ascribed to these two carotenoids as biosynthetic intermediates between lycopene (C₄₀H₅₆) and spirilloxanthin (C₄₂H₆₀O₂)⁶.

Work on the chemical structure of P481 and OH-P481 is in progress.

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