

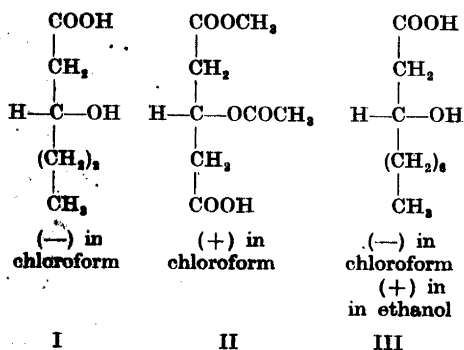
A General Method for the Synthesis of Optically Active β -Hydroxy Acids

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Racemic methyl hydrogen β -acetoxyglutarate, a viscous, slightly hygroscopic liquid with b.p. 150°/0.5 torr (slight decomposition), n_D^{25} 1.4470 and d_4^{25} 1.234, was made from the known anhydride of β -acetoxyglutaric acid^{1,2} and methanol. The optically pure antipodes of this half-ester were obtained in about 30 % yield by fractional crystallization of the cinchonidine and strychnine salts from ethyl acetate and chloroform-ether, respectively. They are liquids with $[\alpha]_D^{25} \pm 6.1^\circ$ (chloroform; c 20; l 1).

Electrolysis of the dextrorotatory antipode with propionic acid gave, via the acetylated methyl ester, β -hydroxyhexanoic acid (I) with m. p. 42° and $[\alpha]_D^{25} - 28^\circ$ (chloroform; c 2; l 1). As this acid is known to have the D configuration³, formula II represents the Fischer projection of the dextrorotatory half-ester. It may be named (+)-methyl 3D-acetoxy-4-carboxybutanoate. Cf.^{4,5}



Electrolysis of the levorotatory half-ester with *n*-heptanoic acid gave similarly 3 L-hydroxydecanoic acid with m. p. 48.4°; $[\alpha]_D^{25} + 20^\circ$ (chloroform; c 2.5; l 1); -3° (ethanol; c 2.5; l 1). The antipode of this acid has been isolated from natural sources⁶⁻⁸ and it follows that this has the D-configuration (III). Cf.³ A specimen kindly supplied by professor S. Bergström gave in equimolecular mixture with the synthetic acid the racemic 3-hydroxydecanoic acid with m. p. 56.6°. Skogh⁹ reports m. p. 56.4–56.6°.

Apparently optically pure β -hydroxy-nonanoic with $[\alpha]_D + 2^\circ 26'$ (ethanol) has been obtained by degradation of ricinoleic acid¹⁰. From the foregoing it seems highly probable that it has the D-configuration. Ricinoleic acid should thus be 12D-hydroxy-*cis*-9-octadecenoic acid.

It is obvious that the optically active half-ester (II) and its antipode can be used as starting material for the synthesis of a great many optically active hydroxy-compounds of the type $\text{RCH}_2\text{CHOHCH}_2\text{R}'$.

A full account of this work will be published later. Grants from *Norges Almenvitenskapelige Forskningsråd* and from *Statens Medicinska Forskningsråd* are gratefully acknowledged.

- Blaise, E. E. *Bull. soc. chim. France* **29** (1903) 1013.
- Böseken, J., Schweizer, A. and van der Want, G. F. *Rec. trav. chim.* **31** (1912) 85.
- Lemieux, R. U. and Giguere, J. *Can. J. Chem.* **29** (1951) 678.
- Klyne, W. *Chemistry & Industry* **1951** 1022.
- Serck-Hanssen, K., Ställberg-Stenhagen, S. and Stenhagen, E. *Arkiv Kemi* **5** (1953) 203.
- Bergström, S., Theorell, H. and Davide, H. *Arch. Biochem.* **10** (1946) 165.
- Bergström, S., Theorell, H. and Davide, H. *Arkiv Kemi, Mineral. Geol.* **23A** (1946) No. 13.
- Ohno, T., Tajima, S., and Toki, K. *J. Agr. Chem. Soc. Japan* **27** (1953) 665.
- Skogh, M. *Acta Chem. Scand.* **6** (1952) 809.
- Haller, A. and Brochet, A. *Compt. rend.* **150** (1910) 496.

Received June, 7, 1955.

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