Short Communications

The Synthesis of Tetraethylthiouram Disulphide (Antabus) labelled with Radioactive Sulphur

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Tetraethylthiouram disulphide is at present in frequent use in Scandinavia as an adjuvans in the treatment of alcoholism. A year only has passed since the discovery ¹ of its peculiar physiological action on the metabolism of ethyl alcohol, and relatively little is yet known about its turnover and way of action.

In order to investigate its fate in the organism, the compound labelled with radioactive sulphur was synthezised in the following way:

 S^{35} was available as sulphuric acid in 0.25 N hydrochloric acid from Oak Ridge National Laboratory, Oak Ridge, Tennessee.

The labelled sulphate together with carrier sulphate was precipitated as calcium sulphate and reduced at 850° C with carbon monoxide 2 to calcium sulphide.

The calcium sulphide was transformed to potassium sulphide by liberating the hydrogen sulphide with hydrochloric acid and receiving it in potassium hydroxide in a diffusion chamber.

The labelled sulphur was brought into carbon disulphide by an exchange reaction between labelled potassium sulphide in water and unlabelled carbon disulphide:

$$S^{*-} + CS_2 \rightleftharpoons CS_2S^{*-} \rightleftharpoons S^- + CSS^*$$

When equilibirum is established, the labelled sulphur will be distributed between the two phases according to their sulphur content. In this way, a yield of 50 % per run of radioactive carbon disulphide was easily obtained.

The labelled carbon disulphide was reacted with diethylamine and potassium hydroxide to give potassium diethyldithiocarbamate. The oxidation to tetraethylthiouram disulphide was achieved by means of sodium tetrathionate⁴.

The compound was recrystallized from absolute ethyl alcohol to give faintly yellow crystalline needles, m. p. 70.4° C.

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